## **ROLLNO RR 210804**

## 2006 JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY

## IV B.TECH II SEMESTER REGULAR EXAMINATIONS INORGANIC CHEMICAL TECHNOLOGY (CHEMICAL ENGINEERING)

**APRI/MAY 2006** 

TIME: 3HOUR MARK:100

## ANSWER ANY FIVE QUESTIONS ALL QUESTIONS CARRY EQUAL MARKS

- 1.Explain the importance of energy and material balance in a process plant. Why they are computed?
- 2. (a) Mention the engineering problems involved in the crystallization process employed for the water conditioning.
- (b) Explain the Reverse osmosis process employed for the separation of water from brine.
- 3. (a) What are the advantages of single pure oxide refractories over the mixed refractories?
- (b) How Beryllia refractories are made? What are their applications?
- 4. (a) What are the various types of Portland cement?
- (b) What are the functions of the ingredients of cement?
- (c) Describe the various micro constituents and their functions in the Portland cement.
- 5. Write in detail the chemical reactions occur in the furnace of Glass manufacturing with temperatures.
- 6. (a) Describe the production of hydrogen by catalytic reforming of petroleum stocks.
- (b) Describe the production of hydrogen from methanol and steam.
- 7. Explain the production of synthetic ammonia along with a neat flow diagram.
- 8. (a) What is Alum? What are its uses?
- (b) Draw the flow sheet for the production of Alum?

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