Qn.
Complete the flow chart


Write A,B, and C

Hint.
A - Alkenes
B - $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}-2}$
C - $\mathrm{CH}_{4}$

B- $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}-2}$

C - $\mathrm{CH}_{4}$

## Qn No. 2

Qn.
The hints regarding a cyclic compound are given.

There are 6 carbon atoms.
There are 12 hydrogen atoms

1. Write its structure
2. Write the molecular formula and IUPAC name of the alkane with the same number of carbon atoms

Hint.
1.

2. $\mathrm{C}_{6} \mathrm{H}_{14}$ Hexane

Marks :(4)

Hide Answer

## Qn No. 3

Qn.
Match the following

| A | B | C |
| :---: | :---: | :---: |


|  | Propyne | $\mathrm{C}_{3} \mathrm{H}_{8}$ |
| :---: | :---: | :---: |
|  | Propane | $\mathrm{C}_{3} \mathrm{H}_{6}$ |
|  | Propene | $\mathrm{C}_{3} \mathrm{H}_{4}$ |

Hint.

| A | B | C |
| :---: | :---: | :---: |
|  |  |  |

Hide Answer

Qn No. 4

Qn.
Two hints regarding a hydrocarbon are given

There are four carbon atoms
The general formula of the family of compound is $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 n+2}$

1. Give the molecular formula of this compound
2. Write the structure
3. What will be the molecular formula of the hydrocarbon with the same number of carbon atoms and having a double bond
4. Write the structure of the cyclic hydrocarbon with the same number of carbon atoms

Hint.

1. $\mathrm{C}_{4} \mathrm{H}_{10}$
2. 


3. $\mathrm{C}_{4} \mathrm{H}_{8}$


Hide Answer

Qn No. 5

Qn.
Choose the odd one out. Give reason
$\left(\mathrm{CH}_{4}, \mathrm{C}_{3} \mathrm{H}_{4}, \mathrm{C}_{2} \mathrm{H}_{2}, \mathrm{C}_{2} \mathrm{H}_{4}\right)$

Hint.
$\mathrm{CH}_{4}$
$\mathrm{CH}_{4}$ is a saturated hydrocarbon where as the others are unsaturated
Marks :(2)

Hide Answer

Qn No. 6

Qn.
The structure of a hydrocarbon is given


1. Give its molecular formula
2. Write the IUPAC name of the compound
3. Write the IUPAC name of the cyclic compound with the same molecular formula

Hint.

1. $\mathrm{C}_{3} \mathrm{H}_{6}$
2. Propene
3. Cyclopropane

Marks :(3)

Hide Answer

Qn No. 7

Qn.
Complete this series

| $\mathrm{C}_{2} \mathrm{H}_{4}$ | $\mathrm{C}_{3} \mathrm{H}_{6}$ | $\mathrm{C}_{4} \mathrm{H}_{8}$ | $\ldots . . . . . . . \mathrm{a} . \ldots . . . . . .$. |
| :---: | :---: | :---: | :---: |


| $\mathrm{CH}_{4}$ | $\mathrm{C}_{2} \mathrm{H}_{6}$ | $\ldots . . . . . . \mathrm{b} . . . . . . . . .$. | $\mathrm{C}_{4} \mathrm{H}_{10}$ |
| :---: | :---: | :---: | :---: |


| $\mathrm{C}_{2} \mathrm{H}_{2}$ | $\ldots \ldots . . . . \mathrm{c} . . . . . . .$. | $\mathrm{C}_{4} \mathrm{H}_{6}$ | $\mathrm{C}_{5} \mathrm{H}_{8}$ |
| :---: | :---: | :---: | :---: |

Hint.
a) $\mathrm{C}_{5} \mathrm{H}_{10}$
b ) $\mathrm{C}_{3} \mathrm{H}_{8}$
c ) $\mathrm{C}_{3} \mathrm{H}_{4}$

## Hide Answer

## Qn No. 8

Qn.
The structure of a hydrocarbon is given

a)Give its molecular formula
b)Write its IUPAC name
c)Write the structure of the unsaturated compound with the same molecular formula

Hint.
a) $\mathrm{C}_{4} \mathrm{H}_{8}$
b)Cyclobutane


1-butene
Marks :(3)

Hide Answer

Qn No. 9

Qn.
Look at the structure of the hydrocarbon

a)To which category of hydrocarbons does this compound belong?
b)Give the molecular formula of this compound
c) Name this compound

## Hint.

1. Cyclic compound
2. $\mathrm{C}_{6} \mathrm{H}_{12}$
3. Cyclohexane

Hide Answer

Qn No. 10

Qn.
The structure of a hydrocarbon is given

a) Write its molecular formula
b) What is the word root used to represent the number of carbon atoms ?
c) Write its IUPAC name

Hint.
a) $\mathrm{C}_{4} \mathrm{H}_{10}$
b)But
c)Butane

Hide Answer

Qn No. 11
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
The structure of a hydrocarbon is given

a) What is the molecular formula of this compound
b) Write its IUPAC name
c) To which homologous series does this compound belong?

Hint.
a) $\mathrm{C}_{3} \mathrm{H}_{4}$
b) Propyne
c) Alkyne

Hide Answer

Qn No. 12

Qn.
To which category does $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ belong?
( Alkane,Alkene, Alkyne, Cyclo alkane )

Hint.
Alkane
Marks :(1)

Hide Answer

Qn No. 13

Qn.
Write the structure of $\mathrm{C}_{3} \mathrm{H}_{8}$

Hint.
$\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$

Hide Answer

Qn No. 14
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
Name the functional group of $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{OH}$ ?

Hint.
Hydroxyl

|  |
| :--- | :--- | :--- |
| Hide Answer |

Qn No. 15
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.

The structure of hydrocarbon is given.
$\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
(a) Write the word root used to represent the number of carbon atoms in this compound ?
(b) Give the IUPAC name of this hydrocarbon

Hint.
(a) Hex
(b) Hexane

Qn.
Structure of a cyclic compound is given

a)Write the molecular formula of the compound
b)Write its IUPAC name
c) Write the structure of an open chain hydrocarbon having the same formula

Hint.

1. $\mathrm{C}_{4} \mathrm{H}_{8}$
2. Cyclobutane
3. $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CH}_{3} / \mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{3}$

Marks :(3)

Hide Answer

Qn No. 17

Qn.
Some hydrocarbons are given in the box

$$
\mathrm{C}_{3} \mathrm{H}_{4}, \mathrm{C}_{2} \mathrm{H}_{6}, \mathrm{C}_{2} \mathrm{H}_{2}, \mathrm{C}_{4} \mathrm{H}_{8}, \mathrm{C}_{5} \mathrm{H}_{10}, \mathrm{C}_{3} \mathrm{H}_{8}
$$

1. Which belong to the family with the general formula $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 n+2}$
2. Which compounds have a triple bond
3. Select the alkenes from the box?

Hint.

1. $\mathrm{C}_{2} \mathrm{H}_{6}, \mathrm{C}_{3} \mathrm{H}_{8}$
2. $\mathrm{C}_{3} \mathrm{H}_{4}, \mathrm{C}_{2} \mathrm{H}_{2}$
3. $\mathrm{C}_{4} \mathrm{H}_{8}, \mathrm{C}_{5} \mathrm{H}_{10}$

Hide Answer

Qn No. 18

Qn.
The details of the hydrocarbon $P$ are given below

1. There are $\mathbf{3}$ carbon atoms
2. The family of compounds with $P$ as a member has a general formula $C_{n} H_{2 n}$
3. The IUPAC name of $P$ is Propene
4. Write the condensed formula of the compound
5. Write the IUPAC name of the compound which is before $\mathbf{P}$ in the homologous series
6. Give the molecular formula of the compound succeeding $P$ in the series

Hint.

1. $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{3}$
2. Ethene
3. $\mathrm{C}_{4} \mathrm{H}_{8}$

Marks :(3)

Hide Answer

Qn No. 19

Qn.
The details of the hydrocarbon $P$ are given below

1. There are 3 carbon atoms
2. The family of compounds with $P$ as a member has a general formula $C_{n} H_{2 n}$
3. The IUPAC name of $P$ is Propene
4. Write the condensed formula of the compound
5. Write the IUPAC name of the compound which is before $P$ in the homologous series
6. Give the molecular formula of the compound succeeding $\mathbf{P}$ in the series

Hint.

1. $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{3}$
2. Ethene
3. $\mathrm{C}_{4} \mathrm{H}_{8}$

Hide Answer

## Qn No. 20

Qn.
Given below is a homologous series

| $\mathrm{C}_{2} \mathrm{H}_{2}$ | $\ldots . . . .$. | $\mathrm{C}_{4} \mathrm{H}_{6}$ | $\ldots . \mathrm{B} \ldots$. |
| :--- | :--- | :--- | :--- |

1. What are $A$ and $B$
2. To which family do this belong?
( Alkane, Alkene, Alkyne )
3. Write the IUPAC name of $A$

Hint.

1. $\mathrm{A}-\mathrm{C}_{3} \mathrm{H}_{4}$

B-C $\mathrm{C}_{8}$
2. Alkyne
3. Propyne

Qn.
Given below is a homologous series

| $\mathrm{C}_{2} \mathrm{H}_{2}$ | $\ldots . \mathrm{A} \ldots$. | $\mathrm{C}_{4} \mathrm{H}_{6}$ | $\ldots . \mathrm{B} \ldots$. |
| :--- | :--- | :--- | :--- |

1. What are A and B
2. To which family do this belong?
( Alkane, Alkene, Alkyne )
3. Write the IUPAC name of $A$

Hint.

1. $\mathrm{A}-\mathrm{C}_{3} \mathrm{H}_{4}$

B $-\mathrm{C}_{5} \mathrm{H}_{8}$
2. Alkyne
3. Propyne

Hide Answer

Qn No. 22

Qn.
The formulae given below are of a homologous series

| $\mathrm{CH}_{4}$ | $\mathrm{C}_{2} \mathrm{H}_{6}$ | $\mathrm{C}_{3} \mathrm{H}_{8}$ |
| :--- | :--- | :--- |

1. To which category does this belong?
( Alkane, Alkene, Alkyne )
2. Write the general formula of this family
3. Write the structure of $\mathrm{C}_{2} \mathrm{H}_{6}$
4. Write the IUPAC name of $\mathrm{CH}_{4}$

Hint.

1. Alkane
2. $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}+2}$
3. 


4. Methane

Hide Answer

## Qn No. 23

Qn.
The molecular formulae of some hydrocarbons are given
$\mathrm{C}_{2} \mathrm{H}_{4}, \mathrm{C}_{2} \mathrm{H}_{2}, \mathrm{C}_{2} \mathrm{H}_{6}, \mathrm{C}_{3} \mathrm{H}_{4}, \mathrm{C}_{3} \mathrm{H}_{8}$

1. Which one belongs to the alkene family?
2. To which family does $\mathrm{C}_{2} \mathrm{H}_{2}$ belong?
3. Which belong to the family with general formula $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 n+2}$

Hint.

1. $\mathrm{C}_{2} \mathrm{H}_{4}$
2. Alkyne
3. $\mathrm{C}_{2} \mathrm{H}_{6}, \mathrm{C}_{3} \mathrm{H}_{8}$

Marks :(3)

Hide Answer

Qn.
Self linking property of carbon atoms is known as

Hint.

## Qn No. 25

Qn.
The molecular formula of a cyclic compound is $\mathrm{C}_{4} \mathrm{H}_{8}$.
a) Write the structure of this compound
b) Write the structure of the open chain hydrocarbon having the same molecular formula

## Hint.

a)

b)


Hide Answer

Qn No. 26

Qn.
What is the minimum number of carbon atoms required to form a cyclic comound.

$$
(4,3,2,5)
$$

## Hint. 3

Hide Answer

## Qn No. 27

Qn.
The formulae given below are of a homologous series

| $\mathrm{CH}_{4}$ | $\mathrm{C}_{2} \mathrm{H}_{6}$ | $\mathrm{C}_{3} \mathrm{H}_{8}$ |
| :---: | :---: | :---: |

1. To which category does this belong?
( Alkane, Alkene, Alkyne )
2. Write the general formula of this family
3. Write the structure of $\mathrm{C}_{2} \mathrm{H}_{6}$
4. Write the IUPAC name of $\mathrm{CH}_{4}$

Hint.

1. Alkane
2. $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}+2}$

3. Methane

Hide Answer

Qn No. 28

Qn.
Match the following

| A | B | C |
| :---: | :---: | :---: |
| Molecular formula | Condensed formula | IUPAC Name |
|  |  |  |
| $\mathrm{C}_{3} \mathrm{H}_{4}$ | $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ | Propyne |
| $\mathrm{C}_{4} \mathrm{H}_{8}$ | $\mathrm{CH} \equiv \mathrm{C}-\mathrm{CH}_{3}$ | Butene |
|  |  |  |
| $\mathrm{C}_{3} \mathrm{H}_{8}$ | $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ | Propane |
|  |  |  |

Hint.

| A | B | C |
| :---: | :---: | :---: |


| Molecular formula | Condensed formula | IUPAC Name |
| :---: | :---: | :---: |
| $\mathrm{C}_{3} \mathrm{H}_{4}$ | $\mathrm{CH} \equiv \mathrm{C}-\mathrm{CH}_{3}$ | Propyne |
|  |  |  |
| $\mathrm{C}_{4} \mathrm{H}_{8}$ | $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ | Butene |
|  |  |  |
| $\mathrm{C}_{3} \mathrm{H}_{8}$ | $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ | Propane |

Hide Answer

Qn No. 29
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
The molecular formula of a hydrocarbon is $\mathrm{C}_{2} \mathrm{H}_{4}$
a) Name the homologous series of which this is a member
b) Write the molecular formula of the Fifth member
c) Write the structure of $\mathrm{C}_{2} \mathrm{H}_{4}$ and give its IUPAC name

Hint.
(a) Alkene
(b) $\mathrm{C}_{6} \mathrm{H}_{12}$
(c) $\mathrm{CH}_{2}=\mathrm{CH}_{2}$; Ethene

Marks :(3)

Hide Answer

Qn No. 30

Qn.
The molecular formula of a hydrocarbon is $\mathrm{C}_{2} \mathrm{H}_{4}$
a) Name the homologous series of which this is a member
b) Write the molecular formula of the Fifth member
c) Write the structure of $\mathrm{C}_{2} \mathrm{H}_{4}$ and give its IUPAC name

Hint.
(a) Alkene
(b) $\mathrm{C}_{6} \mathrm{H}_{12}$
(c) $\mathrm{CH}_{2}=\mathrm{CH}_{2}$; Ethene

Qn No. 31

Qn.
Categorise the given hydrocarbons
$\mathrm{C}_{2} \mathrm{H}_{4}, \mathrm{C}_{3} \mathrm{H}_{8}, \mathrm{C}_{4} \mathrm{H}_{6}, \mathrm{CH}_{4}, \mathrm{C}_{5} \mathrm{H}_{10}, \mathrm{C}_{6} \mathrm{H}_{10}$
(Hint: Hydrocarbons can be catogorised as Alkanes, Alkenes, Alkynes)

Hint.
Alkanes: $\mathrm{CH}_{4}, \mathrm{C}_{3} \mathrm{H}_{8}$
Alkenes: $\mathrm{C}_{2} \mathrm{H}_{4}, \mathrm{C}_{5} \mathrm{H}_{10}$
Alkynes: $\mathrm{C}_{4} \mathrm{H}_{6}, \mathrm{C}_{6} \mathrm{H}_{10}$

Hide Answer

Qn No. 32

Qn.
Molecular formulae of some hydrocarbons are given in the box

$$
\mathrm{C}_{3} \mathrm{H}_{6}, \mathrm{C}_{4} \mathrm{H}_{8}, \mathrm{C}_{5} \mathrm{H}_{10}, \mathrm{C}_{6} \mathrm{H}_{12}
$$

a) To which Homologous series do these belong?
b) Give two reasons for them being homologous.

Hint.
a) Alkene
(b) i. Immediate neighbours differ by $\mathrm{CH}_{2}$
ii. Can be represented by a general formula $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 n}$

Hide Answer

Qn.

The structure of a hydrocarbon is given

a) Write the condensed formula
b) Write its molecular formula
c) Write the structure of the first member of homologous series having this one as a member and give its IUPAC name

Hint.
@) $\mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{CH}$
凹ி) $\mathrm{C}_{3} \mathrm{H}_{4}$
mา) $\mathrm{CH} \equiv \mathrm{CH}$ Ethyne
Marks :(4)

Hide Answer

Qn No. 34

Qn.
The structure of a hydrocarbon is given

a) Write the condensed formula
b) Write its molecular formula
c) Write the structure of the first member of homologous series having this one as a member and give its IUPAC name

Hint.
@) $\mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{CH}$
ゅி) $\mathrm{C}_{3} \mathrm{H}_{4}$
m() $\mathrm{CH} \equiv \mathrm{CH}$ Ethyne

## Qn No. 35

Qn.
$\mathrm{C}_{2} \mathrm{H}_{6}, \mathrm{C}_{3} \mathrm{H}_{8}$ $\qquad$ $\mathrm{C}_{5} \mathrm{H}_{12}$ are the members of a homologous series
a)Write the molecular formula of the missing compound
b)What is the name of this homologous series
c) Write the structure of $\mathrm{C}_{2} \mathrm{H}_{6}$

Hint.
a) $\mathrm{C}_{4} \mathrm{H}_{10}$
b) Alkane
c) $\mathrm{CH}_{3}-\mathrm{CH}_{3}$

Hide Answer

Qn No. 36

Qn.
Name the functional group present in the compound $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{OH}$ ?

Hint.
Hydroxyl

Hide Answer

Qn No. 37
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
The structures written by two students are given
Student 1:


Student 2 :


Write the IUPAC names and say whether these two are isomeric pairs

Hint.
Student 1: Butan -2- OI.
Student 2: Butan -2- OI.
As both are the structure of the same compound they are not isomeric pairs

Hide Answer

## Qn No. 38

Qn.

$$
\mathrm{C}_{3} \mathrm{H}_{6}, \mathrm{C}_{3} \mathrm{H}_{4}, \mathrm{C}_{4} \mathrm{H}_{8}, \mathrm{C}_{4} \mathrm{H}_{10}
$$

Of the given compounds, the name of which one ends with "-yne"

Hint. $\mathrm{C}_{3} \mathrm{H}_{4}$
Marks :(1)

Hide Answer

Qn No. 39

Qn.
$\mathrm{C}_{2} \mathrm{H}_{6}, \mathrm{C}_{3} \mathrm{H}_{8}$, $\qquad$ , $\mathrm{C}_{5} \mathrm{H}_{12}$ are the members of a homologous series
a)Write the molecular formula of the missing compound
b) What is the name of this homologous series
c) Write the structure of $\mathrm{C}_{2} \mathrm{H}_{6}$

## Hint.

a) $\mathrm{C}_{4} \mathrm{H}_{10}$
b) Alkane
c) $\mathrm{CH}_{3}-\mathrm{CH}_{3}$

Qn.
Some molecular formulae are given
(i) $\mathrm{C}_{5} \mathrm{H}_{12}$ (ii) $\mathrm{C}_{5} \mathrm{H}_{10}$ (iii) $\mathrm{C}_{5} \mathrm{H}_{8}$ (iv) $\mathrm{C}_{5} \mathrm{H}_{12} \mathrm{O}$
a) Which of the above is the molecular formula of Pent-2-ene ?
b)Write the stucture of pent-2-ene.
c) Can there be a compound named pent-3-ene

Hint.
(a) $\mathrm{C}_{5} \mathrm{H}_{10}$
(b) correct structure
(b) No

Hide Answer

Qn No. 41

Qn.
Write the two possible structures of compounds with molecular formula $\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}$. Write their IUPAC names.

Hint.
(a) $\mathrm{CH}_{3}-\mathrm{O}-\mathrm{CH}_{3}$ Methoxymethane
(b) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{OH}$ Ethanol

Marks :(4)

Hide Answer

Qn No. 42
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
The molecular formula of the carboxylic acid in vinegar is $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{O}_{2}$
(a) Write the structural formula
(b) Give its IUPAC name

Hint.
(a) $\mathrm{CH}_{3}-\mathrm{COOH}$
(b) Ethanoicacid

## Qn No. 43

Qn.
$\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
(a) Give the IUPAC name of the given open chain compound.
(b) Write the structure of the cyclic compound having the same number of carbon atoms
(c) Write the IUPAC name of this cyclic compound

Hint.
(a) Hexane
(b)

or any other correct answer
(c) Cyclohexane

Marks :(4)

Hide Answer

## Qn No. 44

Qn.
(a) Write the structure of 2,2-dimethylhexane
(b) Write the structure of any one its chain isomer

Hint.
${ }_{1}{ }^{\mathrm{CH}}$
$\mathrm{CH}_{5}-\mathrm{C}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
$\mathrm{CH}_{3}$
(b) any correct one

Marks :(2)

Hide Answer

Qn No. 45

Qn.


The main chain consists of 10 carbon atoms and the same is represented by the word root 'dec'
(a) Give the position of the branches
(b) Write the IUPAC name of the compound

Hint.
(a) 2,7,8
(b) 2,7,8 - Trimethyldecane

Hide Answer

Qn No. 46

Qn

(a) How many carbon atoms are there in the main chain?
(b) Number the position of the carbon with the branch ?
(c) Name the branch?
(d) Write the IUPAC name of the compound

Hint.
(a) 4
(b) 2
(c) Methyl
(d) 2-Methylbutane

Marks :(4)

Hide Answer

Qn No. 47

Qn. (i) $\mathrm{CH}_{3}-\mathrm{O}_{-} \mathrm{CH}_{3}$
(ii) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{OH}$
(a) Write the IUPAC names of the given compounds
(b) Which type of isomers are these compounds ?

Hint.
(a) (i) Methoxymethane
(ii) Ethanol
(b) Functional Isomers

## Qn No. 48

Qn.
Look at the structure

## CH:HH:H:HOH

(a) Write its IUPAC name
(b) Name its position isomer
(c) Write the structure of its functional isomer

## Hint.

a)Propan-1-ol
b) Propan-2-ol
(c. $\mathrm{HH}_{3} \mathrm{CH} \mathrm{CH}_{2} \cdot \mathrm{O}_{2} \cdot \mathrm{CH}_{3}$

Hide Answer

## Qn No. 49

Qn.
Match suitably

| $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ | 2,2-Dimethylpropane |
| :---: | :---: |
|  | Pentane |
|  | 2-Methylbutane |

## Hint.

| $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ | Pentane |
| :---: | :--- |
| $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ <br> CH <br> CH | 2- Methyl butane |
| $\mathrm{CH}_{3}$ |  |
| $\mid$ |  |
| $\mathrm{CH}_{3}-\mathrm{C}-\mathrm{CH}_{3}$ |  |
| \| |  |
| $\mathrm{CH}_{3}$ | 2,2- Di methyl Propane |

## Qn No. 50

Qn.
The chain of a hydrocarbon is given

(a) Complete the structure
(b) How many carbon atoms are there in the longest chain
(c) Give the position of the branch
(d) Write down the IUPAC name of the compound

Hint.

(b) 4
(c) 2,3
(d) 2,3-Dimethylbutane

Marks :(4)

Hide Answer

Qn No. 51

Qn.
The structure of a compound is $\mathrm{CH}_{3}-\mathrm{O}-\mathrm{CH}_{3}$
(a) What is the IUPAC name of the compound
(b) Write the structure of its isomer
(c) What is the IUPAC name of this isomer.
(d) What type of isomers are these compounds?

Hint.
(a) Methoxy methane
(b) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{OH}$
(c) Ethanol
(d) Functional isomers

Qn.

(a) How many carbon atoms are there in the longest chain of the compound given above?
(b) Give the position of the branches ?
(c) Write the IUPAC name of this compound

Hint.
(a) 5
(b) 2,2
(c) 2,2-Di methyl pentane

Marks :(3)

Hide Answer

## Qn No. 53

Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
See the structure given

## $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$

(a) Write the IUPAC name of this compound
(b) Write the molecular formula of the alkene having the same number of
carbon atoms
(c) Write the structures of the position isomers of this alkene.

Hint.
(a) Butane
(b) $\mathrm{C}_{4} \mathrm{H}_{8}$
${ }^{\text {(c) }}{ }^{\mathrm{CH}_{3}}=\mathrm{CH}-\mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{3}-\mathrm{CH}_{3}$
Marks :(4)

Hide Answer

## Qn No. 54

Qn.
a) Choose any pairs showing different types of isomerism from the structures given
. $\mathrm{CH}_{3}-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
3. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{OH}$
4.


OH
b) To which type of isomerism do these pairs belong?

Hint.

1. $\mathrm{CH}_{3}-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}_{3} / \mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{OH}$
|
OH

Marks :(4)

Hide Answer

## Qn No. 55

Qn.
Examine the given structure
$\mathrm{CH}_{3} \mathrm{CH}_{\mathrm{H}} \mathrm{C} \cdot \mathrm{O} \cdot \mathrm{CH}_{2} \cdot \mathrm{CH}_{3}$
(a) Give the name of the functional group?
(b) Write the common name of the category of compounds with this functional group?
(c) Give the IUPAC name of the compound

Hint.
(a)Alkoxy group OR Ethoxy
(b)Ethers
(c)Ethoxyethane

Marks :(3)

Hide Answer

Qn No. 56

Qn. $\begin{array}{r}\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2} \mathrm{CH}_{3} \\ \mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{3}\end{array}$
(a) How many carbon atoms are there in the parent chain of the above compound?
(b) What is the position of the branched carbon ?
(c) Give the name of the branch?
(d) Write the IUPAC name of the compound

Hint.
(a) 8
(b) 4
(c) Ethyl
(d) 4-Ethyloctane

Hide Answer

## Qn No. 57

Qn.
To which category does the compound $\mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}_{2}$ belong?
( Alkane,Alkene, Alkyne, Cyclo alkane )

Hint.
Alkene

Marks :(1)

Hide Answer

Qn No. 58

Qn.

(a) How many branches are there in the compound ?
(b)Give the position of the branches ?
(c) Write the IUPAC name

Hint.
(a) 3
(b) 2,3,6
(c)2,3,6- Trimethyloctane

Marks :(3)

Hide Answer

Qn.
Write the structure of but-2-ene

Hint.
$\mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{3}$
Marks :(1)

Hide Answer

Qn No. 60

Qn.
To which category does $\mathrm{CH} \equiv \mathrm{CH}$ belong?
( Alkane,Alkene, Alkyne, Cyclo alkane )

Hint.
Alkyne
Marks :(1)

Hide Answer

Qn No. 61

Qn.
Examine the given structural formula

(a) What is the molecular formula of the compound.
(b) Identify the functional group?
(c) Give the IUPAC name of the compound
(d) Write the structure of its isomer

## Hint.

(a) $\mathrm{C}_{3} \mathrm{H}_{7} \mathrm{Cl}$
(b) chloro $/-\mathrm{Cl}$
(c) 2-chloropropane
(d) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{Cl}$
Qn.

(a) Name the functional group in this compound?
(b) What is the common name of compounds with this functional group?
(c) Give the IUPAC name of the compound

Hint.
(a) Hydroxyl
(b) Alcohols
(c) Propan -2-ol

Marks :(3)

Hide Answer

## Qn No. 63

Qn.
The IUPAC name of a compound is Pent-2-yne
(a) To which category of hydrocarbons does this belong?
( Alkane,Alkene, Alkyne,)
(b) Give the structure of the compound
(c) What is its molecular formula?

Hint.
(a) Alkyne
(b) $\mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{C}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
(c) $\mathrm{C}_{5} \mathrm{H}_{8}$

Hide Answer

Qn No. 64
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
The structure of a compound is $\mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{C}-\mathrm{CH}_{3}$
(a) What is its molecular formula
(b) To which category of hydrocarbon does this hydrocarbon belong
( Alkane,Alkene, Alkyne,)
(c) Give the IUPAC name of this compound

Hint.
(a) $\mathrm{C}_{4} \mathrm{H}_{6}$
(b) Alkyne
(c) But -2-yne

Hide Answer

Qn No. 65
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
The IUPAC name of a compound is Pent-2-yne
(a) To which category of hydrocarbons does this belong?
( Alkane,Alkene, Alkyne,)
(b) Give the structure of the compound
(c) What is its molecular formula ?

Hint.
(a) Alkyne
(b) $\mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{C}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
(c) $\mathrm{C}_{5} \mathrm{H}_{8}$

Marks :(3)

Hide Answer

Qn No. 66

Qn.
Write the structure of
3- Ethylhexane

Hint. $\begin{array}{r}\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2} \mathrm{CH}_{3} \\ \mathrm{CH}_{2}-\mathrm{CH}_{3}\end{array}$
Marks :(1)

Hide Answer

Qn No. 67
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn. $\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
(a) Write the IUPAC name of the compound
(b) What will be the IUPAC name of the compound, if the double bond were in between the second and third carbon atoms?

Hint.
(a)But-1-ene
(b)But -2-ene
(2):(2)

Hide Answer

Qn No. 68

Qn.
Write the structure of
3,3 - Diethylheptane

Hint.


Marks :(1)

Hide Answer

Qn No. 69

Qn.
Some carbon compounds are given
i) $\mathrm{CH}_{2}=\mathrm{CH}_{2}$
ii)

iii)

iv) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{3}$
a) Catogarise the above as alkane, alkene, alkyne and cyclic compound
b) Most of the compounds in nature contains carbon. Do you agree with this statement? Justify

Hint.
i) Alkene
ii ) Cyclic compounds
iii ) Alkyne
iv ) Alkane
b) Agree.

Carbon forms extremely large number of compounds. Compounds with single,double, and triple bonds between carbon atoms can be formed. Has self linking property catenation to form chains and rings

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Hide Answer
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Qn No. 70

Qn.

(a) How many carbon atoms are there in the main chain?
(b)Give the position of the branches?
(c) Write the IUPAC name

Hint.
(a) 5
(b) 2,4
(c)2,4- Dimethylpentane

Marks :(3)

Hide Answer

Qn No. 71
Chapter Name:Nomenclature of Organic compounds and Isomerism

Qn.
A few structures are given


a) By which name are these compounds known?
b) How many hydrogen atoms will be there in such a compound with five carbon atoms. Write the structure and give its IUPAC name

Hint.
a) Cyclic compounds
b) 10 atoms, Cyclopentane


## Qn No. 72

Qn.
The structure of a hydrocarbon is given
$\mathrm{H}-\mathrm{C} \equiv \stackrel{\mathrm{C}}{\mathrm{H}} \underset{\mathrm{C}}{\mathrm{C}} \underset{\mathrm{C}}{\mathrm{H}} \mathrm{H}$
1.Write the IUPAC name of this compound
2.Write the general formula of the family having this one as a member
( $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}+2}, \mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}}, \mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}-2}$ )
3.Write the molecular formula of the compound after this one in the homologous series

Hint.

1. Propyne
2. $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}-2}$
3. $\mathrm{C}_{4} \mathrm{H}_{6}$

Hide Answer

