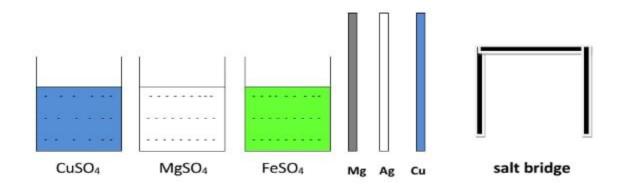
## REACTIVITY SERIES AND ELECTROCHEMISTRY

1 Which reaction is taking place in the galvanic cell? (oxidation, reduction, redox reaction)

2 In galvanic cell the direction of electron flow is from.....?

(anode to cathode, cathode to anode, salt bridge to voltmeter)

3 Few solutions and metal rods are given below



- a) Draw the diagram of a galvanic cell that can be made using these substances?
- b) Find out the anode?
- c) Write the chemical equations for the reaction at cathode?



4 Find out the cathode and anode for each cell given below?

Cell	Anode	Cathode
a) Zn-Cu		
b) Fe-Ag		
c) Cu-Ag		

