

JEE Main Sample Papers

by





FIITJEE - JEE (Main)

SAMPLE TEST - 1

Time Allotted: 3 Hours

Maximum Marks: 360

- Do not open this Test Booklet until you are asked to do so.
- Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.

Important Instructions:

- 1. Immediately fill in the particulars on this page of the Test Booklet with Blue / Black Ball Point Pen. Use of pencil is strictly prohibited.
- 2. The Answer Sheet is kept inside this Test Booklet. When you are directed to open the Test Booklet, take out the Answer Sheet and fill in the particulars carefully.
- 3. The test is of **3 hours** duration.
- 4. The Test Booklet consists of 90 questions. The maximum marks are 360.
- 5. There are *three* sections in the question paper I, II, III consisting of **Physics**, **Chemistry** and **Mathematics** having 30 questions in each section of equal weightage. Each question is allotted **4 (four)** marks for correct response.
- Candidates will be awarded marks as stated above in instruction No.5 for correct response of each question.
 ¼ (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
- 7. There is only one correct response for each question. Filling up more than one response in any question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction 6 above.
- 8. Use *Blue / Black Ball Point Pen only* for writing particulars / marking responses on *Side-1* and *Side-2* of the Answer Sheet. *Use of pencil is strictly prohibited.*
- 9. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. except the Admit Card inside the examination hall / room.
- 10. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty in the Room / Hall. *However, the candidates are allowed to take away this Test Booklet with them.*
- 11. Do not fold or make any stray marks on the Answer Sheet.

Name of the Candidate (in Capital Letters) :			
Enrolment Number :			
Batch :	Date of Examination :		

<u>____</u>

Useful Data Chemistry:		
Gas Constant	R =	8.314 J K ⁻¹ mol ⁻¹
	=	0.0821 Lit atm K ⁻¹ mol ⁻¹
	=	$1.987 \approx 2 \text{ Cal } \text{K}^{-1} \text{ mol}^{-1}$
Avogadro's Number	N _a =	6.023×10^{23}
Planck's Constant	h =	$6.626 \times 10^{-34} \text{ Js}$
	=	6.25 x 10 ⁻²⁷ erg.s
1 Faraday	=	96500 Coulomb
1 calorie	=	4.2 Joule
1 amu	=	1.66 x 10 ⁻²⁷ kg
1 eV	=	1.6 x 10 ⁻¹⁹ J
Atomic No :	H=1, D=1,	Li=3, Na=11, K=19, Rb=37, Cs=55, F=9, Ca=20, He=2, O=8,
	Au=79.	
Atomic Masses:	He=4, Mg=	24, C=12, O=16, N=14, P=31, Br=80, Cu=63.5, Fe=56,
	Mn=55, Si	= 28 Pb=207, u=197, Ag=108, F=19, H=2, Cl=35.5
Useful Data Physics:		
Acceleration due to	gravity g =	10 m/s ²

Section – I (Physics)

1. A particle is projected vertically upwards with a velocity of 20 m/sec. Find the time at which the distance travelled is twice the displacement

(A) $2 + \sqrt{4/3}$ sec. (B) 1 sec.

(C)
$$2 + \sqrt{3/4}$$
 (D) 3 sec

2. Two men who can swim with a speed v_1 in still water start from the middle of a river of width d and move in opposite directions always swimming at an angle θ with the banks. What is the distance between them along the river when they reach the opposite banks, if the velocity of the river is v_2

(A)
$$\frac{dv_1}{dv_2} \cot\theta$$
 (B) $\frac{dv_1 \cos\theta}{v_1 + v_2}$ (C) $\frac{dv_2}{v_1} \tan\theta$ (D) $\frac{v_2 d}{v_1 \sin\theta}$

- 3. A uniform chain of mass m hangs from a light pulley, with unequal lengths of the chain hanging from the two sides of pulley the force exerted by moving chain on the pulley is
 - (A) = mg
 - (B) > mg
 - (C) < mg

(D) either b or c depending upon acceleration of chain

- In the figure ball A is released from rest, when the spring is at its natural length. For the block B of mass M to leave contact with the ground at some stage, the minimum mass of A must be (A) 2M
 - (B) M
 - (C) M/2
 - (D) a function of M and force constant of spring.
- 5. A system is shown in figure pulleys and strings are ideal system is released from rest $a_1 \rightarrow acceleration of 2 \text{ kg}, a_2 = acceleration of 3 \text{ kg}$ (i) $a_1 = 2a_2$ (ii) $a_2 = 2a_1$ (iii) $a_1 = a_2 = 0$





- (B) (iii), (v) are correct(D) (i), (iv) & (vi) are correct
- 6.A bullet of mass 10 gm is fired from a rifle with a velocity of 800 m/s. After passing through a mud
wall 180 cm thick, the velocity drops to 100 m/s. The average resistance of the wall is
(A) 750 N(B) 1250 N(C) 1750 N(D) 2250 N
- 7. A body is gently dropped on a conveyor belt moving 3 m/s. If $\mu = 0.5$ how far will the body move relative to the belt before coming to rest? (g = 10m/s²) (A) 0.3 m (B) 0.6 m (C) 0.9 m (D) 0.8 m
- 8. Consider the disc kept on a rough horizontal surface as shown in the diagram. If a horizontal force 'F' has to be applied such that the disc starts pure rolling, what should be the value of 'h' ?
- R F

(A) R (B) R/3 (C) R / 2

- (D) Body can't start pure rolling for any value of 'h'
- 9. Moment of inertia of a half shell of mass 'M' about an axis tangential to it, as shown, would be
 - (A) $\frac{2}{3}$ MR² (B) $\frac{1}{3}$ MR² (C) MR²
- 10. Consider '3' bodies namely a disc 'A', a sphere 'B' and a hollow cylinder all with same mass and radius being released from top of fixed inclined plane. If t_A , $t_B \& t_c$ be the time they take to reach the bottom, find the correct alternative for each of given situations.

In absence of any friction (A) $t_{\rm C} > t_{\rm A} > t_{\rm B}$ (B) $t_{\rm C} < t_{\rm A} < t_{\rm B}$

(C) $t_{C} = t_{A} = t_{B}$

(D) none of these

(D) none of these

R

h=2F

11. A small ball starts rolling on an inclined track which becomes loop if radius R in vertical plane.

(A) speed of the ball at highest point is zero but and highest point is 2R above the ground.

(B) speed of the ball at highest point is non zero but highest point is 2R above the ground(C) speed of the ball is along horizontal at highest point and highest point is less than the 2R above the ground.

(D) speed of the ball is along horizontal at highest point but height of highest point above the ground can not be calculated.

- 12. A double star consists of two stars having mass M and 2M. The distance between their centre is equal to r. They revolve under their mutual gravitational interaction. Then which of the following statements is/are correct.
 - (A) heavier star revolves in orbit of radius r/3
 - (B) kinetic energy of heavier star is twice of that of the other star.
 - (C) lighter star revolves in orbit of radius $\frac{2r}{5}$
 - (D) all above are correct.
- 13. Three small identical bodies each of mass m are moving in circular orbit around a fixed point with same angular velocity under their gravitational interaction. If the separation between any two bodies is R, the total energy possessed by the system is given by

(A)
$$-\frac{3GM^2}{2R}$$
 (B) $-\frac{3GM^2}{4R}$ (C) $-\frac{3GM^2}{2R\cos 30^{\circ}}$ (D) $-\frac{3GM^2}{R}$

14. A man of mass M stands at one end of a plank of length L which lies at rest on a frictionless surface. The man walks to the other end of the plank. If the mass of the plank is M/3, the distance that the man moves relative to the ground is

(A)
$$\frac{3L}{4}$$
 (B) $\frac{4L}{5}$ (C) $\frac{L}{4}$ (D) $\frac{L}{3}$





space for rough work

Section – II (Chemistry)

1.	For the equations $C(diamond) + 2H_2(g) \rightarrow CH_4(g); \Delta H_1$ $C(graphite) + 2H_2(g) \rightarrow CH_4(g); \Delta H_2$ Predict whether (A) $\Delta H_1 = \Delta H_2$ (C) $\Delta H_1 < \Delta H_2$	(B) $\Delta H_1 > \Delta H_2$ (D) $\Delta H_1 = \Delta H_2 + \Delta H_{vap}$	$(3) + \Delta H_{diss}(H_2)$
2.	Which of the following can behave as both (A) $CH_3 - C \equiv N$ (B) CH_3OH	electrophile and nucleophile (C) $CH_2 = CH - CH_3$	A
3.	Steam reacts with iron at high temperature $3Fe(s) + 4H_2O(g) \Rightarrow Fe_3O_4(s) + 4H_2(g)$ The correct expression for the equilibrium of (A) $\frac{p_{H_2}^2}{p_{H_2O}^2}$ (B) $\frac{(p_{H_2})^4}{(p_{H_2O})^4}$		(D) $\frac{[Fe_3O_4]}{[Fe]}$
4.	In which of the following solvents, AgBr has (A) 10 ⁻³ M NaBr (B) 10 ⁻³ M NH₄OH	the maximum solubility? (C) Pure water	(D) 10 ⁻³ M HBr
5.	The electrode potential of a copper wire dip reduction potential of copper is 0.34V): (A) 0.34V (B) 0.31V	ped in 0.1 M CuSO₄ solution (C) 0.349 V	n at 25 ⁰ C (the standard (D) 0.28 V
6.	The strongest reducing agent among the for (A) F ⁻ (B) Cl ⁻	llowing is (C) Br ⁻	(D) I ⁻
7.	Diazonium salt decomposes as: $C_6H_5N_2^+Cl^- \rightarrow C_6H_5Cl + N_2$ At 0° <i>C</i> , the evolution of N ₂ becomes two tin doubled., Therefore, it is (A) a first order reaction (B) a second order reaction (C) independent of the initial concentration (D) a zero order reaction.		oncentration of the salt is
8.	If 2.68×10^{-3} mole of a solution containing a	an ion $A^{\scriptscriptstyle n+}$ requires $1.61{ imes}1$	$0^{^{-3}}$ mole of $MnO_4^{^-}$ for
	the oxidation of A^{n+} to AO_3^- in an acidic m (A) 3 (B) 2	nedium, then what is the valu (C) 5	ue of n? (D) 4
9.	The equivalent weight of $MnSO_4$ is half its n (A) Mn_2O_3 (B) MnO_2	nolecular weight when it is c (C) MnO_4^-	onverted to (D) MnO_4^{-2}
10.	The largest number of molecules is in (A) 36 g of H_2O (C) 46 g of CH_3CH_2OH (Use atomic weight: O = 16, C = 12, N = 14	(B) 28 g of CO (D) 54 g of N ₂ O ₅ , H = 1)	

Consider a titration of $K_2 Cr_2 O_7$ with acidified Mohr's salt solution $(FeSO_4.(NH_4), SO_4.6H_2O)$ 11. using diphenylamine as indicator. The number of moles of Mohr's salt required per mole of dichromate is $Fe^{+2} + Cr_2O_7^{2-} \xrightarrow{H^+} Fe^{+3} + Cr^{+3}$ (A) 3 (B) 4 (C) 5 (D) 6 The normality of 0.3 M phosphorus acid (H_3PO_3) is 12. (A) 0.1 (B) 0.9 (C) 0.3 (D) 0.6 13. The energy of hydrogen atom in the ground state is $-13.6 \,\text{eV}$. Its energy corresponding to the quantum number n = 5 is (A) -0.54 eV(B) −5.40 *eV* (C) -0.85 eV(D) $-2.72 eV_1$ 14. If r₁ is the radius of first orbit of hydrogen atom, then the radii of second, third and fourth orbit in terms of r1 are (A) $8r_1, 27r_1, 64r_1$ (B) $2r_1, 6r_1, 8r_1$ (C) $4r_1, 9r_1, 16r_1$ (D) $r_1, 2r_1, 3r_1$ 15. If kinetic energy of an electron is increased nine times, the de-Broglie wavelength associated with it would become (C) $\frac{1}{3}$ times (D) $\frac{1}{9}$ times (A) 3 times (B) 9 times 16. Which electronic level allows the hydrogen atom to absorb a photon but not emit a photon? (C) 1s (D) 3d (A) 2s (B) 2p CH₃ NaOMe 17. Product. Product of the reaction is: CH_3OH Br $(E_2 \text{ reaction})$ CH_3 (A) (B) (C) (D) No reaction OMe CH₃ 18. Major product of the reaction is: NaOEt ^{′′′′}′Cl



19. Which of the given options best describes the product of the following reaction? $C_{I}^{H_{3}}$

Ph $CD_3 \xrightarrow{K^+t-BuO^-} Product$

- (A) Absolute configuration has been inverted
- (B) Absolute configuration has been retained
- (C) Racemisation (loss of absolute configuration)
- (D) Loss of chirality has occurred (the product is achiral)

20. Which one of the following undergoes nucleophlic aromatic substitution at the fastest rate?





23. An organic compound 'B' is formed by the reaction of ethyl magnesium iodide (CH_3CH_2MgI) with a substance 'A', followed by treatment with dil. aqueous acid. Compound 'B' doesnot react with PCC. Identify A?

(A)
$$(A) = (A) =$$

Compound (B) is

 (C_4H_7ClO)





- 26. Equimolar solutions of two non electrolytes in the same solvent have
 (A) same b.pt but different f.pt
 (C) same b.pt and same f.pt
 (D) different b.pt and different f.pt
- 27. The degree of dissociation (α) of weak electrolyte $A_x B_y$ is related to van't Hoff's factor (i) by the expression:

(A)
$$\alpha = \frac{i-1}{(x+y-1)}$$
 (B) $\alpha = \frac{i-1}{(x+y+1)}$ (C) $\alpha = \frac{(x+y-1)}{i-1}$ (D) $\alpha = \frac{(x+y+1)}{i-1}$

- 28. When 20 gm of naphthoic acid $(C_{11}H_8O_2)$ is dissolved in 50 gm of benzene $(k_f = 1.72)$, a freezing point depression of 2k is observed. The van't Hoff factor (i) is (A) 0.5 (B) 1 (C) 2 (D) 3
- 29. A 0.004 M solution of Na_2SO_4 is isotonic with a 0.01 M solution of glucose at same temperature. The apparent degree of dissociation of Na_2SO_4 is (A) 25% (B) 50% (C) 75% (D) 85%
- 30. If equal volumes of $BaCl_2$ and NaF solutions are mixed, which of these combination will not give a precipitate. $(K_{sp} \text{ of } BaF_2 = 1.7 \times 10^{-7})$
 - (A) $10^{-3}M \ BaCl_2$ and $2 \times 10^{-2}M \ NaF$ (B) $10^{-3}M \ BaCl_2$ and $1.5 \times 10^{-2}M \ NaF$
 - (C) $1.5 \times 10^{-2} M BaCl_2$ and $10^{-2} M NaF$ (D) $2 \times 10^{-2} M BaCl_2$ and $2 \times 10^{-2} M NaF$

pace for rough work

Section – III (Mathematics)

 $\sum_{r=1}^{n} \left(\frac{{}^{n}C_{r}}{r+1} - \frac{{}^{n+1}C_{r}}{n+1} \right) \text{ equals to}$ 1. (A) $\frac{1}{n+1}$ (B) $\frac{-n}{n+1}$ $(C) \frac{-(n+2)}{n+1}$ (D) $\frac{-1}{n+1}$ 2. Statement -1: If a, b, c are distinct and x, y, z are not all zero, then ax + by + cz = 0bx + cy + az = 0cx + ay + bz = 0Gives $a + b + c \neq 0$ Statement – 2: $a^2 + b^2 + c^2 > ab + bc + ca$ if a. b. c are distinct. (A) Statement-1 is true, Statement-2 is true, but Statement-2 is a correct explanation for Statement-1. (B) Statement-1 is true, Statement-2 is true, but Statement-2 is not a correct explanation for Statement-1. (C) Statement-1 is true, Statement-2 is false. (D) Statement-1 is false, Statement-2 is true. If f(x) is an odd periodic function with period 2, then f(4) equals 3. (A) 0 (B) 2 (C) 4 (D) - 4 The solution of the differential equation $2x\frac{dy}{dx} - y = 3$ represents 4. (A) straight lines (C) parabola (D) ellipse (B) circle If [x] denotes the greatest integer less than or equal to x, then the value of $\int \left[|x-3| \right] dx$ is 5. (D) 8 (A) 1 (B) 2 (C) 4 6. A tower AB leans towards west making an angle α with the vertical. The angular elevation of B, the top most point of the tower, is β , as observed from a point C due east of A at a distance d from A. If the angular elevation of B from a point due east of C is at a distance 2d from C is γ , then (A) $2\tan\alpha = 2\cot\beta - \cot\gamma$ (B) $2\tan\alpha = 3\cot\beta - \cot\gamma$ (C) $\tan \alpha = \cot \beta - \cot \gamma$ (D) None of these Let $f(x) = x^3 + ax^2 + bx + 5\sin^2 x$ be an increasing function in the set of real number R. Then 7. (A) $a^2 - 3b - 15 > 0$ (B) $a^2 - 3b + 15 > 0$ (C) $a^2 - 3b - 15 < 0$ (D) a > 0 and b > 0If $a_1, a_2, a_3,...$ are in H.P. and $f(k) = \sum_{r=1}^n a_r - a_k$, then $\frac{a_1}{f(1)}, \frac{a_2}{f(2)}, \frac{a_3}{f(3)}, ..., \frac{a_n}{f(n)}$ are in 8. (A) A.P. (B) G.P. (D) None of these A problem in mathematics is given to 3 students whose chances of solving individually are $\frac{1}{2}, \frac{1}{2}$ 9. and $\frac{1}{4}$. The probability that the problem will be solved atleast by one is

(A)
$$\frac{1}{4}$$
 (B) $\frac{1}{24}$ (C) $\frac{23}{34}$ (D) $\frac{3}{4}$

10. If $a \in I$ and the equation (x-a)(x-10)+1=0 has integral roots, then the values of a are (A) 6, 8 (B) 8, 10 (C) 10, 12 (D) 8, 12

11. If $2x + \sqrt{6}y = 2$ touches the hyperbola $x^2 - 2y^2 = 4$, then the point of contact is (A) $\left(-2, \sqrt{6}\right)$ (B) $\left(-5, 2\sqrt{6}\right)$ (C) $\left(\frac{1}{2}, \frac{1}{\sqrt{6}}\right)$ (D) $\left(4, -\sqrt{6}\right)$

12. Consider a circle with its centre lying on the focus of the parabola $y^2 = 2 px$ such that it touches the directrix of the parabola. Then a point of intersection of the circle and the parabola is

(A)
$$\left(\frac{p}{2}, p\right)$$
 or $\left(\frac{p}{2}, -p\right)$
(B) $\left(\frac{p}{2}, -\frac{p}{2}\right)$
(C) $\left(-\frac{p}{2}, p\right)$
(D) $\left(-\frac{p}{2}, -\frac{p}{2}\right)$

13. If (a, a^2) falls inside the angle made by the linear equations $y = \frac{x}{2}, x > 0$ and y = 3x, x > 0, then a belongs to

(A)
$$\left(-3, -\frac{1}{2}\right)$$
 (B) $\left(0, \frac{1}{2}\right)$ (C) $(3, \infty)$ (D) $\left(\frac{1}{2}, 3\right)$

- 14. The straight lines whose direction cosines satisfy al + bm + cn = 0, fmn + gnl + hlm = 0 are perpendicular if
 - (A) $\sqrt{af} + \sqrt{bg} + \sqrt{ch} = 0$ (B) $\frac{a^2}{f} + \frac{b^2}{g} + \frac{c^2}{h} = 0$ (C) $\frac{f}{a} + \frac{g}{b} + \frac{h}{c} = 0$ (D) $a^2 f + b^2 g + c^2 h = 0$

15. Domain of derivative of the function $f(x) = |\sin^{-1} (2x^2 - 1)|$ is

16.

(A) [-1, 1]
(B) [-1, 1] ~
$$\left\{0, \pm \frac{1}{\sqrt{2}}\right\}$$

(C) [-1, 1] ~ $\{0\}$
(D) [-1, 1] ~ $\left\{\pm \frac{1}{\sqrt{2}}\right\}$
 $\sum_{r=1}^{\infty} \tan^{-1}\left(\frac{1}{r^2 + 5r + 7}\right)$ equals to
(A) $\tan^{-1} 3$
(B) $\frac{\pi}{4}$
(C) $\sin^{-1}\frac{1}{\sqrt{10}}$
(D) $\cot^{-1} 2$

17. If P_1 , P_2 , P_3 be the lengths of perpendiculars from the vertices of the triangle ABC to the opposite sides, then

(A)
$$P_1P_2P_3 = abc$$

(B) $P_1P_2P_3 = 8R^{\circ}$
(C) $P_1P_2P_3 = \frac{a^2b^2c^2}{R^3}$
(D) $P_1P_2P_3 = \frac{a^2b^2c^2}{8R^3}$

(where R is circumradius of triangle ABC).

18.	A unit vector is orthog vector is	gonal to 5î+2ĵ+6k̂ an		$\hat{k} \text{ and } \hat{i} - \hat{j} + \hat{k}$, then the
	(A) $\frac{3\hat{j}-\hat{k}}{\sqrt{10}}$		$(B) \ \frac{2\hat{i}+5\hat{j}}{\sqrt{29}}$	
	(C) $\frac{6\hat{i}-5\hat{k}}{\sqrt{61}}$		(D) $\frac{2\hat{i}+2\hat{j}-\hat{k}}{3}$	
19.	If the curve $y = x^2 + bx$ by	+ c touches the straight	line y = x at the point (1,	1), then b and c are given
	(A) –1, 1 (C) 2, 1		(B) –1, 2 (D) 1, 1	
20.	In the sequence 1, 2, 2	, 3, 3, 3, 4, 4, 4, 4, 4,	where n consecutive terr	ms have the value n, the
	150 th term is (A) 17	(B) 16	(C) 18	(D) none of these
21.	The value of $\left\{\frac{5^{2n}}{24}\right\}$, n	$\in N$ where {.} denotes the set of the set	he fractional part of x, is	
	(A) 5/24	(B) 9/24	(C) 1/24	(D) None of these
22.	If z_1 and z_2 are two com	nplex numbers satisfying	the equation $\left \frac{z_1 + z_2}{z_1 - z_2} \right =$	1, then $\frac{z_1}{z_2}$ is a number
	which is (A) Positive real (C) Zero or purely imag	jinary	(B) Negative real(D) None of these	
23.	The number of ways of a (A) 61 (C) 63	switching the network su	ich that the bulb glows is (B) 60 (D) None of these	
~ (a a i a a a ² a	: a ⁿ =1 a a ⁿ a1
24.		_	c 3θ +sin 3θ .sec $3^2\theta$ ++s	-
	(A) tan $3^{n}\theta$ -tan θ	(B) tan $3^{n}\theta$ -ntan θ	(C) tan $3^n \theta$ -tan $3^{n-1} \theta$	$ \begin{array}{c} (D) & -(\tan 3 \ \theta \ -\tan \theta) \\ 2 \end{array} $
25.	The determinate $\begin{bmatrix} \cos C \\ \sin B \\ 0 \end{bmatrix}$	$\begin{array}{c c} \tan A & 0 \\ 0 & -\tan A \\ \sin B & \cos C \end{array}$ has the v	value where A, B, C are a	angles of a triangle
	(A) 0	(B) 1	(C) sinA. sinB	(D) cosA cosB cosC
26.	The equation of the imates (A) $x^2 + y^2 + 32x + 4y + 23$ (C) $x^2 + y^2 + 32x + 4y - 23$	65 = 0	16x - 24y + 183 = 0 by the li (B) $x^2 + y^2 - 32x + 4y + 23$ (D) None of these	ne mirror $4x + 7y + 13 = 0$ is $35 = 0$

27. For hyperbola $\frac{x^2}{\cos^2 \alpha} - \frac{y^2}{\sin^2 \alpha} = 1$, which of the following remains constant with change in ' α '? (A) abscissae of vertices (B) abscissae of foci (C) eccentricity (D) directrix 28. If $f(x) = \begin{cases} \frac{\sin[x]}{[x]}, & [x] \neq 0 \\ 0, & [x] = 0 \end{cases}$ Where, [x] denotes the greatest integer less than or equal to x, then $\lim_{x \to 0} f(x)$ equals: (A) 1 (B) 0 (C) -1 (D) None of these

29. For a real number y, let [y] denote the greatest integer less than or equal to y. Then the function $f(x) = \frac{\tan \pi [(x - \pi)]}{1 + [x]^2}$ is:

(A) discontinuous at some x

- (B) continuous at all x, but the derivative f'(x) does not exist for some x
- (C) f'(x) exists for all x, but the derivative f''(x) does not exist for some x
- (D) f''(x) exists for all x

30.
$$\int \frac{1 + (\sin x)^{2/3}}{1 + (\sin x)^{4/3}} d(\sin x)^{1/3} \text{ is equal to}$$
$$(A) \quad \frac{1}{\sqrt{2}} \frac{(\sin x)^{2/3} - 1}{\sqrt{2}(\sin x)^{1/3}} + c$$
$$(C) \quad \frac{1}{\sqrt{2}} \tan^{-1} \left(\frac{(\sin x)^{1/3} - 1}{\sqrt{2}(\sin x)^{2/3}} \right) + c$$

(B)
$$\frac{1}{\sqrt{2}} \tan^{-1} \left(\frac{(\sin x)^{2/3} - 1}{\sqrt{2}(\sin x)^{1/3}} \right) + c$$

(D) none of these

space for rough work

FIITJEE - JEE (Main)

SAMPLE TEST - 2

Time Allotted: 3 Hours

Maximum Marks: 360

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 (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
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- 8. Use *Blue / Black Ball Point Pen only* for writing particulars / marking responses on *Side-1* and *Side-2* of the Answer Sheet. *Use of pencil is strictly prohibited.*
- 9. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. except the Admit Card inside the examination hall / room.
- 10. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty in the Room / Hall. *However, the candidates are allowed to take away this Test Booklet with them.*
- 11. Do not fold or make any stray marks on the Answer Sheet.

Name of the Candidate (in Capital Letters) :			
Enrolment Number :			
Batch :	Date of Examination :		

Useful Data Chemistry:	
Gas Constant	R = $8.314 \text{ J K}^{-1} \text{ mol}^{-1}$
	= 0.0821 Lit atm K^{-1} mol ⁻¹
	= $1.987 \approx 2 \text{ Cal } \text{K}^{-1} \text{ mol}^{-1}$
Avogadro's Number	-
Planck's Constant	h = 6.626×10^{-34} Js
	$= 6.25 \times 10^{-27} \text{ erg.s}$
1 Faraday	= 96500 Coulomb
1 calorie	= 4.2 Joule
1 amu	$= 1.66 \times 10^{-27} \text{ kg}$
1 eV	$= 1.6 \times 10^{-19} \text{ J}$
Atomic No :	H=1, D=1, Li=3, Na=11, K=19, Rb=37, Cs=55, F=9, Ca=20, He=2, O=8,
	Au=79.
Atomic Masses:	He=4, Mg=24, C=12, O=16, N=14, P=31, Br=80, Cu=63.5, Fe=56,
	Mn=55, Si = 28 Pb=207, u=197, Ag=108, F=19, H=2, Cl=35.5
Useful Data Physics:	
Acceleration due to	gravity $g = 10 \text{ m/s}^2$

Section – I (Physics)

- 1. If $\vec{A} \times \vec{B} = \vec{B} \times \vec{A}$ then the angle between A and B is (A) π (B) $\pi/3$ (C) $\pi/2$ (D) $\pi/4$.
- A ball is projected from the top of a tower of height 40 m with a velocity of 20 m/s at an angle of 30⁰ with horizontal. The ratio of time of flight to the time taken to reach the ground is
 (A) 1 : 1
 (B) 1 : 2
 (C) 2 : 1
 (D) 2 : 3
- 3. A particle starting from the origin (0, 0) moves in a straight line in (x, y) plane. Its coordinates at a later time are ($\sqrt{3}$, 3). The path of the particles makes with the x-axis an angle of. (A) 45° (B) 60° (C) 0° (D) 30°.
- A body of mass 1 kg is projected with a velocity of 10m/s at an angle 60⁰ w.r.t. horizontal ground. The maximum value of gravitational potential energy in its motion is

 (A) 50J
 (B) 25J
 (C) 35J
 (D) 37.5J
- 5. A block is kept on a frictionless inclined surface with angle of inclination ' α ' as shown in fig. the incline is given an acceleration 'a' to keep the block stationary. Then a is equal to



- 6. A stationary particle explodes into two particles of masses m_1 and m_2 , which move in opposite directions, with velocities v_1 and v_2 . The ratio of their kinetic energies (E_1/E_2) is (A) m_2 / m_1 (B) m_1 / m_2 (C) 1 (D) $m_1 v_2/m_2 v_1$
- Two spheres of masses m and M are situated in air and the gravitational force between them is F. the space around the masses is now filled with a liquid of specific gravity 3. The gravitational force will now be.
 (A) 3 F
 (B) F
 (C) F/3
 (D) F/9.
- 8. In the fig. a common emitter configuration on NPN transistor with current gain β = 100 is used. The output voltage of the amplifier will be



9. A particle is moving in a X-Y plane under the action of a force such that its instantaneous momentum $\overline{p} = 3\cos t \hat{i} + 3\sin t \hat{j}$. The instantaneous angle between the force and momentum is



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- 10. A cylinder contains 10 kg of gas at pressure of 10^7 Nm^{-2} , the quantity of gas taken out of the cylinder, if final pressure is $2.5 \times 10^6 \text{ Nm}^{-2}$ (A) 7.5 kg (B) 10.5 kg (C) 5.2 kg (D) none of these.
- 11. A particle of mass M and charge Q moving with velocity \vec{v} describes a circular path of radius R when subjected to a uniform transverse magnetic field of induction B. The work done by the field when the particle completes one full circle is

 $(A)\left(\frac{Mv^2}{R}\right) = 2\pi R$ (B) zero (C) BQ $2\pi R$ (D) BQ $v 2\pi R$

- 12. Two masses A of 0.5 kg and B of 0.3 kg having specific heat capacities of 0.85 J/kg K and 0.9 J/kg K respectively are at temperatures 60°C and 90°C respectively. When connected with each other with a conducting rod, heat will flow from
 - (A) A to B
 - (B) B to A
 - (C) Initially from A to B and then from B to A
 - (D) Heat can't flow
- 13. In an oscillating LC circuit, the maximum charge on the capacitor is Q. The charge on the capacitor, when the energy is stored equally between the electric and magnetic field is (A) Q/2 (B) $Q/\sqrt{2}$ (C) $Q/\sqrt{3}$ (D) Q/3
- 14. In YDSE, two slits are made 5 mm apart and the screen is placed 2 m away. What is the fringe separation when light of wavelength 500 nm is used? (A) 0.002 mm (B) 0.02 mm (B) 0.2 mm (D) 2 mm
- A convex lens is dipped in a liquid whose refractive index is equal to the refractive index of the lens. Then its focal length will
 (A) Become zero
 (B) become infinite
 (C) Reduce
 (D) increase.
- 16. Momentum of a photon of energy 1 MeV in kg ms⁻¹ will be (A) 5×10^{-22} (B) 0.33×10^{6} (C) 7×10^{-24} (D) 10^{-22}
- 17. If a star converts all helium in its core to oxygen then energy released per oxygen nuclei is [Mass of He 4.0026 a.m.u., mass of O 15.9994 a.m.u.]
 (A) 10.24 MeV
 (B) 0
 (C) 7.56 MeV
 (D) 5 MeV
- 18. In the middle of the depletion layer of reverse biased p-n junction, the
 (A) Electric Field is zero
 (B) Potential is maximum
 (C) Electric field is maximum
 (D) Potential is zero
- 19. Consider telecommunication through optical fibres. Which of the following statements is not correct?
 - (A) Optical fibres have extremely low transmission loss
 - (B) Optical fibres may have homogeneous core with a suitable cladding
 - (C) Optical fibres can be graded refractive index
 - (D) Optical fibers are subject to electromagnetic interference from outside
- 20. A tuning fork of frequency 256 Hz produces a faint sound when kept near the mouth of a one end closed cylindrical tube. When water is poured, such that 31 cm of air column exists, the sound becomes loud. The speed of sound in air is (A) 317 ms^{-1} (B) 371ms^{-1} (C) 340 ms^{-1} (D) 332 ms^{-1}
- 21. A Carnot engine takes 3×10^6 cals of heat from a reservoir at 627°C and gives it to a sink at 27°C. the work done by the engine is (A) 4.2×10^6 J (B) 8.4×10^6 J (C) 16.8×10^6 J (D) 3×10^6 J

- A car moving with a speed of 50 km/h, can be stopped by brakes after at least 6 m. if the same car is moving at a speed of 100 km/h, the minimum stopping distance is
 (A) 12 m
 (B) 18 m
 (C) 24 m
 (D) 6 m.
- 23. Three charge are placed at the vertices of an equilateral triangle of side a as shown in Fig. The force experienced by the charge placed at the vertex A in a direction parallel to BC is



space for rough work

Section – II (Chemistry)

1.	What is the equivalent (A) 100.5	weight of $HClO_4$? (B) 50.3	(C) 60.1	(D) 90.5
2.		uires 32.5 ml of 1N HCl t	o dissolve it. What is the	equivalent weight of
	metal? (A) 46	(B) 65	(C) 56	(D) 42
3.	For a p-electron, the or (A) $\sqrt{3}\frac{h}{2\pi}$	bital angular momentum (B) $\frac{h}{2\pi}$	is (C) $\sqrt{2} \frac{h}{2\pi}$	(D) $\frac{h}{\pi}$
4.	Calculate the frequency excites to ground state (A) $2.92 \times 10^{15} \text{ sec}^{-1}$		tted when the electron in (C) $3.2 \times 10^{10} \text{ sec}^{-1}$	n = 3 in H-atom de- (D) $1.9 \times 10^{10} \text{ sec}^{-1}$
F			()	(b) 1.9×10 see
5.	(A) Co^{2+}	ving transition metal ions (B) <i>Ni</i> ²⁺	(C) Cu^{2+}	(D) Zn^{2+}
6.	Which of the following it (A) H_3BO_3	s expected not to exist? (B) NaBO ₂	(C) <i>B</i> ₂ <i>N</i>	(D) $B_2 H_6$
7.	The hydrogen bond is s (A) $O - H S$		(C) $F - H F$	(D) <i>F</i> – <i>H</i> – – – <i>O</i>
8.	What is the increasing (A) $MgO < CaO < SrO$ (C) $BaO < SrO < CaO$	e < BaO	of MgO, CaO, SrO and (B) $CaO < MgO < BaO$ (D) $BaO < MgO < CaO$	O < SrO
9.	Which is the correct or (A) $P_1 > P_2 > P_3$ (C) $P_3 > P_2 > P_1$	der for the graph below?	(B) $P_2 > P_1 > P_3$ (D) $P_3 > P_1 > P_2$	P_1 P_2 P_3 $T \rightarrow$
10.	Which is the incorrect (A) P V \rightarrow	graph? (B) P P V→	(C) ↓ ↓ T →	(D) ↑ PV V →

- 11.HI was heated in sealed tube at 400°C till the equilibrium was reached. HI was found to be 22%
decomposed. The equilibrium constant for dissociation is
(A) 1.99(B) 0.0199(C) 0.0796(D) 0.282
- 12. When sulphur in the form of S_8 is heated at 900K, the initial pressure of 1 atm falls by 30% at equilibrium. This is because of conversion of same S_8 to S_2 . Find the value of equilibrium constant for this reaction?

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	(A) 2.55 atm ³	(B) 2.96 atm ³	(C) 0.71 atm ³	(D) 3.4 atm ³		
13.	 B. Catalyst is a substance which (A) Supplies energy to the reaction (B) Increases the equilibrium concentration of the product (C) Changes the equilibrium constant of the reaction (D) Lowers the activation energy barrier 					
14.	What is the pH of solut	ion which have 0.1M NH	I $_3$ and 0.05 M $ NH_4Cl$ G	iven that		
	$K_b(NH_3) = 10^{-5}$					
	(A) 9.3	(B) 5	(C) 4.74	(D) 8.26		
15.	DDT is an example of (A) Fungicide	(B) Herbicide	(C) Insecticide	(D) Analgesic		
16.			to 80 kJmol ⁻¹ . At what te be equal to rate of reac			
	(A) 200 K	(B) 400 K	(C) 625 K	(D) None of these		
17.	The increasing order of	f stability of the following	free radicals is			
	(A) $(CH_3)_2 \dot{C} H < (CH_3)_3 \dot{C} < (C_6H_5)_2 \dot{C} H < (C_6H_5)_3 \dot{C}$					
	(B) $(C_6H_5)_3 \dot{C} < (C_6H_5)_2 \dot{C}H < (CH_3)_3 \dot{C} < (CH_3)_2 \dot{C}H$					
	(C) $(C_6H_5)_2 C H < (C_6H_5)_3 C < (CH_3)_3 C < (CH_3)_2 C H$					
	(D) $(CH_3)_2 \dot{C} H < (C$	$(H_3)_3 \dot{C} < (C_6 H_5)_3 \dot{C} <$	$(C_6H_5)_2\dot{C}H$			
18.	Among the following st	ructures which is not a c	orrect resonance form:			
	(A) H_2C^+	— <u>;;</u> -	(B) H ₂ C==_N=	—ö.		
	$ \begin{array}{c} (A) & H_2C^+ & \ddot{N} \\ & & \\ $		ĊH ₃	3		
	(C) $H_2C = N^+$	— <u>ö:</u>	(D) H ₂ C ⁻ −−−N [±]	— <u>ö</u>		
	L CH ₃		L CH	3		
19.	The reaction					
	H ₃ C CH ₃ +	$SOCl_2 \longrightarrow H_3C$	$CI + SO_2 + HC$	1		
	Proceeds by the (A) S_N i		(C) S _E 2	(D) S _E 1		
20.	-		phates of alkali metal is i metals are soluble in w			
		ali metals excent <i>Li SC</i>				

- (B) All sulphates of alkali metals except Li_2SO_4 forms alum.
- (C) The sulphates of alkali metals cannot be hydrolysed
- (D) All of these

(A) conc. H_2SO_4

21. The compound which undergoes S_N1 reaction most rapidly is



(C) quick lime

(D) None of these

(B) P_2O_5

30. What is the relation between the following compounds?



Section – III (Mathematics)

- 1. A man standing on a horizontal plane, observes the angle of elevation of the top of a tower to be α . After walking a distance equal to double the height of the tower, the angle of the elevation becomes 2α , then α is equal to
- (B) $\frac{\pi}{6}$ (D) $\frac{\pi}{18}$ (A) $\frac{\pi}{2}$ (C) $\frac{\pi}{12}$ If $\cos^{-1} x - \cos^{-1} \frac{y}{2} = \alpha$, then 2. $4x^2 - 4xy\cos\alpha + y^2$ is equal to: (A) $-4\sin^2\alpha$ (B) $4\sin^2\alpha$ (C) 4 (D) $2\sin 2\alpha$ $\sin\left|2\cos^{-1}\left(-\frac{3}{5}\right)\right|$ is equal to 3. (B) $\frac{24}{25}$ (C) $\frac{4}{5}$ 24 (A) $\frac{6}{25}$ (D) 25 If $\cos \alpha = \frac{12}{13}$, $\cos \beta = \frac{3}{5}$, $\cos \gamma = \frac{63}{65}$, then $\cos(\alpha + \beta + \gamma)$ is: 4. (A) 1 (D) 0 (B) 2 (C) 3 5. India plays two matches each with West Indies and Australia. In any match probabilities of India getting points 0, 1 and 2 are 0.45, 0.05 and 0.5 respectively. Assuming that the outcomes are independent, the probability of India getting at least 7 points is (A) 0.8750 (B) 0.0875 (C) 0.0625 (D) 0.0250 The equation of the plane through intersection of planes x + 2y + 3z = 4 and 2x + y - z = -5 and 6. perpendicular to the plane 5x+3y+6z+8=0, is (A) 7x - 2y + 3z + 81 = 0(B) 23x+14y-9z+48=0(C) 23x + 14y - 9z + 48 = 0(D) 51x + 15y - 50z + 173 = 0Image of the point A(1,6,3) in the line $\frac{x}{1} = \frac{y-1}{2} = \frac{z-2}{3}$ is 7. (B) (2,5,7) (A) (7,1,0) (C) (1,0,7) (D) None of these If $\begin{bmatrix} 2\vec{a} + 4\vec{b} \ \vec{c} \ \vec{d} \end{bmatrix} = \lambda \begin{bmatrix} \vec{a} \ \vec{c} \ \vec{d} \end{bmatrix} + \mu \begin{bmatrix} \vec{b} \ \vec{c} \ \vec{d} \end{bmatrix}$, then $\lambda + \mu =$ (A) 6 (B) -6 (C) 8. (C) 10 (D) 8 9. If $\vec{x} + \vec{y} = \vec{a}$, $\vec{x} \times \vec{y} = \vec{b}$ and $\vec{x} \cdot \vec{a} = 1$, then (A) $\vec{x} = \frac{\vec{a} + \vec{a} \times \vec{b}}{r^2}, y = \frac{(a^2 - 1)\vec{a} - (\vec{a} \times b)}{a^2}$ (B) $\vec{x} = \frac{\vec{a} - \left(\vec{a} \times \vec{b}\right)}{2}, y = \frac{a^2 + \left(\vec{a} \times \vec{b}\right)}{2}$ (C) $\vec{x} = \frac{\vec{b} + (\vec{a} \times \vec{b})}{c^2}$, y can have any value (D) $\vec{y} = \frac{(b^2 - 1)\vec{b} - \vec{a} \times \vec{b}}{a^2}$, y can have any value The product of the perpendicular and drawn from any point on a hyperbola to its asymptotes is 10. (B) $\frac{ab}{a^2+b^2}$ (C) $\frac{a^2b^2}{a^2+b^2}$ (D) $\frac{a^2+b^2}{a^2b^2}$ (A) $\frac{ab}{\sqrt{a}+\sqrt{b}}$ The number of values of c such that the straight line y = 4x + c touches the curve $\frac{x^2}{4} + y^2 = 1$ is 11.

(A) 0 (B) 1 (C) 2

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(D) infinite

12. If
$$\frac{x}{a} + \frac{y}{b} = \sqrt{2}$$
 touches the ellipse $\frac{x^2}{a} + \frac{y^2}{b^2} = 1$, then its eccentric angle 0 is equal to
(A) 0° (B) 90° (C) 45° (D) 60°
13. The latus rectum of a parabola whose focal chord is PSQ such that SP = 3 and SQ = 2 is given by
(A) $\frac{24}{5}$ (B) $\frac{12}{5}$ (C) $\frac{6}{5}$ (D) None of these
14. Let PQ and RS be tangent at the extremities of the diameter PR of a circle of radius r. If PS and RQ intersect at a point X on the circumference of the circle, then 2r equals
(A) $\sqrt{PQ.RS}$ (B) $\frac{PQ+RS}{2}$ (C) $\frac{2PQ.RS}{PQ+RS}$ (D) $\sqrt{\frac{PQ^2+RS^2}{2}}$
15. The line of the system $a(2x - y - 2) + b(x - 3y + 4) = 0$ situated farthest from the point (1, 1) is
(A) $x + y - 4 = 0$ (B) $x + 2y - 6 = 0$ (C) $2x + y - 6 = 0$ (D) None of these
16. Solution of the equation
 $x\frac{dy}{dx} = y + x \tan \frac{y}{x}$ is
(A) $\sin \frac{x}{y} - Cx$ (B) $\sin \frac{y}{x} - Cx$ (C) $\sin \frac{x}{y} - Cy$ (D) $\sin \frac{y}{x} - Cy$
17. The solution of the equation
 $(x^2 - xy)dy = (xy + y^2)dx$ is
(A) $xy - Ce^{-y/x}$ (B) $xy - Ce^{-y/x}$ (C) $yx^2 - Ce^{1/x}$ (D) None of these
18. $\int_{0}^{x^2} \frac{\sqrt{\sin x}}{\sqrt{\sin x} + \sqrt{\cos x}} dx =$
(A) $\frac{\pi}{4}$ (B) $\frac{\pi}{2}$ (C) 10 (D) 1
19. $\int_{1}^{10} \frac{\sin x}{(A + x)} \sin (A + 20)$ (B) 8 (C) 10 (D) 18
20. $\int \frac{dx}{\sqrt{1-x^2}}$ is equal to
(A) $\frac{1}{3} \log \left(\frac{\sqrt{1-x^2}}{\sqrt{1-x^2}} + c$ (D) $\frac{1}{3} \log \left(\frac{\sqrt{1-x^2}}{\sqrt{1-x^2}} + c$
21. If $f(x) = \frac{1}{8} \log x - bx + x^2, x > 0$, where $b \ge 0$ is a constant, then
(A) $f(x)$ has local minimum at $x = \frac{1}{4}$ for $b = 1$

(B) f(x) has no extremum for $0 \le b < 1$

	(C) $f(x)$ has a local n	ninimum at $x = \frac{b - \sqrt{b^2 - 1}}{4}$	$\frac{1}{1}, b > 1$	
		naximum at $x = \frac{b + \sqrt{b^2 - 4}}{4}$		
22.	$y = 2x + \cot^{-1} x + \log\left(\sqrt{1}\right)$	$(+x^2-x)$, then y		
	(A) increases in $[0,\infty)$	only	(B) decreases in $[0 \infty)$	
	(C) neither increases r	nor decreases in $[0 \infty)$	(D) increases in $(-\infty, \infty)$))
23.	$\lim_{x \to 0} \left[1^{\frac{1}{\sin^2 x}} + 2^{\frac{1}{\sin^2 x}} + \dots + n \right]$	$\frac{1}{\sin^2 x} \int_{0}^{\sin^2 x} =$		
	(A) ∞	(B) 0	(C) $\frac{(n+1)}{2}$	(D) n
			2	
24.	$C_0^2 - C_1^2 + C_2^2 \dots (-1) C_n^2,$	where n is an even integ		
	(A) ${}^{2n}C_n$	(B) $(-1)^{n} C_n$	(C) $(-1)^{n-2n}C_{n-1}$	(D) None of these
25.			$(11)^{1915} - (23)^{1225}$ is equal (C) 2	
	(A) 0	(B) 1	(C) 2	(D) None of these
26.	The number of parallel another set of three pa		ed from a set of four para	allel lines intersecting
	(A) 6	(B) 10	(C) 12	(D) None of these
27.	The number of triangle (A) 105	s which can be formed fr (B) 210	rom 12 points out of whic (C) 175	h 7 are collinear is (D) 185
28.	One root of the equation	on	>	
	3x-8 3 3			
	$\begin{vmatrix} 3 & 3x - 8 & 3 \\ 3 & 3 & 3x - 8 \end{vmatrix} =$	0 is which of the following	ng?	
	(A) $\frac{8}{3}$	(B) $\frac{2}{3}$	(C) $\frac{1}{3}$	(D) $\frac{16}{3}$.
29.	If α is a complex numl	ber such that $\alpha^2 + \alpha + 1 =$	$0\text{, then }\alpha^{\scriptscriptstyle 31}$ is	
	(A) α	(B) α ²	(C) 0	(D) 1
30.	The domain of			
	$f(x) = \cot^{-1} \frac{x}{\sqrt{x^2 - \left[x^2\right]}}$	$, x \in R$ is		
	(A) <i>R</i>	(B) $R - \{0\}$	(C) $R - \left\{ \pm \sqrt{n}, n \in N \right\}$	(D) None of these.

space for rough work

FIITJEE - JEE (Main)

SAMPLE TEST - 3

Time Allotted: 3 Hours

Maximum Marks: 360

- Do not open this Test Booklet until you are asked to do so.
- Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.

Important Instructions:

- 1. Immediately fill in the particulars on this page of the Test Booklet with Blue / Black Ball Point Pen. Use of pencil is strictly prohibited.
- 2. The Answer Sheet is kept inside this Test Booklet. When you are directed to open the Test Booklet, take out the Answer Sheet and fill in the particulars carefully.
- 3. The test is of **3 hours** duration.
- 4. The Test Booklet consists of 90 questions. The maximum marks are 360.
- 5. There are *three* sections in the question paper I, II, III consisting of **Physics**, **Chemistry** and **Mathematics** having 30 questions in each section of equal weightage. Each question is allotted **4** (four) marks for correct response.
- Candidates will be awarded marks as stated above in instruction No.5 for correct response of each question.
 (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
- 7. There is only one correct response for each question. Filling up more than one response in any question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction 6 above.
- 8. Use *Blue / Black Ball Point Pen only* for writing particulars / marking responses on *Side-1* and *Side-2* of the Answer Sheet. *Use of pencil is strictly prohibited.*
- 9. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. except the Admit Card inside the examination hall / room.
- 10. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty in the Room / Hall. *However, the candidates are allowed to take away this Test Booklet with them.*
- 11. Do not fold or make any stray marks on the Answer Sheet.

Name of the Candidate (in Capital Letters) :		
Enrolment Number :		
Batch :	Date of Examination :	

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Useful Data Chemistry:		
Gas Constant	R =	8.314 J K ⁻¹ mol ⁻¹
	=	0.0821 Lit atm K ⁻¹ mol ⁻¹
	=	$1.987 \approx 2 \text{ Cal K}^{-1} \text{ mol}^{-1}$
Avogadro's Numbe	r N _a =	6.023×10^{23}
Planck's Constant	h =	$6.626 \times 10^{-34} \text{ Js}$
	=	6.25 x 10 ⁻²⁷ erg.s
1 Faraday	=	96500 Coulomb
1 calorie	=	4.2 Joule
1 amu	=	1.66 x 10 ⁻²⁷ kg
1 eV	=	1.6 x 10 ⁻¹⁹ J
Atomic No :	H=1, D=1	, Li=3, Na=11, K=19, Rb=37, Cs=55, F=9, Ca=20, He=2, O=8,
	Au=79.	
Atomic Masses:	He=4, Mg	=24, C=12, O=16, N=14, P=31, Br=80, Cu=63.5, Fe=56,
	Mn=55, S	i = 28 Pb=207, u=197, Ag=108, F=19, H=2, Cl=35.5
Useful Data Physics:		
Acceleration due to	gravity g =	= 10 m/s ²

Section – I (Physics)

1. If L, R, C, V respectively represent inductance, resistance, capacitance and potential difference, then the dimensions of $\frac{L}{RCV}$ are the same as those of (B) $\frac{1}{Current}$ (D) $\frac{1}{\text{Charge}}$ (C) Charge (A) Current 2 A body starting from rest with acceleration α which varies with time t according as $\alpha = At + B$ where A and B are the constants. The velocity of the particle after time t is: (B) $\frac{At^2}{2} + 2Bt$ (C) $\frac{At^2}{2} + Bt$ (D) $At^2 + \frac{Bt}{2}$ (A) (A+B)tA particle starts from origin at t = 0 and moves in the x - y plane with a constant acceleration 6 3. m/s² in y-direction. The equation of motion is $y = \frac{x^2}{3}$, then its velocity component along xdirection at x = 2 is (in m/s) (A) $\frac{3}{2}$ (B) 3 (C) 6 (D) 2 From the surface of a certain planet a body is projected with a certain velocity at a certain angle 4. from the horizontal surface. The horizontal and vertical displacements x and y are given by $x = 10\sqrt{3}t$ and $y = 10t - t^{2}$ where t is the time in second and x and y are in meter. The magnitude and direction of the velocity of projection are: (A) 10 ms^{-1} at 30° from the horizontal (C) 10 ms^{-1} at 60° from the horizontal (B) 20 ms^{-1} at 60° from the horizontal (D) 20 ms^{-1} at 30° from the horizontal From the top of a tower of height 40m, a ball is projected upwards with a speed of 20 ms⁻¹ at an 5. angle of 30° to the horizontal. If $g = 10ms^{-2}$, after how long will the ball hit the ground? (C) 3 s (A) 1 s (B) 2 s (D) 4 s A particle is moving in a circle of radius R. At t = 0 its speed is zero and during its motion speed 6. varies as $v = 2s^2$ where s is the distance travelled. The angle made by acceleration vector with radial direction after one revolution. (C) $\tan^{-1}\left(\frac{1}{\pi}\right)$ (A) $\frac{\pi}{2}$ (B) $\frac{\pi}{4}$ (D) π The displacement x of a particle at any instant is related to its velocity as $v = \sqrt{2x+9}$. Its 7. acceleration and initial velocity are: (A) 1 unit and 3 unit (B) 3 unit and 9 unit (C) 9 unit and 3 unit (D) 2 unit and 9 unit 8. A body is moving down a long inclined plane of angle of inclination θ . The coefficient of friction between the body and the plane varies as $\mu = 0.5x$, where x is the distance moved down the plane. The body will have the maximum velocity when it has travelled a distance x given by (B) $x = \frac{2}{\tan \theta}$ (C) $= \sqrt{2} \cot \theta$ (D) $x = \frac{\sqrt{2}}{\cot \theta}$ (A) $x = 2 \tan \theta$

9. At the instant t = 0 a force F = kt(k is a constant) acts on a small body of

mass m resting on a smooth horizontal surface. The time, when body leaves the surface is: (A) $mgk \sin \alpha$ (B) $k \sin \alpha / mg$

(C) $mg\sin\alpha / k$ (D) $mg / k \sin \alpha$

10. If the potential energy of a gas molecule is $U = \frac{M}{r^6} - \frac{N}{r^{12}},$ M and N being positive constants, then the potential energy at equilibrium must be: (C) $N^2 / 4M$ (B) $M^2/4N$ (D) $MN^2/4$ (A) zero

A body is moving up an inclined plane of angle θ with an initial kinetic energy E. The coefficient 11. of friction between the plane and the body is μ . The work done against friction before the body comes to rest is

(A)
$$\frac{\mu\cos\theta}{E\cos\theta + \sin\theta}$$
 (B) $\mu E\cos\theta$ (C) $\frac{\mu E\cos\theta}{\mu\cos\theta - \sin\theta}$ (D) $\frac{\mu E\cos\theta}{\mu\cos\theta + \sin\theta}$,

12. A body P strikes another body Q of mass that is p times that of body P and moving with a velocity that is $\frac{1}{2}$ of the velocity of body P. If body P comes to rest, the coefficient of restitution is

(A)
$$\frac{p+q}{p-q}$$
 (B) $\frac{p-q}{q(p-1)}$ (C) $\frac{p-q}{p(q-1)}$ (D) $\frac{p+q}{p(q-1)}$

A solid cylinder is rolling without slipping down an incline of inclination θ . Minimum coefficient of 13. friction so that the cylinder does not slip on the incline is

(C) $\frac{\tan\theta}{3}$ $\tan \theta$ (A) $\tan \theta$ (B)



m



$$A \frac{15}{32}MR^{2}$$

$$C \frac{13}{32}MR^{2}$$





15. The centres of a ring of mass m and a sphere of mass M of equal radius R, are at a distance $\sqrt{8R}$ apart as shown in Fig The force of attraction between the ring and the sphere is

(A)
$$\frac{2\sqrt{2}}{27} \frac{GmM}{R^2}$$

(B) $\frac{GmM}{8Rh2}$
(C) $\frac{GmM}{9R^2}$
(D) $\frac{\sqrt{2}}{9} \frac{GmM}{9R^2}$



- 16. The magnitude of gravitational force on a particle of mass m placed at a distance x from the rod of mass M and length *l* as shown in the figure is:
 - (A) $\frac{GMm}{l+x^2}$ (B) $\frac{GMm}{l(l+x)}$ (C) $\frac{GMm}{l^2+x}$ (D) $\frac{GMm}{x(l+x)}$

17. Water from a tap emerges vertically downwards with an initial speed of 1 ms⁻¹. The crosssectional area of top is $10^{-4}m^2$. Assume that the pressure is constant throughout the stream of water and that the flow is steady. The cross-sectional area of the stream 0.15 m below the top is (take g = 10ms⁻²) (A) $5 \times 10^{-4}m^2$ (B) $1 \times 10^{-5}m^2$ (C) $5 \times 10^{-5}m^2$ (D) $2 \times 10^{-5}m^2$

18. One end of a thermally insulated rod is kept at a temperature T_1 and the other at T_2 . The rod is composed of two sections of lengths I_1 and I_2 and thermal conductivities k_1 and k_2 respectively. The temperature at the interface of the section is

$$\mathsf{A} \left(k_{1}l_{2}T_{1} + k_{2}l_{1}T_{2} \right) / \left(k_{1}l_{1} + k_{2}l_{2} \right)$$

$$\mathsf{B} \left(k_{2}l_{1}T_{1} + k_{1}l_{2}T_{2} \right) / \left(k_{2}l_{1} + k_{1}l_{2} \right)$$

$$\mathsf{C} \left(k_{1}l_{1}T_{1} + k_{2}l_{1}T_{2} \right) / \left(k_{1}l_{2} + k_{2}l_{1} \right)$$

$$\mathsf{D} \left(k_{1}l_{1}T_{1} + k_{2}l_{2}T_{2} \right) / \left(k_{1}l_{1} + k_{2}l_{2} \right)$$

- A body cools from 50.0 °C to 49.9 °C in 5s. How long will it take to cool from 40.0 °C to 39.9 °C? Assume the temperature of the surroundings to be 30.0 °C and Newton's law of cooling to be valid.
 (A) 2.5s
 (B) 10 s
 (C) 20 s
 (D) 5 s
- 20. In the following figures, the block of mass m is slightly displaced from its mean position. The ratio of time periods of oscillations in Fig(i) and Fig(ii) is: (A) 1:2 (C) $3: \sqrt{2}$ (D) 1:1
- 21. Third overtone of a closed organ pipe is in unison with fourth harmonic of an open organ pipe. Find the ratio of lengths of the pipes:
 (A) 1:2
 (B) 3:4
 (C) 5:6
 (D) 7:8
- 22. Two point charges q_1 and q_2 ($q_1/2$) are placed at points A(0, 1) and B(1,0) as shown in the figure. The electric field vector at point P(1,1) makes an angle θ with the x-axis, then the angle θ is: (A) $\tan^{-1}\left(\frac{1}{2}\right)$ (B) $\tan^{-1}\left(\frac{1}{4}\right)$ (C) $\tan^{-1}(1)$ (D) $\tan^{-1}(0)$
- 23. Three point charges 4q, Q and q are placed in straight line of length l at points distant $0, \frac{l}{2}, l$ respectively. If the net force on charge q is zero, the magnitude of the force on charge 4q is

o

(A)
$$\frac{q^2}{\pi\varepsilon_0 l^2}$$
 (B) $\frac{2q^2}{\pi\varepsilon_0 l^2}$ (C) $\frac{3q^2}{\pi\varepsilon_0 l^2}$ (D) $\frac{4q^2}{\pi\varepsilon_0 l^2}$

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A $\frac{5}{2}$

24. The capacitance of a sphere of radius R_1 is increased 3 times when it enclosed by an earthed sphere of radius R_2 . The ratio R_2/R_1 is

(A) 2 (B)
$$\frac{3}{2}$$
 (C) $\frac{4}{3}$ (D) 3

	J	
(A) 5 V		(B) 10 V
(C) 15 V		(D) zero

26. A particle of mass m and charge q moves with a constant velocity v along the positive x direction. It enters a region containing a uniform magnetic field B directed along the negative z direction, extending from x = a to x = b. The minimum value of v required so that the particle can just enter the region x > b is

(A) qbB/m (B) q(b-a)B/m (C) qaB/m

27. A square loop is placed in a uniform magnetic field \vec{B} as shown in figure. The power needed to pull it out of the field with a constant velocity v is proportional to (A) $v^{1/2}$ (B) v(C) v^2 (D) $v^{3/2}$



D $\frac{3}{2}$

(D) q(b+a)B/2m

28. An observer can see through a pin-hole the top end of a thin rod of height h, placed as shown in the figure. The beaker height is 3h and its radius h. when the beaker is filled with a liquid up to a height 2h, he can see the lower end of the rod. Then the refractive index of the liquid is:

 $C\sqrt{\frac{3}{2}}$

A U shaped wire is placed in front of a concave mirror of radius of curvature 20 cm as shown in the figure. The total length of the image of the wire ABCD is nearly:
(A) 2.5 cm
(B) 6 cm
(C) 12.5 cm
(D) 15 cm



30. Interference pattern is obtained with two coherent light sources of intensity ratio n. In the interference pattern, the ratio $\frac{I_{\text{max}} - I_{\text{min}}}{I_{\text{max}} + I_{\text{min}}}$ will be (A) $\frac{\sqrt{n}}{(n+1)}$ (B) $\frac{2\sqrt{n}}{(n+1)}$ (C) $\frac{\sqrt{n}}{(\sqrt{n}+1)^2}$ (D) $\frac{2\sqrt{n}}{(\sqrt{n}+1)^2}$

space for rough work

Section – II (Chemistry)

1.	For a hypothetical reac	tion				
	$AB_{2(g)} + \frac{1}{2}B_{2(g)} \rightleftharpoons AB_{3(g)}; \ \Delta H = -x \ kJ$					
	More AB_3 could be produced by					
	(A) using a catalyst		(B) removing some of	<i>B</i> ₂		
	(C) increasing the temp	perature	(D) increasing the pres	ssure		
2.	The pK_b of CN^- is 4.7. The pH of solution prepared by mixing 2.5 mole of KCN and 2.5 moles					
	of HCN in water and m (A) 10.3	aking the total volume up (B) 9.3	oto 500 ml, is (C) 4.7	(D)8.3		
3.	The pH of a solution of	The pH of a solution of weak base at half neutralization with strong base is 8, K_{b} for base is				
	(A) 1×10 ⁻⁴	(B) 1×10 ⁻⁶	(C) 1×10 ⁻⁸	(D) 1×10 ⁻⁷		
4.	HBr and HI reduce sulphuric acid, HCl can reduce $KMnO_4$ and HF can reduce (A) H_2SO_4 (B) $KMnO_4$ (C) $K_2Cr_2O_7$ (D) None of these					
5.	The correct IUPAC name of the compound given below is:					
	H ₃ C CH ₃ CH ₃					
	CH ₃ (A) 4-Ethyl-3-methyloctane (C) 2, 3-Dimethylheptane		(B) 3-Methyl-4-ethyloctane(D) 5-Ethyl-6-methyloctance.			
6.	The transition metal ior	n with least magnetic mo	ment has the electronic	configuration		
	(A) $3d^4$	(B) 3d ⁹	(C) $3d^2$	(D) $3d^8$		
7.	Potassium ferricyanide (A) 2 ions	on ionization produces (B) 1 ions	(C) 5 ions	(D) 4 ions		
8.	Arrange the following free radicals in order of stability: Benzyl (I), Allyl (II), Methyl (III), Vinyl (IV) (A) $IV > III > II > I$ (B) $I > II > III > IV$ (C) $I > III > IV > II$ (D) $IV > III > I > II$					
9.	A hydrocarbon 'A' with molecular mass 84 gives a single monochloride but four dichlorides on photochemical chlorination. The hydrocarbon 'A' is (A) Cyclopentane (B) Cyclohexane (C) 2, 3-Dimethylbutane (D) Methyl cyclopentane					
10.		ected formula of a comp e shell and element Y co (B) XY ₂				
11.	If one million atoms of a (A) 107 g	silver weight 1.79×10 ⁻¹⁶ g (B) 107.2 g	, the gram atomic mass (C) 107.8 g	of silver is (D) 108.2 g		

12. H_2O_2 can act as reducing as well as oxidising agent. In its reaction with NH₂OH and KIO₄, H_2O_2 is acting as (A) Oxidising agent, reducing agent (B) Reducing agent, oxidising agent (C) Oxidising agent, oxidising agent (D) Reducing agent, reducing agent 13. The blue colour developed when Lassaigne's extract is heated with fresh $FeSO_4$ in presence of alkali, cooled and acidified with dil. H₂SO₄ indicates... and is due to the formation of... (A) $N, Fe_4 \left[Fe(CN)_6 \right]_2$ (B) $N, Na_{A} \left[Fe(CN)_{\epsilon} \right]$ (D) N+S, Fe(CNS), (C) $S, Na_4 \left\lceil Fe(CN)_5 NOS \right\rceil$ If the critical temperature of the gas be $T_c = \frac{8a}{27Rb}$ and T_B is the Boyle's temperature, then which 14. of the following, is the correct relation between T_{C} and $T_{B}?$ (B) $T_C = \frac{27}{4} T_B$ (C) $T_C = \frac{8}{27} T_B$ (D) $T_C = \frac{27}{8} T_B$ (A) $T_{c} = \frac{4}{27}T_{B}$ The composition of a sample of wustite is $Fe_{0.93}O$. What is the percentage of iron present as 15. $F \rho^{3+} ?$ (A) 15% (B) 25% (C) 35% (D) 45% The uncertainty in the position of a dust particle with mass equal to 1 mg (if uncertainty in its 16. velocity is $5.5 \times 10^{-20} m s^{-1}$) is: (A) 9.58 Å (B) 958 Å (C) 9.58 nm (D) None. 17. Which of the following has highest bond dissociation energy? (D) O_2^{2-} (A) O₂ (B) O_2^+ (C) O_2^- Given, that the abundance of isotopes ${}^{54}Fe$, ${}^{56}Fe$ and ${}^{57}Fe$ are 5%, 90% and 5% respectively, 18. the atomic mass of Fe is (B) 55.95 (A) 55.85 (C) 55.75 (D) 55.05 The volume strength of $1.5 \text{ N H}_2\text{O}_2$ solution is 19. (C) 3 (B) 8.4 (D) 8 (A) 4.8 Which of the following has the maximum number of atoms? 20. (A) 24 g of C(12) (B) 56 g of Fe(56) (C) 27 g of Al(27) (D) 108 g of Ag(108) 21. An aqueous solution of 6.3 g oxalic acid dihydrate (H₂C₂O₄.2H₂O₁ is made upto 250 ml. The volume of 0.1 N NaOH required to completely neutralize 10 ml of this solution is (A) 40 ml (B) 10 ml (C) 20 ml (D) 4ml 22. In the standardization of Na₂S₂O₃ using $K_2Cr_2O_7$ by iodometry, the equivalent weight of $K_2Cr_2O_7$ is: (A) $\frac{molecular weight}{2}$ (B) $\frac{molecular weight}{6}$ (D) $\frac{molecular weight}{1}$ (C) $\frac{molecular weight}{3}$

23. What volume of CO₂ at STP will evolve when 1 gm of CaCO₃ reacts with excess of dil HCl? (A) 224 ml (B) 112 ml (C) 58 ml (D) 448 ml

- 24. The vapour pressure of benzene at 280 K is 40 mbar. When urea is mixed at the same temperature, the vapour pressure falls by 8 mbar. The mole fraction of benzene in the solution is (B) 0.25 (C) 0.70 (D) 0.80 (A) 0.20
- Ammonia gas is passed into water, yielding a solution of density 0.93 g/cm³ and containing 25. 18.6% NH₃ by weight. The mass of NH₃ per cc of the solution is (D) 0.68 g/cm^3 (A) 0.17 g/cm³ (B) 0.34 g/cm^3 (C) 0.51 g/cm^3
- A sample of NaHCO₃ + Na₂CO₃ required 20 ml of HCl using phenolphthalein as indicator and 35 26. ml more is required if methyl orange is used as indicator. Then molar ratio of NaHCO₃ to Na₂CO₃ is

(A)
$$\frac{1}{2}$$

Br

CH₃



acetone Product. The product of the reaction is: +Nal





(C) $\frac{3}{4}$

(D)



28. Rank the following in order of decreasing rate of solvolysis with aq. ethanol.

(B) $\frac{2}{3}$

H₃C Br CH₃ Br H₃C—ĊH—CH₂CHMe₂ H₂C= Br (2)(3)(1)(A) 2 > 1 > 3 (B) 1 > 2 > 3 (C) 2 > 3 > 1 (D) 1 > 3 > 2

29. In each of the following groups, which is the strongest nucleophile?

(I) In MeOH	(1) CH_3O^-	(2) 0-	(3) $Me - S^-$
(II) In DMF	(1) <i>OH</i> ⁻	(2) H_2O	(3) $O^ O - H$
(A) I, 3; II, 2	(B) I, 2; II, 1	(C) I, 1; II, 2	(D) I, 3; II, 3

30. Which of the following statement is correct regarding the rate of hydrolysis of the compounds (x) and (y) by $S_{N}1$ reaction?



(C) Both x and y reacts at the same rate

(B) y reacts faster than x

(D) Neither x nor y reacts

pace for rough work

Section – III (Mathematics)							
1.	Range of function $f({x}) = \frac{2^{\{x\}} - 1}{2^{\{x\}} + 1}$ is [Where {x} represent fractional part of x]						
	(A) [-2, 3]	(B) [0, 1/3)	(C) $\left(\frac{1}{3}, \frac{1}{2}\right)$	(D) $[\frac{1}{3}, 1]$			
2.	If $0^{\circ} < x < 90^{\circ}$, $\cos x = \frac{3}{\sqrt{10}}$, then value of $\log_{10} \sin x + \log_{10} \cos x + \log_{10} \tan x$ is						
	(A) 0	(B) 1	(C) -1	(D) None of these			
3.	The function f(x)=[x] cos $\left(\frac{(2x-1)\pi}{2}\right)$ (Where [x] denotes the greatest integer function) is						
	discontinuous. (A) at all x (C) at no x		(B) at all integer point (D) at x which is not in				
4.	Suppose a, b, c are in A.P and a^2 , b^2 , c^2 are in G.P. If a < b < c and a + b + c = 3/2, then value of a is						
	(A) $\frac{1}{2\sqrt{2}}$	(B) $\frac{1}{2\sqrt{3}}$	(C) $\frac{1}{2} - \frac{1}{\sqrt{3}}$	(D) $\frac{1}{2} - \frac{1}{\sqrt{2}}$			
5.	If tan x tan y = a and x + y = $\frac{\pi}{6}$ then tan x and tan y satisfy the equation,						
	(A) $x^2 - \sqrt{3}(1-a)x^2$	+a=0	(B) $\sqrt{3}x^2 - (1-a)x$				
	(C) $x^2 + \sqrt{3}(1-a)x$	-a=0	(D) $\sqrt{3}x^2 + (1+a)x - a^2$	$-a\sqrt{3}=0$			
6.	Let $f(x) = \frac{x}{x-1}$ then $\frac{f(P)}{f(P+1)}$ is equal to						
	(A) $f(P^2)$	(B) $f\left(-\frac{P}{P-1}\right)$	(C) $f\left(\frac{1}{P}\right)$	(D) $f(-P)$			
7.	The points (x_1, y_1) , (x_2, y_2) , (x_1, y_2) and (x_2, y_1) are always (A) Vertices of a rhombus (B) Vertices of a square (C) Con-cyclic (D) Collinear						
8.	8. A closet has 5 pair of shoes. The number of ways in which 4 shoes can be drawn such that will be no complete pair is						
	(A) 80	(B) 160	(C) 200	(D) 240			
9.	Solution of the differe	ntial equation $\frac{dy}{dx} = \frac{x^2}{2}$	$\frac{y}{xy}$ is				
	(A) k $(x^2 + y^2) = x$	(B) k $(x^2 - y^2) + x = 0$	(C) k $(x^2 - y^2) = x$	(D) k $(x^2 + y^2) + x = 0$			
10.	The complex number z =1 + i is rotated through an angle $3\pi/2$ in anticlockwise direction at the origin and stretched by additional $\sqrt{2}$ unit, then the new complex number is						
		ed by additional $\sqrt{2}$ unit (B) $\sqrt{2} - \sqrt{2}$ <i>i</i>		number is (D) None of these			

11. If
$$|z|<4$$
, then $|z+3-4i|$ is less than
(A) 4 (B) 5 (C) 6 (D) 9
12. If $4a+2b+c=0$, then the equation $3ax^2 + 2bx + c=0$ has at least one real root lying between
(A) 0 and 1 (B) 1 and 2 (C) 0 and 2 (D) none of these
13. $\int sec^{r/3} x cos e^{r/3} xdx$ is equal to
(A) $3(tanx)^{1/3}$ (B) $3(catx)^{3/3}$ (C) $-3(tanx)^{-3/3}$ (D) $-3(catx)^{-7/2}$
14. The value of $\int_{0}^{10} {\sqrt{x}} dx$ (where (x) is the fractional part of x) is
(A) 50 (B) 1 (C) 100 (D) None of these
15. If $f(x) = \begin{vmatrix} x & cos x & e^{r^2} \\ tan x & 1 & 2 \end{vmatrix}$ then the value of $\int_{-r^2}^{r/2} f(x) dx$ is equal to
(A) 0 (B) 1 (C) 2 (D) None of these
16. The lines $p(p^2+1)x - y + q = 0$ and $(p^2+1)^2 + (p^2+1) + y + 2q = 0$ are perpendicular to a common line for
(A) No values of p (D) More than two values of p
17. Tangents to the circle $x^2 + y^2 = a^2$ cut the circle $x^2 + y^2 = 2a^2$ at P and Q. The tangents at P and Q to the circle $x^2 + y^2 = 2a^2$ intersect at angle θ , then θ is equal to
(A) $\frac{\pi}{4}$ (B) $\frac{\pi}{6}$ (C) $\frac{\pi}{3}$ (D) $\frac{\pi}{2}$
18. The length of the latus rectum of parabola $4y^2 + 2x - 20y + 17 = 0$ is
(A) 3 (B) 6 (D) 1/3 (C) 1/2 (D) 2/3
20. The tangent at any point P on the hyperbola $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ meets the straight lines $bx - ay = 0$ and $bx + ay = 0$ in the points Q and R respectively. If C is the center of the hyperbola, then CQ.CR is
(A) $a^2 - b^2$ (B) $a^2 + b^2$ (C) ab (D) $\frac{a}{b}$
21. The number of points common to the line $\frac{x+3}{2} = \frac{y-4}{3} = \frac{z+5}{2}$ and the plane $4x - 2y - z = 1$ is
(A) 0 (B) 1 (C) infinite (D) none of these
22. If $\frac{4a}{a} + 5b + 9c = 0$, then $(a \times b) \times [(b \times c) \times (c \times a)]$ is equal to
(A) A vector perpendicular to plane of \overline{a} , \overline{b} and \overline{c}
(B) A scalar quantity
(C) $\overline{0}$

(D) None of these

A unit vector in xy-plane which makes an angle of 45° with the vector $\vec{i} + \vec{j}$ and an angle of 60° 23. with the vector $\vec{3i} - \vec{4j}$ is (B) $\frac{i+j}{\sqrt{2}}$ (C) $\frac{i-j}{\sqrt{2}}$ (A) \vec{i} (D) None of these If x + y = a + b, $x^2 + y^2 = a^2 + b^2$, then $x^n + y^n = a^n + b^n$ is true for (A) $\forall n \in N$ (B) $n \ge 4$ (C) $n \ge 3$ 24. (D) None of these If $1 + |\sin x| + \sin^2 x + |\sin^3 x| + \dots = 4 + 2\sqrt{3}, 0 < x < \pi, x \neq \pi/2$, then 25. (A) $x = \frac{\pi}{6}$ (B) $x = \frac{\pi}{3}, \frac{2\pi}{3}$ (C) $x = \frac{2\pi}{3}, \frac{5\pi}{6}$ (D) $x = \frac{5\pi}{6}$ The value of $\sum_{r=0}^{\infty} \tan^{-1} \left(\frac{1}{1+r+r^2} \right)$ is equal to 26. (D) 0 (A) $\pi/2$ (B) $\pi/4$ (C) π The centre and radius of the circle $z\overline{z} + (1-i)z + (1+i)\overline{z} - 7 = 0$ is 27. (B) 1 + i, 3 $(A) - 1 - i \cdot 3$ (C) 2 – i, 4 (D) 2 + i, 4If |x| < 1 and |y| < 1, then the sum to infinity of the series 28. $(x + y) + (x^{2} + xy + y^{2}) + (x^{3} + x^{2}y + xy^{2} + y^{3}) + \dots$ to ∞ is (A) $\frac{(x+y-xy)}{(1-x)(1-y)}$ (B) $\frac{(x-y+xy)}{(1-x)(1-y)}$ (C) $\frac{(x+y-xy)}{(1+x)(1+y)}$ (D) infinite If the roots of the equation $bx^2 + cx + a = 0$ be imaginary, then for all real values of x, the 29. expression $3b^2x^2 + 6bcx + 2c^2$ is (A) Greater than 4ab (B) Less than 4ab (C) Greater than – 4ab (D) Less than – 4ab The coefficient of x ⁹⁸ in the expansion of $\frac{1+x}{1-x}if|x| < 1$ 30. (A) 1 (D) 0 (B) 2 (C) -1

space for rough work