From Classroom/Integrated School Programs 7 in Top 20, 23 in Top 100, 54 in Top 300, 106 in Top 500 All India Ranks \& 2314 Students from Classroom /Integrated School Programs \& 3723 Students from All Programs have been Awarded a Rank in JEE (Advanced), 2013
FIIT EE JEE (Main), 2014

## FULL TEST - I

- Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.
- You are not allowed to leave the Examination Hall before the end of the test.


## INSTRUCTIONS

## A. General Instructions

1. Attempt ALL the questions. Answers have to be marked on the OMR sheets.
2. This question paper contains Three Parts.
3. Part-I is Physics, Part-II is Chemistry and Part-III is Mathematics.
4. Each part has only one section: Section-A.
5. Rough spaces are provided for rough work inside the question paper. No additional sheets will be provided for rough work.
6. Blank Papers, clip boards, log tables, slide rule, calculator, cellular phones, pagers and electronic devices, in any form, are not allowed.
B. Filling of OMR Sheet
7. Ensure matching of OMR sheet with the Question paper before you start marking your answers on OMR sheet.
8. On the OMR sheet, darken the appropriate bubble with black pen for each character of your Enrolment No. and write your Name, Test Centre and other details at the designated places.
9. OMR sheet contains alphabets, numerals \& special characters for marking answers.
C. Marking Scheme For All Three Parts.
(i) Section-A ( $\mathbf{0 1}$ to 03 and 10 to 12) contains 6 multiple choice questions which have only one correct answer. Each question carries +8 marks for correct answer and - $\mathbf{2}$ mark for wrong answer.
Section-A ( $\mathbf{0 4}$ to 09 and 13 to 30 ) contains 24 multiple choice questions which have only one correct answer. Each question carries $\mathbf{+ 4}$ marks for correct answer and - $\mathbf{1}$ mark for wrong answer.


## Useful Data

## PHYSICS

| Acceleration due to gravity | $\mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}$ |
| :--- | :--- |
| Planck constant | $\mathrm{h}=6.6 \times 10^{-34} \mathrm{~J}-\mathrm{s}$ |
| Charge of electron | $\mathrm{e}=1.6 \times 10^{-19} \mathrm{C}$ |
| Mass of electron | $\mathrm{m}_{\mathrm{e}}=9.1 \times 10^{-31} \mathrm{~kg}$ |

Permittivity of free space $\quad \varepsilon_{0}=8.85 \times 10^{-12} \mathrm{C}^{2} / \mathrm{N}-\mathrm{m}^{2}$
Density of water $\quad \rho_{\text {water }}=10^{3} \mathrm{~kg} / \mathrm{m}^{3}$
Atmospheric pressure $\quad P_{a}=10^{5} \mathrm{~N} / \mathrm{m}^{2}$
Gas constant
$\mathrm{R}=8.314 \mathrm{~J} \mathrm{~K}^{-1} \mathrm{~mol}^{-1}$

## CHEMISTRY

| Gas Constant $\quad \mathrm{R}$ | $=$ | $8.314 \mathrm{~J} \mathrm{~K}^{-1} \mathrm{~mol}^{-1}$ |
| ---: | :--- | :--- |
|  | $=$ | $0.0821 \mathrm{Lit} \mathrm{atm} \mathrm{K}^{-1} \mathrm{~mol}^{-1}$ |
|  | $=$ | $1.987 \approx 2 \mathrm{Cal} \mathrm{K}^{-1} \mathrm{~mol}^{-1}$ |
| Avogadro's Number $\mathrm{N}_{\mathrm{a}}$ | $=$ | $6.023 \times 10^{23}$ |
| Planck's constant h | $=$ | $6.625 \times 10^{-34} \mathrm{~J} \cdot \mathrm{~s}$ |
|  |  | $6.625 \times 10^{-27} \mathrm{erg} \cdot \mathrm{s}$ |
| 1 Faraday | $=$ | 96500 coulomb |
| 1 calorie | $=$ | 4.2 joule |
| 1 amu | $=$ | $1.66 \times 10^{-27} \mathrm{~kg}$ |
| 1 eV |  | $1.6 \times 10^{-19} \mathrm{~J}$ |

Atomic No: $\mathrm{H}=1, \mathrm{He}=2, \mathrm{Li}=3, \mathrm{Be}=4, \mathrm{~B}=5, \mathrm{C}=6, \mathrm{~N}=7, \mathrm{O}=8$, $\mathrm{N}=9, \mathrm{Na}=11, \mathrm{Mg}=12, \mathrm{Si}=14, \mathrm{Al}=13, \mathrm{P}=15, \mathrm{~S}=16$, $\mathrm{Cl}=17, \mathrm{Ar}=18, \mathrm{~K}=19, \mathrm{Ca}=20, \mathrm{Cr}=24, \mathrm{Mn}=25$, $\mathrm{Fe}=26, \mathrm{Co}=27, \mathrm{Ni}=28, \mathrm{Cu}=29, \mathrm{Zn}=30, \mathrm{As}=33$, $\mathrm{Br}=35, \quad \mathrm{Ag}=47, \quad \mathrm{Sn}=50, \mathrm{l}=53, \quad \mathrm{Xe}=54, \quad \mathrm{Ba}=56$, $\mathrm{Pb}=82, \mathrm{U}=92$.
Atomic masses: $\mathrm{H}=1, \mathrm{He}=4, \mathrm{Li}=7, \mathrm{Be}=9, \mathrm{~B}=11, \mathrm{C}=12, \mathrm{~N}=14, \mathrm{O}=16$, $\mathrm{F}=19, \mathrm{Na}=23, \mathrm{Mg}=24, \mathrm{Al}=27, \mathrm{Si}=28, \mathrm{P}=31, \mathrm{~S}=32$, $\mathrm{Cl}=35.5, \mathrm{~K}=39, \mathrm{Ca}=40, \mathrm{Cr}=52, \mathrm{Mn}=55, \mathrm{Fe}=56, \mathrm{Co}=59$, $\mathrm{Ni}=58.7, \mathrm{Cu}=63.5, \mathrm{Zn}=65.4, \mathrm{As}=75, \mathrm{Br}=80, \mathrm{Ag}=108$, $\mathrm{Sn}=118.7, \mathrm{l}=127, \mathrm{Xe}=131, \mathrm{Ba}=137, \mathrm{~Pb}=207, \mathrm{U}=238$.

## Physics

## PART - I

## SECTION - A

## Single Correct Choice Type

This section contains 30 multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONLY ONE is correct.

1. One quarter of the plate is cut from a square plate as shown in the figure. If ' $M$ ' is the mass of the plate and ' $\ell$ ' is the length of each side, then the moment of inertia of the plate about an axis passing through 'O' and perpendicular to the plate is
(A) $M \ell^{2} / 8$
(B) $3 M \ell^{2} / 4$
(C) $M \ell^{2} / 3$
(D) $3 \mathrm{M} \ell^{2}$

2. A block is suspended by an ideal spring constant $K$. If the block is pulled down by constant force F and if maximum displacement of block from it's initial position of rest is z , then
(A) $z=F / K$
(B) $z=2 F / K$
(C) work done by force F is equal to 2 Fz .
(D) increase in potential energy of the spring is $\frac{1}{2} \mathrm{Kz}^{2}$
3. A particle starts from rest and moves with an acceleration of $a=\{2+|t-2|\} \mathrm{m} / \mathrm{s}^{2}$, the velocity of the particle at $t=4 \mathrm{sec}$ is
(A) $2 \mathrm{~m} / \mathrm{s}$
(B) $4 \mathrm{~m} / \mathrm{s}$
(C) zero
(D) $12 \mathrm{~m} / \mathrm{s}$.
4. A large tank filled with water of density $\rho$ and surface tension $\gamma$ is being siphoned out using a glass capillary tube of cross-sectional radius $r$. If the siphon action starts without any external agent, then (angle of contact is zero).
(A) $\mathrm{H}<\frac{\gamma}{\rho g \mathrm{r}}$
(B) $\mathrm{H}<\frac{2 \gamma}{\rho \mathrm{gr}}$
(C) $\mathrm{H}-\mathrm{h}<\frac{2 \gamma}{\rho g \mathrm{~g}}$
(D) $\sqrt{\mathrm{H}^{2}-\mathrm{h}^{2}}<\frac{\gamma}{\rho g r}$

5. A copper rod and a steel rod of equal cross-sections and lengths (L) are joined side by side and connected between two heat baths as shown in the figure. If heat flows through
 them from $x=0$ to $x=2 L$ at a steady rate, and conductivities of the metals are $\mathrm{K}_{\mathrm{cu}}$ \& $\mathrm{K}_{\text {steel }}\left(\mathrm{K}_{\mathrm{cu}}>\mathrm{K}_{\text {steel }}\right)$, then the temperature varies as: (convection and radiation are negligible)
(A)

(B)

(C)

(D)

6. A non-conducting rod $A B$ of length / has a total charge q . The rod is rotated about an axis passing through its center of mass with a constant angular velocity $\omega$ as shown in the figure. The magnetic moment of the rod is
(A) $\frac{\left.\mathrm{q} \omega\right|^{2}}{12}$
(B) $\frac{\left.q \omega\right|^{2}}{3}$
(C) $\frac{\left.q \omega\right|^{2}}{24}$
(D) $\frac{\left.q \omega\right|^{2}}{6}$

7. The figure shows a potentiometer arrangement. $D$ is the driving cell, $C$ is the cell whose e.m.f. is to be determined. $A B$ is the potentiometer wire and $G$ is a galvanometer. $J$ is a sliding contact which can touch any point on $A B$. Which of the following are essential conditions for obtaining balance?
(A) The e.m.f. of $D$ must be greater than the e.m.f. of $C$
(B) The positive terminals of $D$ and $C$ both must be joined to $A$
(C) The resistance of $G$ must be less than the resistance of $A B$
(D) None of these
8. Three capacitors each having capacitance $C=2 \mu F$ are connected with a battery of e.m.f. 30 V as shown in the figure. When the switch $S$ is closed. The amount of charge flown through the battery is
(A) $20 \mu \mathrm{C}$
(B) $24 \mu \mathrm{C}$
(C) $10 \mu \mathrm{C}$
(D) $40 \mu \mathrm{C}$

9. A conducting rod of length $/$ is hinged at point $O$. It is free to rotate in a vertical plane. There exists a uniform magnetic field $\vec{B}$ in horizontal direction. The rod is released from the position shown in the figure. The potential difference between the two ends of the rod when it rotates through an angle $\theta$ is proportional to:
(A) $\ell^{2}$
(B) $\ell^{3 / 2}$
(C) $\sin \theta$
(D) None of these

10. If $\vec{a}$ and $\vec{b}$ are two non parallel vectors then $[\vec{a} \times(\vec{a} \times \vec{b})] .[\vec{a} \times \vec{b}]$ is equal to
(A) $a^{3} b^{2}$
(B) $a^{2}(\vec{a} \cdot \vec{b})$
(C) zero
(D) None of these
11. A boat goes downstream for half an hour and then goes upstream for half an hour. The total distance travelled by the boat in the ground frame for this is 20 km . li is known that speed of the boat relative to the river for the whole trip was constant and greater than the speed of the river. The distance travelled by the boat in the frame of the river for this is
(A) zero
(B) 20 km
(C) 10 km
(D) can't be determined
12. A particle of mass $m$ is attached to a rod of length $L$ and it rotates in a circle with a constant angular velocity $\omega$. An observer $P$ is rigidly fixed on the rod at a distance $L / 2$ from the centre. The acceleration of $m$ and the pseudo
 force on $m$ from the frame of reference of $P$ must be respectively
(A) zero, zero
(B) zero, $m \omega^{2} \frac{L}{2}$
(C) $\omega^{2} \frac{L}{2}, m \omega^{2} \frac{L}{2}$
(D) zero, $m \omega^{2} \mathrm{~L}$
13. For a certain organ pipe open at both ends, the successive resonance frequencies are obtained at 510,680 and 850 Hz . The velocity of sound in air is $340 \mathrm{~m} / \mathrm{s}$. The length of the pipe must be
(A) 2 m
(B) 0.5 m
(C) m
(D) 0.25 m
14. A short linear object of length 1 mm lies along the axis of a biconvex lens of focal length 5 cm at a distance of 15 cm . The length of the image will be approximately
(A) $1 / 16 \mathrm{~mm}$
(B) $1 / 4 \mathrm{~mm}$
(C) $1 / 2 \mathrm{~mm}$
(D) $1 / 8 \mathrm{~mm}$
15. The left plate of the capacitor shown in the figure above carries a charge $+Q$ while the right plate is uncharged at $t=0$. The total charge on the right plate after closing the switch will be
(A) $\frac{Q}{2}+C \varepsilon$
(B) $\frac{Q}{2}-C \varepsilon$
(C) $-\frac{Q}{2}$
(D) $-\mathrm{C} \varepsilon$

16. A small sphere and a big sphere are released from rest with a very small gap from height h as shown in the figure. The mass of bigger sphere is very large as compared to mass of smaller sphere the height from the point of collision of smaller sphere with the bigger sphere to which the smaller sphere will rise if all the collisions are elastic
(A) 2 h
(B) 4 h
(C) 6 h
(D) 9 h

17. The distance between two parallel plates of a capacitor is a. A conductor of thickness $b(b<a)$ is inserted between the plates as shown in the figure. The variation of effective capacitance between the surfaces of conductor and plate as a function of the distance $(x)$ is best represented by

(A)

(B)

(C)

(D)

18. Three blocks A, B and C having masses $m \mathrm{~kg} 2 \mathrm{~kg}$ and 3 kg are attached by massless strings and ideal pulleys as shown in the figure. When the system is released from rest if the block ' $A$ ' remains stationary, the mass of block ' $A$ ' is
(A) 2.2 kg
(B) 2.6 kg
(C) 2.4 kg
(D) 4.8 kg

19. A block of mass 2 kg is attached to one end of a massless rod of length $\frac{1}{\pi} \mathrm{~m}$. The rod is fixed to a horizontal plane at the other end such that the block and rod are free to revolve on a horizontal plane. The coefficient of friction between the block and surface is 0.1 . Block is made to rotate with uniform speed by applying a constant external force in tangential direction on the block. The work done by external force when the rod rotates by $90^{\circ}$ is
(A) 0
(B) 10 joule
(C) $\frac{\pi}{2}$ joule
(D) 1 joule
20. A solid sphere of radius $R$, and dielectric constant ' $k$ ' has spherical cavity of radius $R / 4$. A point charge $\mathrm{q}_{1}$ is placed in the cavity. Another charge $\mathrm{q}_{2}$ is placed outside the sphere at a distance of $r$ from q . Then Coulombic force of interaction between them is found to be ' $F_{1}$ '. When the same charges are separated by same distance in vacuum then the force of interaction between them is found to be $F_{2}$ then
(A) $F_{1}=F_{2} / k$
(B) $F_{2}=F_{1} / k$
(C) $F_{1} \cdot F_{2}=\frac{1}{k}$
(D) $F_{1}=F_{2}$
21. A solid cylinder is wrapped with a string and placed on an inclined plane as shown in the figure. Then the frictional force acting between cylinder and plane is
(A) zero
(B) 5 mg
(C) $\frac{7 \mathrm{mg}}{2}$
(D) $\frac{\mathrm{mg}}{5}$

22. The relation between $R$ and $r$ (internal resistance of the battery) for which the power consumed in the external part of the circuit is maximum.
(A) $R=r$
(B) $R=\frac{r}{2}$
(C) $R=2 r$
(D) $R=1.5 r$

23. An equilateral triangular loop having a resistance $R$ and length of each side ' $\ell$ ' is placed in a magnetic field which is varying at $\frac{\mathrm{dB}}{\mathrm{dt}}=1 \mathrm{~T} / \mathrm{s}$. The induced current in the loop will be

(A) $\frac{\sqrt{3}}{4} \frac{\ell^{2}}{R}$
(B) $\frac{4}{\sqrt{3}} \frac{\ell^{2}}{\mathrm{R}}$
(C) $\frac{\sqrt{3}}{4} \frac{R}{\ell^{2}}$
(D) $\frac{4}{\sqrt{3}} \frac{\mathrm{R}}{\ell^{2}}$
24. The temperature of a mono-atomic gas in an uniform container of length ' $L$ ' varies linearly from $T_{0}$ to $T_{L}$ as shown in the figure. If the molecular weight of the gas is $\mathrm{M}_{0}$, then the time taken by a wave pulse in travelling from end $A$ to end $B$ is

(A) $\frac{2 \mathrm{~L}}{\left(\sqrt{\mathrm{~T}_{\mathrm{L}}}+\sqrt{\mathrm{T}_{0}}\right)} \sqrt{\frac{3 \mathrm{M}}{5 \mathrm{R}}}$
(B) $\sqrt{\frac{3\left(\mathrm{~T}_{\mathrm{L}}-\mathrm{T}_{0}\right)}{5 \mathrm{RM}_{0} \mathrm{~L}}}$
(C) $\frac{2 L}{\left(\sqrt{T_{L}}-\sqrt{T_{0}}\right)} \sqrt{\frac{3 M}{5 R}}$
(D) $L \sqrt{\frac{M_{0}}{2 R\left(T_{L}-T_{0}\right)}}$
25. In the given circuit the reading of ideal voltmeter is $\mathrm{E} / 2$. The internal resistance of the battery is
(A) $1 \Omega$
(B) $\frac{2}{3} \Omega$
(C) $\frac{2}{5} \Omega$
(D) $\frac{5}{2} \Omega$

26. The specific heat capacity of a monoatomic gas for the process $\mathrm{TV}^{2}=$ constant is (where R is gas constant)
(A) $\frac{R}{2}$
(B) $2 R$
(C) $\frac{R}{3}$
(D) $\frac{R}{4}$
27. 12 identical rods made of same material are arranged in the form of a cube. The temperature of ' $P$ ' and ' $R$ ' are maintained at $90^{\circ} \mathrm{C}$ and $30^{\circ} \mathrm{C}$ respectively. Then the temperature of point ' $V$ ', when steady state is reached,
(A) $65^{\circ} \mathrm{C}$
(B) $60^{\circ} \mathrm{C}$
(C) $20^{\circ} \mathrm{C}$
(D) $50^{\circ} \mathrm{C}$

28. A block is placed at the the bottom of an inclined plane and slided upwards with some initial speed. It slides up the incline, stops after time $t_{1}$, and slides back in a further time $t_{2}$. The angle of inclination of the plane with the horizontal is $\theta$ and the coefficient of friction is $\mu$. Then

(A) $t_{1}>t_{2}$
(B) $\mathrm{t}_{1}<\mathrm{t}_{2}$
(C) $\mathrm{t}_{1}=\mathrm{t}_{2}$
(D) $\mathrm{t}_{1}=\mathrm{t}_{2}$ only if $\theta=\frac{\pi}{4}$
29. A projectile is thrown from the base of an inclined plane at an angle $45^{\circ}$ with the plane. At what angle may it hit the plane again?
(A) $30^{\circ}$
(B) $45^{\circ}$
(C) $60^{\circ}$
(D) all of the above.

30. In an artificial satellite which of the following process of heat transfer will not take place?
(A) Conduction
(B) Convection
(C) Radiation
(D) All of the above

## Chemistry

## PART - II

## SECTION - A <br> Straight Objective Type

This section contains 30 multiple choice questions numbered 1 to 30 . Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

1. Which one of the following is correct?
(A) If one component of a binary liquid-liquid solution deviate negatively from Raoult's law, then the solution will form minimum boiling point azeotrope.
(B) for a solution of two liquids ( $A$ and $B$ ), $V_{\text {sol }}>V_{A}+V_{B}$; then $A$ and $B$ can be separated by fractional distillation.
(C) Deep sea divers use a mixture of He and $\mathrm{O}_{2}$ instead of air due to less solubility of He in blood in comparison to $\mathrm{N}_{2}$, which cause the bends, i.e. pain due to bubbling of untreated gas dissolved in blood.
(D) All of the above.
2. A compound is composed of two elements $A$ and $B$, element $A$ constitute f.c.c. lattice, while $B$ occupy all the tetrahedral voids, in this way another simple cubic is constituted by element $B$, inside the fcc unit cell of element A. If all the points/particles along any one edge of inner cube of every unit cell, are missing then what is the new empirical formula of the compound
(A) $\mathrm{A}_{4} \mathrm{~B}_{6}$
(B) $\mathrm{A}_{2} \mathrm{~B}_{8}$
(C) $\mathrm{A}_{2} \mathrm{~B}_{3}$
(D) $A B_{4}$
3. Which of the following gives yellow precipitate with aq. $\mathrm{AgNO}_{3}$ ?
(A) $\mathrm{KIO}_{3}$
(B) KI
(C) $\mathrm{CHI}_{3}$
(D) $\mathrm{CH}_{2} \mathrm{I}_{2}$
4. How many planes of symmetry and axis of symmetry are there in one molecule of $\mathrm{PCl}_{3} \mathrm{~F}_{2}$, respectively?
(A) 4,7
(B) 7,13
(C) 7,10
(D) 4,10
5. Two cylinders $A$ and $B$ of equal volumes are filled and equal masses of $N_{2}$ and $O_{2}$, respectively. Cylinder A is kept at 300 K while B is kept at 600 K , then (assume ideal behaviour of both gases)
(A) Average kinetic energy of $\mathrm{N}_{2}$ (per mole) in $A$ is equal to that of $\mathrm{O}_{2}$ in $B$.
(B) Molecules in cylinder B move twice as fast as those in A.
(C) Pressure in flask $A$ is less than that of $B$.
(D) Average velocity of $\mathrm{N}_{2}$ is equal to rms velocity of $\mathrm{O}_{2}$, in the given conditions.

## Space for Rough work

6. The ratio of time periods taken by electron in $1^{\text {st }}$ and $3^{\text {rd }}$ orbits of $\mathrm{He}^{+}$ion, for each revolution is.....
(A) $1: 9$
(B) $1: 27$
(C) $8: 27$
(D) $8: 9$
7. The equilibrium constant of mutarotation of $\alpha-D$-glucose to $\beta-D-$ glucose is 1.8 . If specific rotation of $\alpha$-D-glucose is $112^{\circ}$ and that of mixture is $59^{\circ}$; what is specific rotation of $\beta$-D-glucose?
(A) $-112^{\circ}$
(B) $-59^{\circ}$
(C) $+29.6^{\circ}$
(D) $-29.6^{\circ}$
8. Which of the following is correct?
(A) cis-cycloalkane is always more stable than trans cycloalkane.
(B)

(C)

(D)

9. Which of the following will undergo $\mathrm{E} 1_{\mathrm{CB}}$ ?
(A)

(B)

(C)

(D)

10. From which of the following halogen can be replaced by $\mathrm{H}_{2} \mathrm{O}$ by $\mathrm{S}_{1}$ 1?
(A) $\mathrm{CH}_{3}-\mathrm{Br}$
(C)

(B)
(D)

11. 



Product X is :
(A)

(B)

(C)

(D)

12. Which of the following can not be used for protection of carbonyl group?
(A) $\mathrm{CH}_{2}(\mathrm{OH}) \mathrm{CH}_{2} \mathrm{CHO}$
(B) $\mathrm{CH}_{2}(\mathrm{OH})-\mathrm{CH}_{2} \mathrm{CH}_{2}(\mathrm{OH}) / \mathrm{H}^{+}$
(C) $\mathrm{CH}_{2}(\mathrm{OH}) \mathrm{CH}_{2}(\mathrm{OH}) / \mathrm{H}^{+}$
(D) $\mathrm{HS}\left(\mathrm{CH}_{2}\right)_{3} \mathrm{SH}$
13. The configuration of the C-2 epimer of D-glucose is
(A) 2R, 3S, 4R, 5S
(B) $2 R, 3 \mathrm{~S}, 4 \mathrm{R}, 5 \mathrm{R}$
(C) $2 \mathrm{~S}, 3 \mathrm{~S}, 4 \mathrm{R}, 5 \mathrm{R}$
(D) 2S, 3R, 4R, 5S
14.


The product is
(A)

(B)

(C)

(D)

15. Acetophenone $\xrightarrow{\mathrm{HCO}_{3} \mathrm{H}} \underset{\text { (Major) }}{\mathrm{A}} \xrightarrow{\mathrm{H}_{3} \mathrm{O}^{+}} \mathrm{B}+\mathrm{C} \xrightarrow{\text { Phthalic Anhydride/ / }{ }^{+}}$Indicator(D) is
(A) PhOH
(B)

(C)


(D)

16. The complex $\left[\mathrm{Fe}\left(\mathrm{H}_{2} \mathrm{O}\right)_{5}(\mathrm{NO})\right]^{2+}$ is formed in the ring test for nitrate when freshly prepared $\mathrm{FeSO}_{4}$ solution is added to aqueous solution of $\mathrm{NO}_{3}^{-}$followed by addition of conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$. The complex is formed by charge transfer in which:
(A) $\mathrm{Fe}^{2+}$ charges to $\mathrm{Fe}^{3+}$, NO to $\mathrm{NO}^{+}$
(B) $\mathrm{Fe}^{3+}$ charge to $\mathrm{Fe}^{2+}$ and NO changes to $\mathrm{NO}^{+}$
(C) $\mathrm{Fe}^{2+}$ changes to $\mathrm{Fe}^{+}$and NO changes to $\mathrm{NO}^{+}$
(D) $\mathrm{Fe}^{2+}$ changes to $\mathrm{Fe}^{3+}$ and $\mathrm{NO}^{+}$changes to NO
17. Which of the following match is correct?

## Ore/ Minerals

## Element

(A) Bauxite B
(B) Magnetite Mn
(C) Bornite B
(D) Cerussite Pb
18. Which of the following is correct about Boron nitride $\left\{(\mathrm{BN})_{\mathrm{x}}\right\}$ ?
(A) All the atoms of the ring are uncharged
(B) It is insulator, unlike graphite
(C) Inter-layer distance in its graphite like structures, is less than that of graphite
(D) Resonance energy of each hexagon of $(\mathrm{BN})_{\mathrm{x}}$ is greater than that of Benzene.
19. Which of the following is true?
(A) Oxides of $\mathrm{Ge}, \mathrm{Sn}$ and Pb are amphoteric in nature
(B) Producer gas is produdced in incomplete combustion of C by air $\left(\mathrm{O}_{2}+\mathrm{N}_{2}\right)$
(C) SnO is much better reducing agent than PbO .
(D) All are correct
20. Which of the following can not be used as nitrating agent in Electrophilic Aromatic Substitution of benzene?
(A) $\mathrm{N}_{2} \mathrm{O}_{5} / \mathrm{MeCN}$
(B) $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{ONO}_{2}$
(C) $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{NO}_{2}$
(D) $\mathrm{NO}_{2} \cdot \mathrm{SO}_{3}$
21.


What is the product ' $P$ ' of the above reaction
(A)

(B)

(C)

(D)

22. Which of the following is tetrahedral and paramagnetic complex?
(A) $\left[\mathrm{NiCl}_{4}\right]^{2-}$
(B) $\left[\mathrm{Ni}(\mathrm{CN})_{4}\right]^{2-}$
(C) $\left[\mathrm{Cu}\left(\mathrm{NH}_{3}\right)_{4}\right]^{2+}$
(D) $\left[\mathrm{Ni}(\mathrm{CO})_{4}\right]$
23. Which of the following statement is correct?
(A) Free energy ( $G$ ) is a path function
(B) In the expression $\mathrm{G}=\mathrm{H}-\mathrm{TS}$, TS is the part of the system's energy that is disordered, already.
(C) $\Delta \mathrm{G}$ for a system is equal to $\mathrm{T} \Delta \mathrm{S}_{\text {Totaa }}$ in magnitude
(D) Statement A is incorrect but statement B and C are correct

Space for Rough work
24. All the elements $A, B, C$ and $D$ form homonuclear diatomic molecules and conform anion of type $\mathrm{X}^{-}$also. In a series of experiments the following observations are made:
(i) $2 \mathrm{~B}^{-}+\mathrm{C}_{2} \longrightarrow 2 \mathrm{C}^{-}+\mathrm{B}_{2}$
(ii) $2 \mathrm{~A}^{-}+\mathrm{C}_{2} \longrightarrow$ no reaction
(iii) $2 \mathrm{D}^{-}+\mathrm{C}_{2} \longrightarrow 2 \mathrm{C}^{-}+\mathrm{D}_{2}$
(iv) $2 \mathrm{~B}^{-}+\mathrm{D}_{2} \longrightarrow \mathrm{~B}_{2}+2 \mathrm{D}^{-}$

Which of the following statement is correct?
(A) $A_{2}$ is the strongest oxidizing agent among all $A_{2}, B_{2}, C_{2}$ and $D_{2}$.
(B) Correct order of reduction potentials is $A_{2}>C_{2}>D_{2}>B_{2}$.
(C) (A) and (B) are correct statements as well as it can also be concluded that $B$ corrode at fastest rate among all the four elements.
(D) All the above statements are correct.
25. What is the solubility product of $\mathrm{CaF}_{2}$ at room temperature, if ${ }^{\circ} \mathrm{m}\left(\mathrm{Ca}^{2+}\right)=1.04 \times 10^{-2} \mathrm{Sm}^{2} / \mathrm{mol}$
$\stackrel{\circ}{\mathrm{m}}_{\mathrm{m}}\left(\mathrm{F}^{-}\right)=4.8 \times 10^{-3} \mathrm{Sm}^{2} / \mathrm{mol}$
$\kappa_{\left.\mathrm{CaF}_{2}(\text { Saturated Solution })\right)}=4.25 \times 10^{-3} \mathrm{~S} / \mathrm{m}$ at room temperature .
$\kappa_{\mathrm{H}_{2} \mathrm{O}}=2 \times 10^{-4} \mathrm{~S} / \mathrm{m}$
(A) $4.05 \times 10^{-4} \mathrm{M}^{3}$
(B) $2.025 \times 10^{-4} \mathrm{M}^{3}$
(C) $3.32 \times 10^{-11} \mathrm{M}^{3}$
(D) $4.05 \times 10^{10} \mathrm{M}^{3}$
26. Some times yellow turbidity appears while passing $\mathrm{H}_{2} \mathrm{~S}$ gas even in the absence of $\mathrm{II}_{\text {nd }}$ group of basic radicals. This is because:
(A) Sulphur is present in the mixture as impurity.
(B) IV group radicals are precipitated as sulphides.
(C) The oxidation of $\mathrm{H}_{2} \mathrm{~S}$ gas by some acid radicals.
(D) $\mathrm{III}^{\text {rd }}$ group radicals are precipitated as hydroxides.
27. In low temperature range fraction of heat supplied to an ideal diatomic gas system, at constant pressure, which bring change in its internal energy is approx. ...?
(A) 0.71
(B) 0.75
(C) 0.60
(D) 0.50

Space for Rough work
28. By which of the following methods, $\mathrm{Cl}_{2}$ can be produced?
(A) By treating $\mathrm{KClO}_{3}$ with iodine ( $\mathrm{I}_{2}$ )
(B) By heating a mixture of NaBr and $\mathrm{NaBrO}_{3}$ with HCl
(C) By treating $\mathrm{CaOCl}_{2}$ and NaI mixture with HCl solution.
(D) By treating NaCl with $\mathrm{H}_{2} \mathrm{SO}_{4}$
29. Which of the following is optically active?
(A)

(B)

(C)

(D)

30. Which of the following will cause symmetric cleavage of $\mathrm{B}_{2} \mathrm{H}_{6}$ ?
(A) $\mathrm{NH}_{3}$
(B) $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{3}$
(C) $\mathrm{NH}_{2}-\mathrm{C}_{2} \mathrm{H}_{5}$
(D) $\mathrm{HN}\left(\mathrm{CH}_{3}\right)_{2}$

## Mathematics

## PART - III

## SECTION - A

## Straight Objective Type

This section contains 30 multiple choice questions numbered 1 to 30 . Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

1. For $\mathrm{n} \geq 3$ circles, the value of n for which the number of radical axis is equal to number of radical centres is
(A) 3
(B) 4
(C) 5
(D) none of these
2. Consider a circle, $x^{2}+y^{2}=1$ and point $P(1, \sqrt{3}) \cdot P A B$ is a secant drawn from $P$ intersecting circle in $A$ and $B$ (distinct) then range of $|P A|+|P B|$ is
(A) $[2,2 \sqrt{3}]$
(B) $(2 \sqrt{3}, 4]$
(C) $(0,4]$
(D) none of these
3. Consider a curve $|z-i|=2$ and a point $z_{1}=3-i$, then the length of tangent made from the point $\left(z_{1}\right)$ to the curve is
(A) 2
(B) 3
(C) 5
(D) none of these
4. The total number of 1 word, 2 word, 3 word sentences that can be formed using the letters of the word SAMSUNG is
(A) 8 !
(B) $18 \times 7$ !
(C) $11 \times 7$ !
(D) none of these
5. Consider a line $z(i-1)+\bar{z}(i+1)=0$ in the argand plane and a point $z_{1}=2+3 i$ then the reflection of $z_{1}$ in the given line is
(A) $2-3 \mathrm{i}$
(B) $-2+3 i$
(C) $3+2 \mathrm{i}$
(D) none of these

Space for Rough work
6. The sequence $a_{n}$ is defined by $a_{1}=\frac{1}{2}, a_{n+1}=a_{n}^{2}+a_{n}$. Also, $S=\frac{1}{a_{1}+1}+\frac{1}{a_{2}+1}+\ldots . .+\frac{1}{a_{100}+1}$ then [ S ] (where [.] denotes the greatest integer function) is
(A) 1
(B) 2
(C) 3
(D) none of these
7. Consider a plane $\bar{r} \cdot(2 \hat{i}+2 \hat{j}-\hat{k})=5$ which cuts a circular section from the sphere $|\bar{r}|=9$ then the volume of the cone formed by taking circular cross section as base and vertex as centre of sphere is
(A) $\frac{3520 \pi}{27}$
(B) $\frac{40}{9} \sqrt{11} \pi$
(C) $\frac{64}{9} \pi$
(D) none of these
8. Let, $\mathrm{t}_{\mathrm{r}}=\mathrm{r}!$ and $\mathrm{S}_{\mathrm{n}}=\sum_{\mathrm{r}=1}^{\mathrm{n}} \mathrm{r}$ ! then $\frac{\mathrm{S}_{\mathrm{n}}}{24}=\mathrm{a}+\frac{\lambda}{24} ; \mathrm{a}, \lambda \in \mathrm{N}$ where $\lambda$ is
(A) 7
(B) 23
(C) 9
(D) none of these
9. If $z=\left[\begin{array}{ll}O & 1 \\ 1 & O\end{array}\right]$ where $O, I$ are $2 \times 2$ null and identity matrix then $\operatorname{det}(|z|)$ is
(A) 1
(B) -1
(C) 0
(D) none of these
10. Given that $f(x)+f\left(\frac{1}{1-x}\right)=\frac{2(1-2 x)}{x(1-x)}, x \in R, x \neq 0,1$ then the given $f(x)$ is
(A) one-one
(B) many one
(C) $f(0)=1$
(D) none of these
11. The area between the curve $y^{2}(a+x)=(a-x)^{3}$ and its vertical asymptote is
(A) $\frac{\pi}{2} a^{2}$
(B) $2 \pi a^{2}$
(C) $3 \pi a^{2}$
(D) none of these
12. If the given curve satisfies the differential equation $e^{y} d x+\left(x e^{y}+2 y\right) d y=0$ and also passes through $(0,0)$ then the possible equation of curve can be
(A) $x e^{y}+y=0$
(B) $x+y^{2} e^{y}=0$
(C) $x^{2} e^{x}+y e^{y}=1$
(D) none of these
13. Let $\bar{v}$ be a unit vector which follows the equation, $\overline{\mathrm{v}} \times \overline{\mathrm{b}}=\overline{\mathrm{c}}$. Also, $|\overline{\mathrm{b}}|=2$ and $|\overline{\mathrm{c}}|=\sqrt{3}$ then
(A) $\overline{\mathrm{v}}=-\overline{\mathrm{b}}+\overline{\mathrm{b}} \times \overline{\mathrm{c}}$
(B) $\overline{\mathrm{v}}=\frac{3}{4}(\overline{\mathrm{~b}}+2 \overline{\mathrm{~b}} \times \overline{\mathrm{c}})$
(C) $\overline{\mathrm{v}}=\frac{1}{4}(\overline{\mathrm{~b}}+\overline{\mathrm{b}} \times \overline{\mathrm{c}})$
(D) none of these
14. If the tangent to the curve $y=1-x^{2}$ at $x=\alpha(0<\alpha<1)$ meets the axes at $P$ and $Q$. Also $\alpha$ varies, the minimum value of the area of the triangle OPQ is $k$ times the area bounded by the axes and the part of the curve for which $0<x<1$, then $k$ is
(A) $\frac{\sqrt{3}}{2}$
(B) $\frac{2}{\sqrt{3}}$
(C) $\frac{1}{2}$
(D) $\frac{3}{2}$
15. Consider a parabola $y^{2}=\alpha x$ and a point $\left(-\frac{\alpha}{4}, 0\right)$ then midpoint of centres of the circles touching the tangents from given point and its chord of contact is
(A) $\frac{\alpha}{2}$
(B) $\frac{\alpha}{4}$
(C) $\frac{3 \alpha}{2}$
(D) none of these
16. Consider a curve $(2 x+y-1)^{2}=5(x-2 y-3)$ then the possible coordinates of foci are
(A) $\left(\frac{5}{4}, \frac{3}{4}\right)$
(B) $\left(\frac{3}{4},-\frac{1}{2}\right)$
(C) $\left(\frac{5}{4}, \frac{3}{2}\right)$
(D) none of these
17. Consider the curves $C_{1}: x^{2}+y^{2}=1$ and $C_{2}: \frac{x^{2}}{\sin ^{2} \theta}+\frac{y^{2}}{\cos ^{2} \theta}=1$. If a common tangent $y=m x+c$ is drawn to $\mathrm{C}_{1}, \mathrm{C}_{2}$ then $\left(\frac{\pi}{4}<\theta<\frac{\pi}{2}\right)$
(A) $m \in \phi$
(B) $\mathrm{c}=1$
(C) $m=\frac{1}{\sqrt{2}}$
(D) none of these
18. Consider a parabola $x^{2}=4 y$ and a hyperbola $x y=1$. A tangent is drawn to parabola meets the hyperbola in $A$ and $B$ then locus of midpoint of $A B$ is
(A) straight line
(B) parabola
(C) ellipse
(D) none of these
19. If the following represent the two lines with real point of intersection then $\alpha_{1} x+\alpha_{2} y=k_{1}$, $\beta_{1} x+\beta_{2} y=k_{2}$. Also, $\left(\alpha_{1}, \alpha_{2}\right),\left(\beta_{1}, \beta_{2}\right)$ lie on $x^{2}+y^{2}=1$ then
(A) $\alpha_{1} \beta_{1}+\alpha_{2} \beta_{2} \in[0,2]$
(B) lines can represent asymptotes of rectangular hyperbola
(C) $\alpha_{1} \beta_{1}+\alpha_{2} \beta_{2} \in[-2,0]$
(D) none of these
20. $g(n)=\int_{0}^{n^{2}+n+1} e^{x / 2-[x / 2]}\left(\frac{x}{2}-\left[\frac{x}{2}\right]\right) d(x-[x]) ; n \in N$ then $g(n)$
(A) has minimum value as $\frac{1}{4}+\sqrt{e}$
(B) has maximum value as $3-\sqrt{e}$
(C) has minimum value as $\frac{3}{4}-\sqrt{\frac{\mathrm{e}}{4}}$
(D) none of these
21. Consider a sequence $f(x), n \in I^{+} \cup\{0\}$ defined by $f(0)=0, f(n+1)=\sqrt{6+f(n)}$ then $\lim _{n \rightarrow \infty} f(n)$ is
(A) 1
(B) 2
(C) 3
(D) none of these
22. If $\left(a-a^{\prime}\right)^{2}+\left(b-b^{\prime}\right)^{2}+\left(c-c^{\prime}\right)^{2}=p$ and $\left(a b^{\prime}-a^{\prime} b\right)^{2}+\left(b c^{\prime}-b^{\prime} c\right)^{2}+\left(c a^{\prime}-c^{\prime} a\right)^{2}=q$, then the perpendicular distance of the line $a x+b y+c z=1, a^{\prime} x+b^{\prime} y+c^{\prime} z=1$ from origin, is
(A) $\sqrt{\frac{p}{q}}$
(B) $\sqrt{\frac{q}{p}}$
(C) $\frac{p}{\sqrt{q}}$
(D) $\frac{q}{\sqrt{p}}$
23. Consider a function, $f: R \rightarrow R$ such that $f(x+a)=\frac{1}{2}+\sqrt{f(x)-f^{2}(x)}$, $a$ is a real constant. If $f(x)$ is periodic then its period can be
(A) $\frac{a}{2}$
(B) $a$
(C) 2 a
(D) none of these
24. Consider an identity matrix $(n \times n) I_{n} ; \lambda \in R^{+}$then $\left|\operatorname{Adj}\left(\lambda I_{n}\right)\right|$ is
(A) $\lambda^{n-1}$
(B) $\lambda^{n(n-1)}$
(C) $\lambda^{n}$
(D) $\lambda^{2(n-1)}$
25. Let $A_{1}, A_{2}, A_{3}, \ldots ., A_{7}$ be skew symmetric matrices of same order then $1 \cdot\left(A_{1}\right)^{1}+3 \cdot\left(A_{2}\right)^{3}+5\left(A_{3}\right)^{7}+\ldots .+13\left(A_{7}\right)^{13}$ is
(A) symmetric
(B) skew symmetric
(C) neither symmetric nor skew symmetric
(D) none of these
26. Given the expansion $(a+p)^{m-1}+(a+p)^{m-2}(a+q)+(a+p)^{m-3}(a+q)^{2}+\ldots . .+(a+q)^{m-1}$ [ $a \neq-q$ and $p \neq q$ ] then coefficient of $a^{t}$ is
(A) $\frac{{ }^{m} C_{t}\left[p^{t}-q^{t}\right]}{p-q}$
(B) ${ }^{m} C_{t} \frac{p^{m-t}-q^{m-t}}{p-q}$
(C) $\frac{{ }^{m} C_{t}\left[p^{t}+q^{t}\right]}{p-q}$
(D) ${ }^{m} C_{t} \frac{p^{m-t}+q^{m-t}}{p-q}$
27. Consider a curve $C: y^{2}=8 x$ and line $L: 2 x+y=3$. If three distinct points $A, B, C$ are selected at random on $L$, then the maximum probability that the tangents drawn from $A, B, C$ to $y^{2}=8 x$ are mutually perpendicular is [For points $A, B, C, S_{1}>0$ ]
(A) 1
(B) $\frac{1}{3}$
(C) $\frac{2}{3}$
(D) none of these
28. Let $F, F^{\prime}, F^{\prime \prime}$ be continuous in $[0, \ln 2]$ and $f(0)=0, f^{\prime}(0)=3, f(\ln 2)=6, f^{\prime}(\ln 2)=4$ and $\int_{0}^{\ln 2} e^{-2 x} f(x) d x=3$ then $\int_{0}^{\ln 2} e^{-2 x^{\prime \prime \prime}}(x) d x$ is
(A) 10
(B) 13
(C) 12
(D) none of these
29. Consider circles $S: x^{2}+y^{2}=1$ and $S^{\prime}:(x-4)^{2}+(y-0)^{2}=9$. Now a circle is drawn touching $S, S^{\prime}$ externally and also their direct common tangent then its radius is
(A) $3(4+2 \sqrt{3})$
(B) $\frac{3}{4}(4-2 \sqrt{3})$
(C) $4-2 \sqrt{3}$
(D) none of these
30. Given that $a, b, c$ are sides of the triangle $\triangle A B C$ such that $z=\log _{2^{a}+2^{-a}}\left[4(a b+b c+c a)-(a+b+c)^{2}\right]$ then $z$ has a real value if and only if
(A) $a=b=2 c$
(B) $3 \mathrm{a}=2 \mathrm{~b}=\mathrm{c}$
(C) $a-b=3 c$
(D) none of these

