CHEMISTRY

Standard : VIII

1.

Time: 120 Minutes

Attempt any three of the following questions from 1 to 4 (one mark each)

Molecular arrangements of solid, liquid and gas are depicted below.



Identify the molecular arrangement of gas

- 2. Find out the correct statement from the following?
 - A. Oxygen is a monoatomic molecule.
 - B. Chlorine is an example for a compound molecule.
 - C. Water splits into Hydrogen and Oxygen by passing electricity
 - D The element Chromium got its symbol from the name of a planet

E 806 - 2/6

Which one is an example for polyatomic molecule?

A Hydrogen	C. Oxygen
B. Sulphur	D. Helium

4.

3



 $1 \times 3 = 3$

2

1

Attempt any four of the following questions from five to nine (two mark each)

- 5. Pick out the correct changes during the tranformation of ice to water
 - A Possibility of diffusion increases
 - B. Energy of particles decreases
 - C. Distance between the particles decreases
 - D. Attraction between the particles decreases
- 6. Symbol of Chlorine is Cl. Represent two Chlorine atoms and one Chlorine molecule by using this symbol?
- 7. Complete the table.

Element	Basis of Naming	Symbol
Copper	Latin name	(A)
(B)	· Name of a scientist	Md

- 8. Identify the reactants and products from the chemical reaction given NaOH+H₂SO₄ \longrightarrow Na₂SO₄+H₂O
- 9. Which method is used to separate the mixture on the basis of the difference in rate of adsorption of components? Write an example for this type of separation. $4 \ge 8$

Attempt any three of the following questions from 10 to 13 (three marks each)

10. Samples of some molecules are given below.

		Abox 2007 100-100000 1000000	
5NO ₂	NH ₃	5HNO ₃	7KCl
-			

- a) Which sample has more number of atoms?
- b) Arrange the samples in the increasing order of number of atoms?
- 11.

- a) Write the name of this apparatus?
- b) Can we adopt this method for the separation of Petrol and Kerosine? Justify your answer.



E 806 - 3/6

12. Match with column A from column B and C of the given table.

A	B	С	
Mixture	Properties of components	Method of separation	
1. Ethanol+Methanol	Mass difference	Magnetic separation	
2. Butter from Curd	Magnetic property	Fractional distillation	
3. Aluminium powder Narrow difference in and Iron powder		Centrifugation boiling points	
	Chromatography	all a start	

13. Distillation of Salt solution is picturised below:



a) Identify the apparatus A

b) Name the products B and C after distillation 2

111

4

1