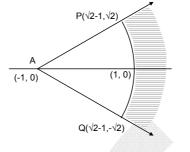
Note: **FIITJ€€** solutions to IIT–JEE, 2005 Screening Test is based on Screening Test paper created using memory retention of select **FIITJ€€** students appeared in this test and hence may not exactly be the same as the original paper. However, every effort has been made to reproduce the original paper in the interest of the aspiring students.

FIITJ€€ solutions to IIT–JEE, 2005 Screening

1. The locus of z which lies in shaded region is best represented by (A) z : |z + 1| > 2, $|\arg(z + 1)| < \pi/4$ (B) z : |z - 1| > 2, $|\arg(z - 1)| < \pi/4$ (C) z : |z + 1| < 2, $|\arg(z + 1)| < \pi/2$

(D) z : |z - 1| < 2, $|\arg(z - 1)| < \pi/2$



Ans. A

2.

Sol. The points (1, 0), $(\sqrt{2} - 1, -\sqrt{2})$ and $(\sqrt{2} - 1, \sqrt{2})$ are equidistant from the point (-1, 0).

The shaded area belongs to the region outside the sector of circle |z + 1| = 2, lying between the line rays

 $\arg(z+1) = \frac{\pi}{4}$ and $\arg(z+1) = \frac{-\pi}{4}$.

In an equilateral triangle, 3 coins of radii 1 unit each are kept so that they touch each other and also the sides of the triangle. Area of the triangle is (A) $4 + 2\sqrt{3}$ (B) $6 + 4\sqrt{3}$ (C) $12 + \frac{7\sqrt{3}}{4}$ (D) $3 + \frac{7\sqrt{3}}{4}$

Ans. B

Sol. The line joining the vertex of the triangle and the centre of the coin makes angle $\frac{\pi}{6}$ with the sides of the triangle. The length of each of the sides of the equilateral triangle is $2 + 2 \cot \frac{\pi}{6} = 2 (1 + \sqrt{3})$.

(B) 1

(D) $\frac{1}{2}$

Hence its area is
$$\frac{\sqrt{3}}{4}4(1+\sqrt{3})^2 = 6+4\sqrt{3}$$
.

3. If a, b, c are integers not all equal and w is a cube root of unity $(w \neq 1)$, then the minimum value of $|a + bw + cw^2|$ is

(A) 0 (C) $\frac{\sqrt{3}}{2}$

Ans.

B

Sol.
$$|a + bw + cw^2| = \sqrt{\left(a - \frac{b}{2} - \frac{c}{2}\right)^2 + \frac{3}{4}(c - b)^2} = \sqrt{\frac{1}{2}\left((a - b)^2 + (b - c)^2 + (c - a)^2\right)}$$
.

This is minimum when a = b and $(b - c)^2 = (c - a)^2 = 1 \Rightarrow$ The minimum value is 1.

4. A rectangle with sides 2m - 1 and 2n - 1 is divided into squares of unit length by drawing parallel lines as shown in the diagram, then the number of rectangles possible with odd side lengths is (A) $(m + n + 1)^2$ (B) 4^{m+n-1} (C) $m^2 n^2$ (D) mn(m + 1)(n + 1)

Ans. C

Sol. There are 2m vertical (numbered 1, 2, ..., 2m) and 2n horizontal lines (numbered 1, 2, ...,2n). To form the required rectangle we must select two horizontal lines, one even numbered and one odd numbered and similarly two vertical lines. The number of rectangles is then ${}^{m}C_{1}$. ${}^{m}C_{1}$. ${}^{n}C_{1} = m^{2}n^{2}$. **Alternate solution:** Number of rectangles possible is $(1 + 3 + 5 + ... + (2m - 1))(1 + 3 + 5 + ... + (2n - 1)) = m^{2}n^{2}$.

5. A circle is given by $x^2 + (y-1)^2 = 1$, another circle C touches it externally and also the x-axis, then the locus of its centre is

(A) $\{(x, y) : x^2 = 4y\} \cup \{(x, y) : y \le 0\}$ (B) $\{(x, y) : x^2 + (y - 1)^2 = 4\} \cup \{x, y) : y \le 0\}$ (C) $\{(x, y) : x^2 = y\} \cup \{(0, y) : y \le 0\}$ (D) $\{(x, y) : x^2 = 4y\} \cup \{(0, y) : y \le 0\}$

Ans.

D

Sol. Let the circle touching the x-axis be $x^2 + y^2 - 2ax - 2by + a^2 = 0$ with centre at (a, b) and radius b. Since it touches the circle $x^2 + (y-1)^2 = 1$, $|b+1| = \sqrt{a^2 + (b-1)^2}$. $\Rightarrow b^2 + 2b + 1 = a^2 + b^2 - 2b + 1$ $\Rightarrow 4b = a^2$ so that locus of (a, b) is $x^2 = 4y$. If the centre of the circle lies on the y-axis, then $y \le 0$.

6. $\cos (\alpha - \beta) = 1$ and $\cos (\alpha + \beta) = 1/e$, where $\alpha, \beta \in [-\pi, \pi]$. Pairs of α, β which satisfy both the equations is/ are (A) 0 (B) 1 (C) 2 (D) 4

Ans. D

- Sol. For $\cos (\alpha \beta) = 1$, $\alpha = \beta$ so that $\cos (\alpha + \beta) = 1/e \Rightarrow \alpha + \beta = \pm \cos^{-1} 1/e$ $\Rightarrow 2\alpha = \pm \cos^{-1} \left(\frac{1}{e}\right) \in [-2\pi, 2\pi]$. $\Rightarrow \alpha, \beta$ can be satisfied by 4 sets of values.
- 7. In \triangle ABC, a, b, c are the lengths of its sides and A, B, C are the angles of triangle ABC. The correct relation is given by

(A)
$$(b-c) \sin\left(\frac{B-C}{2}\right) = a \cos\frac{A}{2}$$

(B) $(b-c) \cos\frac{A}{2} = a \sin\left(\frac{B-C}{2}\right)$
(C) $(b+c) \sin\left(\frac{B+C}{2}\right) = a \cos\frac{A}{2}$
(D) $(b-c) \cos\left(\frac{A}{2}\right) = 2a \sin\left(\frac{B+C}{2}\right)$
B

Ans.

Sol. Here
$$\frac{b-c}{a} = \frac{\sin B - \sin C}{\sin A} = \frac{2 \sin \frac{B-C}{2} \cos \frac{B+C}{2}}{2 \sin \frac{A}{2} \cos \frac{A}{2}} = \frac{\sin \left(\frac{B-C}{2}\right)}{\cos \frac{A}{2}}$$
.
8. The value of $\binom{30}{0}\binom{30}{10} - \binom{30}{1}\binom{30}{11} + \binom{30}{2}\binom{30}{12} \dots + \binom{30}{20}\binom{30}{30}$ is, where $\binom{n}{r} = {}^{n}C_{r}$.
(A) $\binom{30}{10}$ (B) $\binom{30}{15}$
(C) $\binom{60}{30}$ (D) $\binom{31}{10}$

Ans.

Sol. The given expression is the coefficient of x^{20} in the product $(1 + x)^{30} (1 - x)^{30} = (1 - x^2)^{30}$ \Rightarrow the given expression $= {}^{30}C_{10}$.

9. A variable plane at a distance of 1 unit from the origin cuts the co-ordinate axes at A, B and C. If the centroid D (x, y, z) of triangle ABC satisfies the relation $\frac{1}{x^2} + \frac{1}{y^2} + \frac{1}{z^2} = k$, then the value of k is (A) 3 (C) 1/3 (D) 9

IIT-JEE-2005-S-3

Ans. D

Sol. Let
$$\frac{x}{a} + \frac{y}{b} + \frac{z}{c} = 1$$
 be the variable plane so that $\left| \frac{1}{\sqrt{\frac{1}{a^2} + \frac{1}{b^2} + \frac{1}{c^2}}} \right| = 1$.

The plane meets the coordinate axes at A (a, 0, 0), B (0, b, 0), C (0, 0, c). The centroid D of the triangle ABC is $\left(\frac{a}{3}, \frac{b}{3}, \frac{c}{3}\right)$

$$\Rightarrow x = \frac{a}{3}, y = \frac{b}{3}, z = \frac{c}{3} \text{ and } \frac{1}{a^2} + \frac{1}{b^2} + \frac{1}{c^2} = 1 \Rightarrow \frac{1}{x^2} + \frac{1}{y^2} + \frac{1}{z^2} = 9$$

10. If $\int_{\sin x}^{1} t^2 (f(t)) dt = (1 - \sin x)$, then $f\left(\frac{1}{\sqrt{3}}\right)$ is
(A) 1/3 (B) $1/\sqrt{3}$
(C) 3 (D) $\sqrt{3}$

Ans.

С

Sol. Differentiating both sides with respect to x, we get

$$-\sin^{2} x f(\sin x) \cdot \cos x = -\cos x \implies f(\sin x) = \frac{1}{\sin^{2} x}$$
$$\implies f(x) = \frac{1}{x^{2}} \implies f\left(\frac{1}{\sqrt{3}}\right) = 3.$$

11. In the quadratic equation $ax^2 + bx + c = 0$, if $\Delta = b^2 - 4ac$ and $\alpha + \beta$, $\alpha^2 + \beta^2$, $\alpha^3 + \beta^3$ are in G.P. where α , β are the roots of $ax^2 + bx + c = 0$, then (A) $\Delta \neq 0$ (B) $b\Delta = 0$

(A)
$$\Delta \neq 0$$
(B) $b\Delta = 0$ (C) $c\Delta = 0$ (D) $\Delta = 0$ C

Ans.

Sol. We have
$$(\alpha^2 + \beta^2)^2 = (\alpha + \beta) (\alpha^3 + \beta^3) \Rightarrow \alpha\beta (\alpha - \beta)^2 = 0$$

 $\Rightarrow c\Delta = 0.$

12. A six faced fair dice is thrown until 1 comes, then the probability that 1 comes in even no. of trials is (A) 5/11 (B) 5/6 (C) 6/11 (D) 1/6

Ans.

Α

Sol. The required probability
$$= \frac{1}{6} \cdot \frac{5}{6} + \frac{1}{6} \cdot \left(\frac{5}{6}\right)^5 + \dots$$

 $= \frac{1}{6} \cdot \frac{5}{6} \left[1 + \left(\frac{5}{6}\right)^2 + \dots \right] = \frac{5}{11}.$

13. If f(x) is a twice differentiable function and given that f(1) = 1, f(2) = 4, f(3) = 9, then (A) f''(x) = 2, for $\forall x \in (1, 3)$ (B) f''(x) = f'(x) = 5 for some $x \in (2, 3)$ (C) f''(x) = 3, $\forall x \in (2, 3)$ (D) f''(x) = 2, for some $x \in (1, 3)$ Ans. D Set L at g(x) = f(x) x^2

Sol. Let $g(x) = f(x) - x^2$. We have g(1) = 0, g(2) = 0, g(3) = 0. Hence by Rolle's theorem g'(x) = 0 for some $c \in (1, 2)$ and g'(x) = 0 for some $d \in (2, 3)$. Again, by Rolle's theorem g''(x) = 0 at some $x \in (c, d)$ $\Rightarrow f''(x) = 2$ for some values $x \in (1, 3)$. 14. $\int_{-2}^{0} (x^3 + 3x^2 + 3x + 3 + (x + 1)\cos(x + 1)) dx$ is equal to (A) - 4 (B) 0 (C) 4 (D) 6

Ans. С Here I = $\int_{0}^{0} [x^{3} + 3x^{2} + 3x + 3 + (x+1)\cos(x+1)]dx$ Sol. Put x + 1 = t $= \int_{1}^{1} \left[\left(t^{3} + t \cos t \right) + 2 \right] dt$ $=\int 2dt = 4$. If P (x) is a polynomial of degree less than or equal to 2 and S is the set of all such polynomials so that P (1) = 1, 15. $P(0) = 0 \text{ and } P'(x) > 0 \quad \forall x \in [0, 1], \text{ then}$ (B) $S = \{(1-a) x^2 + ax \quad 0 < a < 2$ (D) $S = \{(1-a) x^2 + ax \quad 0 < a < 1$ (A) $S = \phi$ (C) (1- a) $x^2 + ax \ a \in (0, \infty)$ В Ans. Let the polynomial be $P(x) = ax^2 + bx + c$ Sol. $P(0) = 0 \implies c = 0$ and $P(1) = 1 \implies a + b = 1$ so that $P'(x) = 2(1-b)x + b > 0 \forall x$ \Rightarrow b \in (0, 2). \Rightarrow S = {(1-a)x² + ax, a \in (0, 2)} The minimum area of triangle formed by the tangent to the ellipse $\frac{x^2}{x^2} + \frac{y^2}{x^2} = 1$ and coordinate axes is 16. (B) $\frac{a^2 + b^2}{2}$ sq. units (A) ab sq. units (D) $\frac{a^2 + ab + b^2}{3}$ sq. units (C) $\frac{(a+b)^2}{2}$ sq. units Ans. Α A tangent of the given ellipse is $y = mx + \sqrt{a^2m^2 + b^2}$. Sol. It meets the axes at $\left(\frac{-\sqrt{a^2m^2+b^2}}{m}, 0\right)$ and $\left(0, \sqrt{a^2m^2+b^2}\right)$. Hence the area of the triangle is $\frac{1}{2} \left| \frac{a^2 m^2 + b^2}{m} \right| = \frac{1}{2} \left| a^2 m + \frac{b^2}{m} \right| \ge ab$. Alternate: The equation of tangent at $(a\cos\theta, b\sin\theta)$ is $\frac{x\cos\theta}{a} + \frac{y\sin\theta}{b} = 1$. It meets the coordinate axes at $A \equiv (0, b \operatorname{cosec} \theta), B \equiv (a \sec \theta, 0).$ Area of triangle = $\frac{ab}{2\sin\theta\cos\theta} = \frac{ab}{\sin 2\theta} \ge ab$. If the functions f(x) and g(x) are defined on $R \rightarrow R$ such that 17. $x \in \text{rational} \\ x \in \text{irrational}, g(x) = \begin{cases} 0, & x \in \text{irrational} \\ x, & x \in \text{rational} \end{cases}$ f(x) = , then (f-g)(x) is (A) one-one and onto (B) neither one-one nor onto (C) one-one but not onto (D) onto but not one-one Ans. A Let $h(x) = f(x) - g(x) = \begin{cases} x; & x \in \text{irrational} \\ -x; & x \in \text{rational} \end{cases}$ Sol. \Rightarrow the function h(x) is one-one and onto.

IIT-JEE-2005-S-5

18. The area bounded by the parabolas $y = (x + 1)^2$ and $y = (x - 1)^2$ and the line y = 1/4 is (A) 4 sq. units (C) 4/3 sq. units (D) 1/3 sq. units

Ans. D

Sol. The parabolas meet at (0, 1) and intersect the line
$$y = 1/4$$
 at $x = -\frac{3}{2}, -\frac{1}{2}, \frac{1}{2}$ and $\frac{3}{2}$

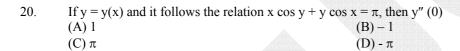
Hence the required area = $2\left[\int_{0}^{1/2} (x-1)^2 dx\right] - \frac{1}{4} = \frac{2}{3}(x-1)^3 \Big|_{0}^{1/2} - \frac{1}{4} = \frac{1}{3}$

19. The function given by y = ||x| - 1| is differentiable for all real numbers except the points (A) $\{0, 1, -1\}$ (B) ± 1 (C) 1 (D) -1

Ans.

A

Sol. From the graph, the function is not differentiable at x = -1, 0, 1.



Ans. C

Sol. $x \cos y + y \cos x = \pi$, $y(0) = \pi$. $\Rightarrow -x \sin y \frac{dy}{dx} + \cos y - y \sin x + \cos x \frac{dy}{dx} = 0 \Rightarrow y'(0) = 1$. Again differentiating and using y'(0) = 1 and $y(0) = \pi$, we get $y''(0) = \pi$.

21. The solution of primitive integral equation $(x^2 + y^2) dy = xy dx$, is y = y(x). If y(1) = 1 and $y(x_0) = e$, then x_0 is (A) $\sqrt{2(e^2 - 1)}$ (B) $\sqrt{2(e^2 + 1)}$

(0, 1

С

(1, 0)

(-1, 0)

(C)
$$\sqrt{3}e$$
 (D) $\sqrt{\frac{e^2 + 1}{2}}$

Ans.

Sol. We have $\frac{dy}{dx} = \frac{xy}{x^2 + y^2}$

С

Solving the homogenous differential equation by writing y = vx, we get

$$-\frac{x^{2}}{2y^{2}} + \ln y = -\frac{1}{2}.$$

For $y = e$, $\frac{-x_{0}^{2}}{2e^{2}} + \ln e = -\frac{1}{2} \Rightarrow x_{0}^{2} = 3e^{2} \Rightarrow x_{0} = \sqrt{3} e.$
22. $A = \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 1 \\ 0 & -2 & 4 \end{bmatrix}, I = \begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{bmatrix} \text{ and } A^{-1} = \begin{bmatrix} \frac{1}{6} (A^{2} + cA + dI) \end{bmatrix}, \text{ then the value of } c \text{ and } d \text{ are}$
(A) - 6, -11
(C) - 6, 11
(D) 6, -11

Ans. C

IIT-JEE-2005-S-6

Sol. We evaluate A^2 and A^3 and write the given equation as $AA^{-1} = I = \frac{1}{6} [A^3 + cA^2 + dA]$. Comparing the corresponding elements on both the sides we get c = -6, d = 11.

Alternatively, we may use Cayley Hamilton Theorem.

23. If
$$P = \begin{bmatrix} \frac{\sqrt{3}}{2} & \frac{1}{2} \\ -\frac{1}{2} & \frac{\sqrt{3}}{2} \end{bmatrix}$$
, $A = \begin{bmatrix} 1 & 1 \\ 0 & 1 \end{bmatrix}$ and $Q = PAP^{T}$ and $x = P^{T} Q^{2005} P$, then x is equal to
(A) $\begin{bmatrix} 1 & 2005 \\ 0 & 1 \end{bmatrix}$
(B) $\begin{bmatrix} 4 + 2005\sqrt{3} & 6015 \\ 2005 & 4 - 2005\sqrt{3} \end{bmatrix}$
(C) $\frac{1}{4} \begin{bmatrix} 2+\sqrt{3} & 1 \\ -1 & 2-\sqrt{3} \end{bmatrix}$
(D) $\frac{1}{4} \begin{bmatrix} 2005 & 2-\sqrt{3} \\ 2+\sqrt{3} & 2005 \end{bmatrix}$

Ans. Sol.

А

1

$$P^{T}P = I$$

$$Q = PAP^{T} \text{ so that}$$

$$x = P^{T}Q^{2005}P = P^{T}(PAP^{T})^{2005}P$$

$$= P^{T}PAP^{T}(PAP^{T})^{2004}P$$

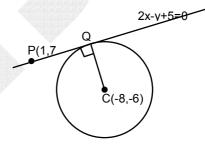
$$= A^{2005} = \begin{bmatrix} 1 & 2005\\ 0 & 1 \end{bmatrix}$$

24. Tangent to the curve $y = x^2 + 6$ at a point P (1, 7) touches the circle $x^2 + y^2 + 16x + 12y + c = 0$ at a point Q. Then the coordinates of Q are (A) (-6, -11) (C) (-10, -15) (D) (-6, -7)

Ans.

Sol. Equation of tangent to the parabola at (1, 7) is

 $x - \frac{(y+7)}{2} + 6 = 0 \implies 2x - y + 5 = 0.$ $\Rightarrow \text{Centre} \equiv (-8, -6)$ Equation of CQ = x + 2y + k = 0 $-8 - 12 + k = 0 \implies k = 20$ PQ = 4x - 2y + 10 = 0 CQ = x + 2y + 20 = 0 $= 5x + 30 = 0 \implies x = -6$ $\Rightarrow -6 + 2y + 20 = 0 \implies y = -7$ Hence the point of contact is (-6, -7).



25.

If f(x) is a continuous and differentiable function and $f\left(\frac{1}{n}\right) = 0 \quad \forall n \ge 1 \text{ and } n \in I$, then

(A) $f(x) = 0, x \in (0, 1]$ (C) $f'(0) = 0 = f''(0), x \in (0, 1]$ (B) f (0) = 0, f'(0) = 0 (D) f(0) = 0 and f'(0) need not to be zero

Ans.

B

Sol. Given $f\left(\frac{1}{n}\right) = 0 \quad \forall n \ge 1 \text{ and } n \in I.$

This indicates that f(x) has a wavy behaviour. Amplitude of the wave either (a) is constant (b) increases or (c) decreases. In case of (a) and (b), function will not be differentiable at 0. \Rightarrow Amplitude has to decrease such that f'(0) = 0.

IIT-JEE-2005-S-7

If \vec{a} , \vec{b} , \vec{c} are three non-zero, non-coplanar vectors and $\vec{b}_1 = \vec{b} - \frac{\vec{b} \cdot \vec{a}}{|\vec{a}|^2} \vec{a}$, $\vec{b}_2 = \vec{b} + \frac{\vec{b} \cdot \vec{a}}{|\vec{a}|^2} \vec{a}$, 26. $\vec{c}_{1} = \vec{c} - \frac{\vec{c} \cdot \vec{a}}{|\vec{a}|^{2}} \vec{a} + \frac{\vec{b} \cdot \vec{c}}{|\vec{c}|^{2}} \vec{b}_{1}, \quad \vec{c}_{2} = \vec{c} - \frac{\vec{c} \cdot \vec{a}}{|\vec{a}|^{2}} \vec{a} - \frac{\vec{b}_{1} \cdot \vec{c}}{|\vec{b}_{1}|^{2}} \vec{b}_{1}, \quad \vec{c}_{3} = \vec{c} - \frac{\vec{c} \cdot \vec{a}}{|\vec{c}|^{2}} \vec{a} + \frac{\vec{b} \cdot \vec{c}}{|\vec{c}|^{2}} \vec{b}_{1}, \quad \vec{c}_{4} = \vec{c} - \frac{\vec{c} \cdot \vec{a}}{|\vec{c}|^{2}} \vec{a} - \frac{\vec{b} \cdot \vec{c}}{|\vec{b}|^{2}} \vec{b}_{1}, \quad \text{then}$ the set of orthogonal vectors is (A) $(\vec{a}, \vec{b}_1, \vec{c}_3)$ (B) $(\vec{a}, \vec{b}_1, \vec{c}_2)$ (D) $(\vec{a}, \vec{b}_2, \vec{c}_2)$ (C) $(\vec{a}, \vec{b}_1, \vec{c}_1)$ Ans. В Obviously $\vec{a} \cdot \vec{b}_1 = \left(\vec{b} \cdot \vec{a} - \frac{\vec{b} \cdot \vec{a}}{|\vec{a}|^2} \vec{a} \cdot \vec{a} \right) = 0$ Sol. & $\vec{a} \cdot \vec{c}_2 = 0$ and $\vec{b}_1 \cdot \vec{c}_2 = 0$. \Rightarrow ($\vec{a}, \vec{b}_1, \vec{c}_2$) are orthogonal vectors. For the primitive integral equation $ydx + y^2 dy = x dy$; $x \in R$, y > 0, y = y(x), y(1) = 1, then y(-3) is 27. (B) 2 (A) 3 (C) 1 (D) 5 Ans. $y\frac{dx - xdy}{v^2} = -dy$ Sol. $\frac{x}{y} = -y + c$ $y(1) = 1 \implies c = 2$ $y^2 - 2y + x = 0$ y(-3): $y^2 - 2y - 3 = 0 \implies (y - 3)(y + 1) = 0$ y = 3, -1X and Y are two sets and $f: X \rightarrow Y$. If $\{f(c) = y; c \subset X, y \subset Y\}$ and 28. $\{f^{1}(d) = x; d \subset Y, x \subset X\}$, then the true statement is (B) $f^{1}(f(a)) = a$ (A) $f(f^{1}(b)) = b$ (C) $f(f^{1}(b)) = b, b \subset y$ (D) $f^{1}(f(a)) = a, a \subset x$ Ans. D Х The given data is shown in the figure Sol. Since $f^{-1}(d) = x$ $\Rightarrow f(x) = d$ Now, if $a \subset x$, $f(a) \subset d$ \Rightarrow f⁻¹(f(a)) = a. d Ģ **f**⁻¹ '(a

Analyse your performance in Screening Test for evaluation of your preparation for Mains. A comprehensive analysis of your preparation on different topics would be couriered to you. Fill this sheet as per answers you have made in the IIT-JEE Screening Examination as per the sequencing provided in the solution booklet and send to nearest **FIITJEE**'s office immediately.

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FIITJ€€ solutions to IIT–JEE, 2005 Screening

PHYSICS

(B) 2 A

(D) 1 A

57. Find current in 2Ω resistor (A) 0 (C) 4 A

Ans.

A

- Sol. According to Kirchoff's junction rule no current passes through 2Ω resistor. $\therefore i = 0$
- 58. In Young's double slit experiment the angular position of a point on the central maxima whose intensity is one fourth of maximum intensity

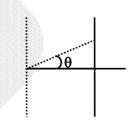
(A) $\sin^{-1}(\lambda/d)$	(B) $\sin^{-1}(\lambda/2d)$	
(C) $\sin^{-1}(\lambda/3d)$	(D) $\sin^{-1}(\lambda/4d)$	

Ans.

Sol. $I = I_{max} \cos^2 \phi / 2$

С

$$\Rightarrow \phi = 2\pi/3 \text{ and } \frac{2\pi}{\lambda} d\sin \theta = \frac{2\pi}{3}$$
$$\therefore \theta = \sin^{-1} \left(\frac{\lambda}{3d}\right)$$



10

59. Ratio of area of hole to beaker is 0.1. Height of liquid in beaker is 3m, and hole is at the height of 52.5 cm from the bottom of beaker, find the square of the velocity of liquid coming out from the hole (A) 50 (m/s)^2 (B) 50.5 (m/s)^2

(A) $50 (m/s)^2$	(B) 50.5 (m/s)
(C) 51 $(m/s)^2$	(D) $42 (m/s)^2$

 $= 50 (m/s)^2$

Ans.

Sol.
$$u^2 = \frac{1}{1}$$

↑ 3 m ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ 52.5 cm

60.In the figure shown, a cubical block is held stationary against a rough wall by
applying force 'F' then *incorrect* statement among the following is
(A) frictional force, f = Mg
(B) F = N, N is normal reaction
(C) F does not apply any torque(D) N does not apply any torque

Ans. D

Sol. For equilibrium, f = Mg F = N
For rotational equilibrium normal will shift downward. Hence torque due to friction about centre of mass = Torque due to Normal reaction about centre of mass.



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61.

Three infinitely charged sheets are kept parallel to x - y plane having charge densities as shown. Then the value of electric field at 'P' is $(B)\;\frac{4\sigma}{\varepsilon_0}\hat{k}$

(A)
$$\frac{-4\sigma}{\epsilon_0} \hat{k}$$

(C) $\frac{-2\sigma}{\epsilon_0} \hat{k}$

Ans.

С

Sol.
$$\vec{E}_{P} = \frac{\sigma}{2 \epsilon_{0}} \left(-\hat{k}\right) + \frac{\left(-2\sigma\right)}{2 \epsilon_{0}} \left(\hat{k}\right) + \frac{\left(-\sigma\right)}{2 \epsilon_{0}} \left(\hat{k}\right)$$
$$= \frac{-2\sigma}{\epsilon_{0}} \hat{k}$$

62. A cylindrical conducting rod is kept with its axis along positive z-axis, where a uniform magnetic field exists parallel to z-axis. The current induced in the cylinder is

(D) $\frac{2\sigma}{\epsilon_0}\hat{k}$

(B) $9MR^2$

 $=4MR^{2}$

(D) $\frac{40}{9}$ MR²

(A) zero

- (B) clockwise as seen from +z axis
- (C) anti-clockwise as seen from +z axis
- (D) opposite to the direction of magnetic field.

2R/3

Ans.

Sol. Since B is constant

Α

$$\frac{d\phi}{dt} = 0$$

$$\therefore i = 0$$

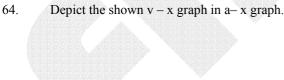
63. A circular disc of radius R/3 is cut from a circular disc of radius R and mass 9 M as shown. Then moment of inertia of remaining disc about 'O' perpendicular to the plane of the disc is

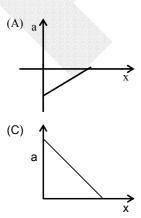
(A) 4 MR²
(C)
$$\frac{37}{9}$$
 MR²

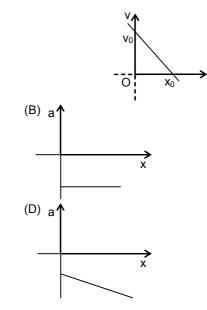
Ans.

A

Sol.
$$I_0 = \frac{9MR^2}{2} - \left[\frac{M(R/3)^2}{2} + M\left(\frac{2H}{3}\right)^2\right]$$







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Ans. A

.'

Sol. Equation of curve is

$$\frac{\mathbf{v}}{\mathbf{v}_0} + \frac{\mathbf{x}}{\mathbf{x}_0} = 1$$

$$\mathbf{v} = \left(1 - \frac{\mathbf{x}}{\mathbf{x}_0}\right) \mathbf{v}_0$$

:
$$a = \frac{dv}{dt} = -\frac{v_0}{x_0}(v) = -\frac{v_0^2}{x_0} \left(1 - \frac{x}{x_0}\right)$$

Alternative: $a = -v \left(\frac{dv}{dx}\right)$; but dv/dx is negative and v is decreasing with the increase in x.

Hence 'a' should increase with increase of 'x'.

- 65. A particle is confined to rotate in a circular path with decreasing linear speed, then which of the following is correct?
 - (A) \tilde{L} (angular momentum) is conserved about the centre.
 - (B) only direction of angular momentum \vec{L} is conserved.
 - (C) It spirals towards the centre.
 - (D) its acceleration is towards the centre.

Ans. B

66. The atomic number (Z) of an element whose k_{α} wavelength is λ is 11. The atomic number of an element whose k_{α} wavelength is 4λ is equal to

(A) 6	(B) 11
(C) 44	(D) 4

Ans.

A

Sol. $(Z-1)^2 \lambda = \text{constant}$ $\therefore (10^2) \lambda = 4\lambda (Z-1)^2 \Rightarrow Z = 6$

67. The graph shown in the figure represents energy density E_{λ} versus λ for three sources sun, welding arc, tungsten filament. For λ_{max} , correct combination will be

- (A) 1 Tungsten, 2 Welding arc, 3 Sun
- (B) 1 Sun, 2 Tungsten, 3 Welding arc.
- (C) 1 Sun, 2 Welding arc, 3 Tungsten
- (D) 1 Welding arc, 2 Sun, 3 Tungsten

Ans.

68.

A

Sol. Temperature of sun would be maximum out of the given three as $\lambda_m T = constant$ λ_m for Sun is minimum

 T_1 is the time period of simple pendulum. The point of suspension moves vertically upwards according to y =

kt², where k = 1 m/s². New time period is T₂, then
$$\frac{T_1^2}{T_2^2} = ? (g = 10 \text{ m/s}^2)$$

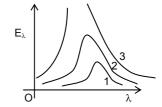
(A) 4/5 (B) 6/5
(C) 5/6 (D) 1

Ans.

B

Sol. Acceleration of the point of suspension

$$a = \frac{d^2 y}{dt^2} = 2k = 2 \text{ m/s}^2$$
$$T = 2\pi \sqrt{\frac{L}{g_{eff}}} \Rightarrow T_1 = 2\pi \sqrt{\frac{L}{10}} \text{ and } T_2 = 2\pi \sqrt{\frac{L}{12}}$$
$$\therefore \frac{T_1^2}{T_2^2} = \frac{6}{5}$$



- 69. Which of the following does not have the same dimension?
 - (A) Electric flux, Electric field, Electric dipole moment
 - (B) Pressure, stress, Young's modulus
 - (C) Electromotive force, Potential difference, Electric voltage.
 - (D) Heat, Potential energy, Work done

Ans. A

70. A capacitor (C = $4.0 \ \mu\text{F}$) is connected through a resistor (R = $2.5 \ \text{M}\Omega$) across a battery of negligible internal resistance of voltage 12 volts. The time after which the potential difference across the capacitor becomes three times to that of resistor is (ln 2 = 0.693)

(A) 13.86 sec.

(C) 3.24 sec.

(B) 6.48 sec. (D) 20.52 sec.

Ans.

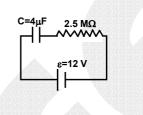
Α

$$q = C\varepsilon \left(1 - e^{-\frac{t}{RC}}\right) \implies i = \frac{\varepsilon}{R} e^{-\frac{t}{RC}}$$

$$3V_R = V_C$$

$$\implies \varepsilon \left(1 - e^{-\frac{t}{RC}}\right) = 3\varepsilon e^{-\frac{t}{RC}} \implies e^{-t/RC} = 1/4$$

$$t/RC = 2\ln2 \therefore t = 20 \times (0.693) = 13.86 \text{ sec}$$



- 71. A photon of energy 10.2 eV collides inelastically with a Hydrogen atom in ground state. After a certain time interval of few micro seconds another photon of energy 15.0 eV collides inelastically with the same hydrogen atom, then the observation made by a suitable detector is
 - (A) 1 photon with energy 10.2 eV and an electron with energy 1.4 eV
 - (B) 2 photon with energy 10.2 eV
 - (C) 2 photon with energy 1.4 eV
 - (D) one photon with energy 3.4 eV and 1 electron with energy 1.4 eV

Ans.

Α

- Sol. 10.2 eV photon on collision will excite H-atom to first excited state but Hydrogen atom will return to ground state before next collision. Second photon will provide ionization energy to Hydrogen atom, i.e., electron will be ejected with energy = 1.4 eV
- 72. In a resonance tube with tuning fork of frequency 512Hz, first resonance occurs at water level equal to 30.3 cm and second resonance occurs at 63.7 cm. The maximum possible error in the speed of sound is (A) 51.2 cm/s
 (B) 102.4 cm/s
 (C) 204.8 cm/s
 (D) 153.6 cm/s

Ans.

Sol.
$$\ell_1 + e = \frac{v}{4f}$$
$$\ell_2 - \ell_1 = \frac{2v}{4f}$$
$$\frac{\Delta(\ell_2 - \ell_1)}{(\ell_2 - \ell_1)}$$

C

 $\begin{pmatrix} \ell_2 - \ell_1 \end{pmatrix}^{-1} \mathbf{v}$ $\Delta \mathbf{v} = 2f\Delta(\ell_2 - \ell_1) = 2f(\Delta \ell_1 + \Delta \ell_2)$ $= 2 \times 512 \times 0.2 = 204.8 \text{ cm/s}.$

Δv

and $\ell_2 + e$

(For maximum error)

73. A thin concave and a thin convex lens are in contact. The ratio of the magnitude of power of two lenses is 2/3 and focal length of combination is 30cm, then the focal length of individual lenses are
(A) - 15 cm, 10 cm
(B) - 75 cm, 50 cm
(C) 75 cm, - 50 cm
(D) 75 cm, 50 cm

Ans.

Sol	P _{concave}	$=\frac{2}{2}$
501.	P _{convex}	3

A

$$\frac{1}{F} = \frac{1}{f_{\text{concave}}} + \frac{1}{f_{\text{convex}}}$$
$$\frac{1}{30} = \frac{-2}{3f} + \frac{1}{f} = \frac{1}{3f} \implies f = 10 \text{ cm, where f is focal length of convex lens}$$

74. Which of the following process does not occur through convection

(A) Boiling of water
(B) Land breeze and Sea breeze
(C) Circulation of air around furnace
(D) Heating of glass bulb through filament

Ans. D

Sol. Heating of glass bulb is by radiation.

75. A tank of height 33.25 cm is completely filled with liquid ($\mu = 1.33$). An object is placed at the bottom of tank on the axis of concave mirror as shown in the figure. Image of the object is formed 25 cm below the surface of the liquid, then focal length of the mirror is (A) 10 cm (B) 15 cm

Ans. C

(C)

Sol. After first refraction, position of the image = $\frac{33.25}{1.33}$ = 25cm

From reflection, $\frac{1}{f} = \frac{1}{v} - \frac{1}{25 + 15} = \frac{1}{v} - \frac{1}{40}$

From second refraction position of the object = $\frac{25}{1.33}$

$$\frac{1}{f} = -\frac{1}{15 + \frac{25}{1.33}} - \frac{1}{40} \Rightarrow f = -18.31 \text{ cm}$$

Hence magnitude of focal length of convex lens is 18.31 cm. The nearest possible matching answer is 20 cm.

- 76. In YDSE, an electron beam is used to obtain interference pattern. If speed of electron is increased then (A) no interference pattern will be observed.
 - (B) distance between two consecutive fringes will increase.
 - (C) distance between two consecutive fringes will decrease.
 - (D) distance between two consecutive fringes remains same.

Ans.

Sol. $\lambda =$

С

h

mv

v is increased, λ is decreased. $\beta = \lambda D/d \implies \beta$ decreases.

77. A spherical body of area A and emissivity e = 0.6 is kept inside a perfectly black body. Total heat radiated by the body at temperature T

(A) 0.4 e AT^4	(B) $0.8eAT^4$
(C) 0.6 eAT^4	(D) $1.0eAT^4$

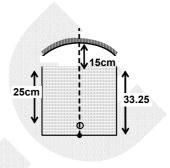
Ans.

D

- Sol. When a non black body is placed inside a hollow enclosure the total radiation from the body is the sum of what it would emit in the open (with e < 1) and the part (1-a) of the incident radiation from the walls reflected by it. The two add up to a black body radiation. Hence the total radiation emitted by the body is $1.0\sigma AT^4$. Probably in the examination paper ' σ ' is misprinted as 'e'
- 78. An open organ pipe resonated with frequency ' f_1 ' and 2^{nd} harmonic. Now one end is closed and the frequency is slowly increased then it resonates with frequency f_2 and nth harmonic then

(A)
$$n = 3, f_2 = \frac{3}{4} f_1$$

(B) $n = 5, f_2 = \frac{3}{4} f_1$
(C) $n = 3, f_2 = \frac{5}{4} f_1$
(D) $n = 5, f_2 = \frac{5}{4} f_1$



Ans. D $f_1 = \frac{1}{B}$ Sol.

$$\ell \bigvee \rho$$

$$f_2 = \frac{n}{4\ell} \sqrt{\frac{B}{\rho}}$$

$$\frac{f_1}{f_2} = \frac{4}{n} \Longrightarrow f_2 = \frac{n}{4} f_1$$

For the first resonance n = 5, $f_2 = \frac{5}{4}f_1$ (as frequency increases)

Temperature of a gas is 20°C and pressure is changed from 1.01×10⁵ Pa to 1.165×10⁵ Pa . If volume is decreased 79. isothermally by 10%. Bulk modulus of gas is

(A) 1.55×10^{5}	(B) 0.155×10^{5}
(C) 1.4×10^5	(D) 1.01×10^5

Ans.

A

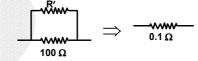
Sol.
$$B = -\Delta P/(\Delta V/V) = -\frac{(1.165 - 1.01) \times 10^5}{0.1} = 1.55 \times 10^5$$

80. A galvanometer with resistance 100Ω is converted to ammeter with a resistance of 0.1Ω . The galvanometer shows full scale deflection with a current of 100 μ A. Then the minimum current in the circuit for full scale deflection of galvanometer will be (A) 100.1mA (B) 10.01mA (D) 0.1001mA

(C) 1.001mA

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Ans.
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 $0.1 = \frac{100R'}{100 + R'} \Longrightarrow R' = \frac{100}{1001}$ Sol. $(100) (100 \times 10^{-6}) = R'(I - 100 \times 10^{-6})$ \therefore I = 100.1 mA



One calorie is defined as the heat required to raise the temperature of 1 gm of water by 1°C in a certain interval 81. of temperature and at certain pressure. The temperature interval and pressure is (A) 13.5° C to 14.5° C & 76 mm of Hg (B) 6.5° C to 7.5° C & 76 mm of Hg (C) 14.5° C to 15.5° C & 760 mm of Hg (D) 98.5°C to 99.5°C & 760 mm of Hg

Ans.

By definition. Sol.

С

C

If a star converts all of its Helium into oxygen nucleus, find the amount of energy released per nucleus of 82. oxygen. He = 4.0026 amu, O = 15.9994 amu (A) 7.26 MeV (B) 7 MeV (C) 10.24 MeV (D) 5.12 MeV

Ans.

 $E = \Delta mc^2 = [4 \times 4.0026 - 15.9994] \times 931.5 = 10.24 \text{ MeV}$ Sol.

Two litre of water at initial temperature of 27^oC is heated by a heater of power 1 kW. If the lid of kettle is 83. opened, then heat is lost at the constant rate of 160 J/s. Find the time required to raise the temperature of water to 77° C with the lid open (Specific heat of water 4.2 kJ/kg) (A) 5 min 40 sec (B) 14 min 20 sec (C) 8 min 20 sec (D) 16 min 10 sec

Ans. С

Rate of heat gain = 1000 - 160 = 840 J/s Sol.

Required time = $\frac{2 \times 4.2 \times 10^3 \times (77 - 27)}{840}$ = 500 sec = 8 min 20 sec

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84. Ideal gas is contained in a thermally insulated and rigid container and it is heated through a resistance 100Ω by passing a current of 1A for five minutes, then change in internal energy of the gas is
 (A) 0 kJ
 (B) 30 kJ
 (C) 10 kJ
 (D) 20 kJ

Ans. B

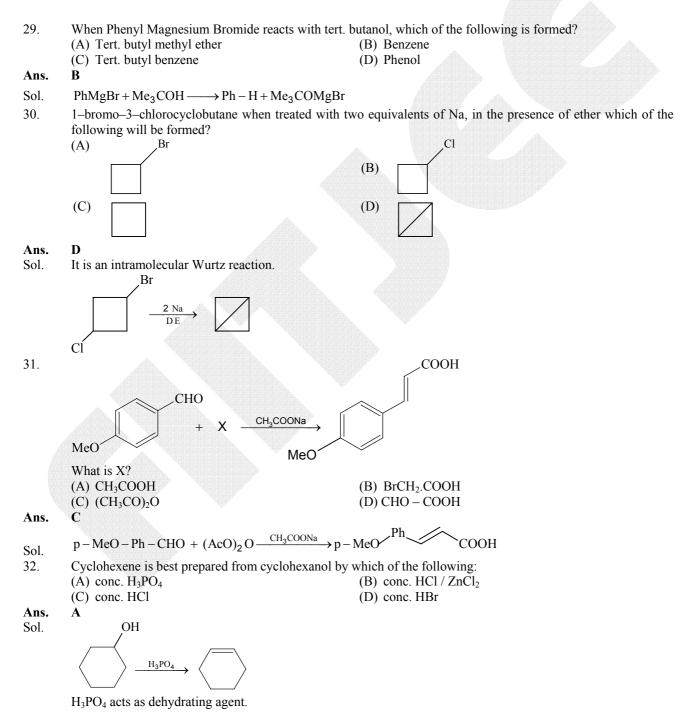
Sol. $\Delta W = 0 \therefore \Delta Q = \Delta U$ $\Delta Q = \Delta U = I^2 R \Delta t = (1)^2 (100)(5 \times 60) = 30 \text{ kJ}$

Analyse your performance in Screening Test for evaluation of your preparation for Mains. A comprehensive analysis of your preparation on different topics would be couriered to you. Fill this sheet as per answers you have made in the IIT-JEE Screening Examination as per the sequencing provided in the solution booklet and send to nearest **FIITJEE**'s office immediately.

Note: **FIITJEE** solutions to IIT–JEE, 2005 Screening Test is based on Screening Test paper created using memory retention of select **FIITJEE** students appeared in this test and hence may not exactly be the same as the original paper. However, every effort has been made to reproduce the original paper in the interest of the aspiring students.

FIITJ€€ solutions to IIT–JEE, 2005 Screening

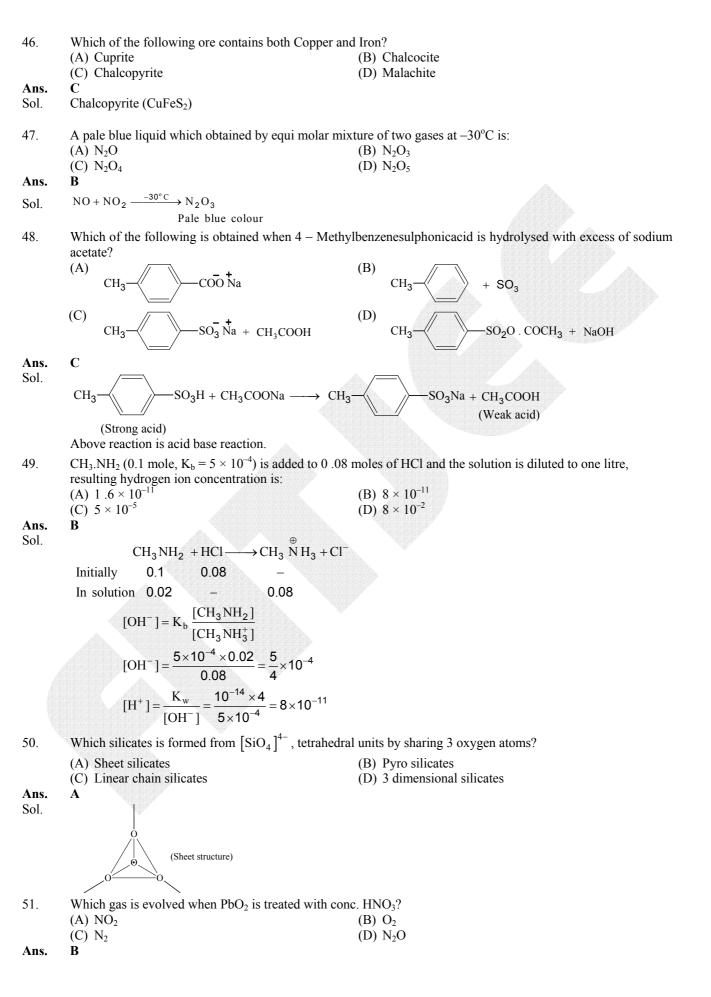
CHEMISTRY



$H_{S}C \longrightarrow (-1) \xrightarrow{(-1)}{H_{S}} H \xrightarrow{(-1)}$	33.	H_3C CH_3 H CH_3 NO_2 on hydrolysis in aqueous acetone gives,	
$\begin{array}{c} \begin{array}{c} & & & & \\ & & & & \\ & & & & \\ & & & & $		О NO ₂ (К)	
$\int_{H} \int_{H} \int_{H$		0	
(A) K and I. (B) Only K (C) L and M (D) Only M Ans. A Sol. S _x 1 and S _x 2, both reactions are possible due to aqueous acetone solution. 34. For 1-methoxy-1,3-butadiene, which of the following resonating structure is the least stable? (A) H_2C —CH—CH—CH—O—CH ₃ (B) H_2C —CH—CH—CH—O—CH ₃ (C) H_2C —CH—CH—CH—O—CH ₃ (D) H_2C —CH—CH—CH—O—CH ₃ (A) NaOH, NaI / H ⁺ (B) Fehling Solution (C) NaOH, I_2 / H ⁺ (D) Tollen's reagent Ans. C Sol. H_3C —C+CH ₃ H_3C +C+COOH Iddoform test. 36. Two forms of D – glucopyranose, are called. (A) Enantioners (B) Anomers (C) Epimers (D) Diastereomers Ans. B Sol. D – glucopyranose is cyclic form of glucose. Around C – 1 (Newly formed chiral centre, due to cycle formation) two isomers are observed. They are called as α and β – Anomers. 37. Which of the following pair is expected to exhibit same colour in solution? (A) VOCl ₂ ; FeCl ₂ (D) FeCl ₂ ; CuCl ₂ Ans. B Sol. V ⁺⁺ and Cu ²⁺ both have one unpaired electron available. 38. Which of the following isomers of phosphorus is thermodynamically most stable? (A) Red (D) Yellow		0 NO ₂ (M)	
(C) L and M (D) Only M Ans. A Sol. S _N ! and S _N 2, both reactions are possible due to aqueous acetone solution. 34. For 1-methoxy-1,3-butadiene, which of the following resonating structure is the least stable? (A) $H_{2}C$ —CH—CH—CH—CH—O—CH ₃ (B) $H_{2}C$ —CH—CH—CH—CH—O—CH ₃ (C) $H_{2}C$ —CH—CH—CH—CH—O—CH ₃ (D) $H_{2}C$ —CH—CH—CH—CH—O—CH ₃ (C) $H_{2}C$ —CH—CH—CH—CH—CH—O—CH ₃ (D) $H_{2}C$ —CH—CH—CH—O—CH ₃ (A) NaoCh, Nal / H ⁺ (B) Fehling Solution (C) NaOH, Nal / H ⁺ (B) Fehling Solution (C) NaOH, I_{2} / H ⁺ (D) Tollen's reagent Ans. C Sol. $H_{3}C$ (C) $H_{3}C$ (C) $H_{3}C$ (D) $H_{3}C$ (C) $H_{3}C$ (D) $Diastereomers$ Ans. B Sol. D – glucopyranose is cyclic form of glucose. Around C – 1 (Newly formed chiral centre, due to cycle formation) two isomers are observed. They are called as α and β – Anomers. 37. Which of the following pair is expected to exhibit same colour in solution? (A) $VOCI_{2}$; $FeCI_{2}$ (B) $CuCI_{2}$; $VOCI_{2}$ (A) $VOCI_{2}$; $FeCI_{2}$ (B) $CuCI_{2}$; $VOCI_{2}$ (A) $VOCI_{2}$; $FeCI_{2}$ (B) $CuCI_{2}$; $VOCI_{2}$ (C) $MaCI_{2}$; $FeCI_{2}$ (B) $CuCI_{2}$; $VOCI_{2}$ (A) $VCH = have one unpaired electron available. 38. Which of the following isomers of phosphorus is thermodynamically most stable? (A) Red (B) White (C) Black (D) Yetlow$			
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(A) H_2C — CH — CH — CH — CH — O — CH_3 (B) H_2C — CH — CH — CH — CH — CH_3 (C) H_2C — CH — CH — CH — CH — O — CH_3 (C) H_2C — CH — CH — CH — O — CH_3 (D) H_2C — CH — CH — CH — O — CH_3 Ans. C Sol. Point of difference is nature of carbanion. 2° carbanions are less stable than 1° – carbanions generally. 35. But–2–one can be converted to propanoic acid by which of the following: (A) NaOH, NaI / H ⁺ (B) Fehling Solution (C) NaOH, I_2 / H ⁺ (D) Tollen's reagent Ans. C Sol. C $H_3C \longrightarrow CH_3$ — $H_3C \longrightarrow COOH$ Iodoform test. 36. Two forms of D – glucopyranose, are called. (A) Enantiomers (B) Anomers (C) Epimers (D) Diastereomers Ans. B Sol. D – glucopyranose is cyclic form of glucose. Around C – 1 (Newly formed chiral centre, due to cycle formation) two isomers are observed. They are called as α and β – Anomers. 37. Which of the following pair is expected to exhibit same colour in solution? (A) VOCl ₂ ; FeCl ₂ (D) FeCl ₂ ; CuCl ₂ Ans. B Sol. V ⁺⁺ and Cu ²⁺ both have one unpaired electron available. 38. Which of the following isomers of phosphorus is thermodynamically most stable? (A) Red (C) Black (D) Yellow			
(A) H_2C $-CH$ $-CH$ $-CH$ $-CH$ $-O$ $-CH_3$ (B) H_2C $-CH$	34.		
(C) $H_2C = CH - CH - CH - O - CH_3$ (D) $H_2C = CH - CH - CH - CH - CH_3$ Ans. C Sol. Point of difference is nature of carbanion. 2° carbanions are less stable than 1° – carbanions generally. 35. But-2-one can be converted to propanoic acid by which of the following: (A) NaOH, NaI/H ⁺ (B) Fehling Solution (C) NaOH, I_2 / H ⁺ (D) Tollen's reagent Ans. C Sol. $H_3C - (CH_3 - MaOH, I_2) + H_3C - COOH$ Iddoform test. 36. Two forms of D – glucopyranose, are called. (A) Enantiomers (D) Diastercomers Ans. B Sol. D – glucopyranose is cyclic form of glucose. Around C – 1 (Newly formed chiral centre, due to cycle formation) two isomers are observed. They are called as α and β – Anomers. 37. Which of the following pair is expected to exhibit same colour in solution? (A) VOCI ₂ ; FeCI ₂ (B) CuCI ₂ ; VOCI ₂ (C) MnCI ₂ ; FeCI ₂ (D) FeCI ₂ ; CuCI ₂ Ans. B Sol. V ⁺ and Cu ²⁺ both have one unpaired electron available. 38. Which of the following isomers of phosphorus is thermodynamically most stable? (A) Red (B) White (C) Black (D) Yellow		(A) H_2C — CH — CH — CH — O — CH ₃ (B) H_2C — CH — CH — CH — O — CH ₃	
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(C) NaOH, I_2 / H^+ (D) Tollen's reagent Ans. C Sol. $H_3C \xrightarrow{-} CH_3 \xrightarrow{NaOH, I_2} H_3C \xrightarrow{-} COOH$ Iodoform test. 36. Two forms of D – glucopyranose, are called. (A) Enantiomers (B) Anomers (C) Epimers (D) Diastereomers Ans. B Sol. D – glucopyranose is cyclic form of glucose. Around C – 1 (Newly formed chiral centre, due to cycle formation) two isomers are observed. They are called as α and β – Anomers. 37. Which of the following pair is expected to exhibit same colour in solution? (A) VOCI ₂ ; FeCI ₂ (B) CuCI ₂ ; VOCI ₂ (C) MnCI ₂ ; FeCI ₂ (D) FeCI ₂ ; CuCI ₂ Ans. B Sol. V ⁴⁺ and Cu ²⁺ both have one unpaired electron available. 38. Which of the following isomers of phosphorus is thermodynamically most stable? (A) Red (B) White (C) Black (D) Yellow	35.	But-2-one can be converted to propanoic acid by which of the following:	
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 (C) MnCl₂; FeCl₂ (D) FeCl₂; CuCl₂ Ans. B Sol. V⁴⁺ and Cu²⁺ both have one unpaired electron available. 38. Which of the following isomers of phosphorus is thermodynamically most stable? (A) Red (B) White (C) Black (D) Yellow 	37.		
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(A) Red(B) White(C) Black(D) Yellow			
(C) Black (D) Yellow	38.		
	Ane	(C) Black (D) Yellow	
Sol. Due to layered structure in Black phosphorous, it is most stable.			

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39.	A metal nitrate gives black ppt. with KI and on add	ing excess of KI it gives orange colour. It is:
	(A) Hg^{+2}	(B) Bi^{+3}_{+2}
	(C) Sn^{+2}	(D) Pb^{+2}
Ans.	В	
Sol.	$\mathrm{Bi}^{+3} + \mathrm{KI} \longrightarrow \mathrm{BiI}_3 \downarrow$	
501.	Black	
	$\operatorname{BiI}_3 + \operatorname{KI} \longrightarrow \operatorname{K}[\operatorname{BiI}_4]$	
	Orange solution	
40.	Which of the following will not be oxidised by O_3 ?	
	(A) KI	(B) FeSO ₄
	(C) KMnO ₄	(D) K_2MnO_4
Ans.	C	
Sol.	KMnO ₄ can't be oxidised by any oxidising agents.	Mn is in maximum possible oxidation state of VI.
41.	Which type of isomerism is shown by Co(NH ₃) ₄ Br ₂	Cl?
	(A) Geometrical and Ionisation	(B) Optical and Ionisation
	(C) Geometrical and Optical	(D) Geometrical only
Ans.	A	
Sol.	$[Co(NH_3)_4(Br)_2]Cl$ can show both Geometrical and	I Ionisation isomerism.
42.	Which of the following FCC structure contains cati	ons in alternate tetrahedral voids?
	(A) NaCl	(B) ZnS
	(C) Na_2O	(D) CaF_2
Ans.	B (2^{-2}) (2^{-2}) (2^{-2}) (2^{-2})	
Sol.	In ZnS, Anions (S ⁻) are placed in fcc manner and c	cations (Zn^{+2}) are placed in alternate tetrahedral voids.
43.	[Some useful data, $Kb = 0.52 \text{ kg K mol}^{-1}$, molecula (A) 0.05	(B) 0.1
•	(C) 0.16	(D) 0.21
Ans.	C (12.44, 1000)	
Sol.	$\Delta T_{b} = i \times K_{b} \times m = 3 \times 0.52 \times \left(\frac{13.44}{134.4} \times \frac{1000}{1000}\right) = 0$.16
44.	The half cell reactions for rusting of iron are:	
	$2H^{+} + \frac{1}{2}O_{2} + 2e^{-} \longrightarrow H_{2}O; E^{0} = +1.23V, F_{0}$	$e^{2+} + 2e^- \longrightarrow Fe_{(S)}; E^0 = -0.44V$
	ΔG^0 (in kJ) for the reaction is:	
	(A) – 76	(B) -322
	(C) - 122	(D) – 176
Ans.	B	
Sol.	$2H^+ + \frac{1}{2}O_2 + 2e^- \longrightarrow H_2O; E^\circ = +1.23 V$	
	$Fe^{+2} + 2e^- \longrightarrow Fe_{(s)}; E^o = -0.44 V$	
	$\Gamma e + 2e \longrightarrow \Gamma e_{(s)}, E = -0.44 V$	
	$\operatorname{Fe}_{(s)} + 2\operatorname{H}^{+} + \frac{1}{2}\operatorname{O}_{2} \longrightarrow \operatorname{Fe}^{+2} + \operatorname{H}_{2}\operatorname{O}; \operatorname{E}_{cell}^{o} = 1.6^{\circ}$	7 V
	2^{10} 2^{10} 10^{10} 10^{10} 10^{10} 10^{10}	· ·
	$\Delta G^{o} = -n F E_{cell}^{o} = -2 \times 96.500 \times 1.67 = -322 \text{ kJ}$	
45.	The number of radial nodes in 3s and 2p respective	lv are:
	(A) 2 and 0	(B) 1 and 2
	(C) 0 and 2	(D) 2 and 1
Ans.	A	
Sol.	Number of radial nodes $= n - \ell - 1$	
	so, for 3s: $3 - 0 - 1 = 2$	
	For 2p: $2 - 1 - 1 = 0$	



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 $PbO_2 + 2HNO_3 \longrightarrow Pb(NO_3)_3 + H_2O + \frac{1}{2}O_2$ Sol.

52. If helium and methane are allowed to diffuse out of the container under the similar conditions of temperature and pressure, then the ratio of rate of diffusion of helium to methane is: (B) 1.0 (A) 2.0 (C) 0.5 (D) 4.0

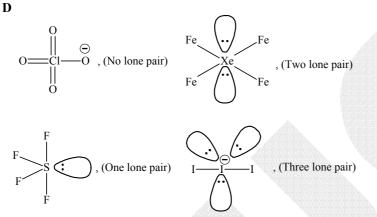
 $\frac{r_{\rm He}}{r_{\rm CH}} = \sqrt{\frac{16}{4}} = 2$: 1 Sol.

Α

Which of the following contains maximum number of lone pairs on the central atom? 53. $(A) ClO_3^-$ (B) XeF₄ (D) I_3^-

(C)
$$SF_4$$

Ans. Sol.



54. Which of the following is correct for lyophilic sols?

(A) They are irreversible

- (B) They are formed by inorganic substances
- (C) They are readily coagulated by addition of electrolytes
- (D) They are self stabilized
- Ans.

D

D

- Lyophilic sols are solvent loving in nature. Due to this property, such kind of sols are self stabilised. Sol.
- 55. Which of the following statement is incorrect about order of reaction?
 - (A) Order of reaction is determined experimentally
 - (B) It is the sum of power of concentration terms in the rate law expression
 - (C) It does not necessarily depend on stoichiometric coefficients
 - (D) Order of the reaction can not have fractional value.

Ans.

- Order of reaction is determined experimentally. It may be fractional. Sol.
- One mole of monoatomic ideal gas expands adiabatically at initial temperature T against a constant external 56. pressure of 1 atm. from one litre to two litre. Find out the final temperature (R = 0.0821 lt. atm K⁻¹ mole⁻¹)

(A) T

С

(C)
$$T - \frac{2}{3 \times 0.0821}$$

(B) $\frac{T}{(2)^{\frac{5}{3}-1}}$ (D) $T + \frac{2}{3 \times 0.0821}$

Ans.

Sol. Work done against constant external pressure $= P_{ext} (V_2 - V_1)$ In adiabatic condition $\Delta q = 0$ therefore $w = \Delta u$

$$\therefore -P_{ext} (V_2 - V_1) = \frac{3}{2} R (T_2 - T_1) \quad [Expansion work is negative]$$

On solving, $T_2 = T_1 - \frac{2}{3 \times 0.0821}$.