# fIITJE 

## Solutions to IIT-JEE-2012

## CODE

## PAPER 1

Time: 3 Hours
Maximum Marks: 210

## Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.

## INSTRUCTIONS

## A. General:

1. This booklet is your Question paper. Do not break the seats of his booklet before being instructed to do so by the invigilators.
2. The question paper CODE is printed on the right hand top corner of this page and on the back page of this booklet.
3. Blank spaces and blank pages are provided in this booklet for your rough work. No additional sheets will be provided for rough work.
4 Blank papers, clipboards, log tables, slide rules, calculators, cameras, cellular phones, pagers, and electronic gadgets are NOT allowed inside the examination hall.
4. Answers to the questions and personal details are to be filled on a two-part carbon-less paper, which is provided separately. You should not separate these parts. The invigilator will separate them at the end of examination. The upper sheet is machine-gradable Objective Response Sheet (ORS) which will be taken back by the invigilator.
5. Using a black ball point pen, darken the bubbles on the upper original sheet. Apply sufficient pressure so that the impression is created on the bottom sheet.
6. DO NOT TAMPER WITH /MUTILATE THE ORS OR THE BOOKLET.
7. On breaking the seals of the booklet check that it contains 28 pages and all 60 questions and corresponding answer choices are legible. Read carefully the instructions printed at the beginning of each section.
B. Filling the Right Part of the ORS:
8. The ORS also has a CODES printed on its left and right parts.
9. Check that the same CODE is printed on the ORS and on this booklet. IF IT IS NOT THEN ASK FOR A CHANGE OF THE BOOKLET. Sign at the place provided on the ORS affirming that you have verified that all the code are same.
10. Write your Name, Registration Number and the name of examination centre and sign with pen in the boxes provided on the right part of the ORS. Do not write any of this information anywhere else. Darken the appropriate bubble UNDER each digit of your Registration Number in such a way that the impression is created on the bottom sheet. Also darken the paper CODE given on the right side of $\operatorname{ORS}\left(\mathrm{R}_{4}\right)$.
C. Question paper format and Marking scheme:

The question paper consists of $\mathbf{3}$ parts (Physics, Chemistry and Mathematics). Each part consists of three sections.
12. Section I contains $\mathbf{1 0}$ multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONLY ONE is correct.
13. Section II contains $\mathbf{5}$ multiple choice questions. Each question has four choice (A), (B), (C) and (D) out of which ONE or MORE are correct.
14. Section III contains 5 questions. The answer to each question is a single digit integer, ranging from 0 to 9 (both inclusive).
D. Marking Scheme
15. For each question in Section I, you will be awarded $\mathbf{3}$ marks if you darken the bubble corresponding to the correct answer ONLY and zero marks if no bubbles are darkened. In all other cases, minus one ( $\mathbf{- 1}$ ) mark will be awarded in this section.
16. For each question in Section II, you will be awarded $\mathbf{4}$ marks if you darken ALL the bubble(s) corresponding to the correct answer(s) ONLY. In all other cases zero (0) marks will be awarded. No negative marks will be awarded for incorrect answer in this section.
17. For each question in Section III, you will be awarded $\mathbf{4}$ marks if you darken the bubble corresponding to the correct answer ONLY. In all other cases zero (0) marks will be awarded. No negative marks will be awarded for incorrect answer in this section.

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Write your Name, Registration Number and sign in the space provided on the back page of this booklet.
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## PAPER-1 [Code - 8] <br> IITJEE 2012

## PART - I: PHYSICS

## SECTION I : Single Correct Answer Type

This section contains $\mathbf{1 0}$ multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONLY ONE is correct.

1. In the determination of Young's modulus $\left(Y=\frac{4 \mathrm{MLg}}{\pi \ell \mathrm{d}^{2}}\right)$ by using Searle's method, a wire of length $\mathrm{L}=2 \mathrm{~m}$ and diameter $\mathrm{d}=0.5 \mathrm{~mm}$ is used. For a load $\mathrm{M}=2.5 \mathrm{~kg}$, an extension $\ell=0.25 \mathrm{~mm}$ in the length of the wire is observed. Quantities d and $\ell$ are measured using a screw gauge and a micrometer, respectively. They have the same pitch of 0.5 mm . The number of divisions on their circular scale is 100 . The contributions to the maximum probable error of the Y measurement
(A) due to the errors in the measurements of d and $\ell$ are the same.
(B) due to the error in the measurement of d is twice that due to the error in the measurement of $\ell$.
(C) due to the error in the measurement of $\ell$ is twice that due to the error in the measurement of d .
(D) due to the error in the measurement of d is four times that due to the error in the measurement of $\ell$.

Sol. (A)
L.C. $=\frac{0.5}{100}=0.005 \mathrm{~mm}$
$\frac{\Delta \mathrm{Y}}{\mathrm{Y}}=\frac{\Delta \ell}{\ell}+\frac{2 \Delta(\mathrm{~d})}{\mathrm{d}}$
$\frac{\Delta \ell}{\ell}=\frac{0.005 \times 10^{-3}}{0.25 \times 10^{-3}}=\frac{1}{50}$
$2 \frac{\Delta(\mathrm{~d})}{\mathrm{d}}=\frac{2 \times 0.005 \times 10^{-3}}{0.5 \times 10^{-3}}=\frac{1}{50}$
2. A small mass $m$ is attached to a massless string whose other end is fixed at $P$ as shown in the figure. The mass is undergoing circular motion in the $x-y$ plane with centre at $O$ and constant angular speed $\omega$. If the angular momentum of the system, calculated about $O$ and $P$ are denoted by $\vec{L}_{O}$ and $\vec{L}_{P}$ respectively, then

(A) $\overrightarrow{\mathrm{L}}_{\mathrm{O}}$ and $\overrightarrow{\mathrm{L}}_{\mathrm{P}}$ do not vary with time.
(B) $\overrightarrow{\mathrm{L}}_{\mathrm{O}}$ varies with time while $\overrightarrow{\mathrm{L}}_{\mathrm{P}}$ remains constant.
(C) $\overrightarrow{\mathrm{L}}_{\mathrm{O}}$ remains constant while $\overrightarrow{\mathrm{L}}_{\mathrm{P}}$ varies with time.
(D) $\overrightarrow{\mathrm{L}}_{\mathrm{O}}$ and $\overrightarrow{\mathrm{L}}_{\mathrm{P}}$ both vary with time.

Sol. (C)

3. A bi-convex lens is formed with two thin plano-convex lenses as shown in the figure. Refractive index $n$ of the first lens is 1.5 and that of the second lens is 1.2 . Both the curved surface are of the same radius of curvature $\mathrm{R}=14 \mathrm{~cm}$. For this bi-convex lens, for an object distance of 40 cm , the image distance will be
$\mathrm{n}=1.5$
(A) -280.0 cm
(B) 40.0 cm
(C) 21.5 cm
(D) 13.3 cm

Sol. (B)
$\mathrm{P}_{\mathrm{T}}=(1.5-1)\left(\frac{1}{14}-0\right)+(1.2-1)\left(0-\frac{1}{-14}\right)=\frac{0.5}{14}+\frac{0.2}{14}=\frac{1}{20}$
$\mathrm{f}=+20 \mathrm{~cm}$
$\frac{1}{\mathrm{v}}-\frac{1}{-40}=\frac{1}{20}$
$\frac{1}{\mathrm{v}}=\frac{1}{20}-\frac{1}{40}=\frac{1}{40}$
$\therefore \mathrm{v}=40 \mathrm{~cm}$
4. A thin uniform rod, pivoted at O , is rotating in the horizontal plane with constant angular speed $\omega$, as shown in the figure. At time $\mathrm{t}=0$, a small insect starts from O and moves with constant speed v , with respect to the rod towards the other end. It reaches the end of the rod at $t=T$ and stops. The angular speed of the system remains $\omega$ throughout. The magnitude of the torque $(|\vec{\tau}|)$ about O , as a function of time is best represented by which plot?

(A)

(C)
(B)

(D)


## Sol. (B)

$\tau=\omega \frac{\mathrm{dI}}{\mathrm{dt}}=\omega \frac{\mathrm{d}}{\mathrm{dt}}\left(\mathrm{C}+\mathrm{mv}^{2} \mathrm{t}^{2}\right)$
$=m \omega v^{2} 2 \mathrm{t}$.
5. A mixture of 2 moles of helium gas (atomic mass $=4 \mathrm{amu}$ ) and 1 mole of argon gas (atomic mass $=40$ amu) is kept at 300 K in a container. The ratio of the rms speeds $\left(\frac{\mathrm{v}_{\mathrm{rms}}(\text { helium })}{\mathrm{v}_{\mathrm{rms}}(\operatorname{argon})}\right)$ is
(A) 0.32
(B) 0.45
(C) 2.24
(D) 3.16

Sol. (D)
$\mathrm{v}_{\mathrm{rms}}=\sqrt{\frac{3 \mathrm{RT}}{\mathrm{M}}}$
Required ratio $=\sqrt{\frac{\mathrm{M}_{\mathrm{Ar}}}{\mathrm{M}_{\mathrm{He}}}}=\sqrt{\frac{40}{4}}=\sqrt{10}$

$$
=3.16
$$

6. Two large vertical and parallel metal plates having a separation of 1 cm are connected to a DC voltage source of potential difference X. A proton is released at rest midway between the two plates. It is found to move at $45^{0}$ to the vertical JUST after release. Then X is nearly
(A) $1 \times 10^{-5} \mathrm{~V}$
(B) $1 \times 10^{-7} \mathrm{~V}$
(C) $1 \times 10^{-9} \mathrm{~V}$
(D) $1 \times 10^{-10} \mathrm{~V}$

Sol. (C)
$\mathrm{qE}=\mathrm{mg}$
$\mathrm{q}(\mathrm{V} / \mathrm{d})=\mathrm{mg}$

$$
\begin{aligned}
V & =\frac{\mathrm{mgd}}{\mathrm{q}} \\
& =\frac{1.67 \times 10^{-27} \times 10 \times 10^{-2}}{1.6 \times 10^{-19}} \\
& =\frac{10^{-28}}{10^{-19}}=10^{-9} \mathrm{~V}
\end{aligned}
$$

7. Three very large plates of same area are kept parallel and close to each other. They are considered as ideal black surfaces and have very high thermal conductivity. The first and third plates are maintained at temperatures 2 T and 3 T respectively. The temperature of the middle (i.e. second) plate under steady state condition is
(A) $\left(\frac{65}{2}\right)^{1 / 4} \mathrm{~T}$
(B) $\left(\frac{97}{4}\right)^{1 / 4} \mathrm{~T}$
(C) $\left(\frac{97}{2}\right)^{1 / 4} \mathrm{~T}$
(D) $(97)^{1 / 4} \mathrm{~T}$

Sol. (C)
$\sigma \mathrm{A}(2 \mathrm{~T})^{4}+\sigma \mathrm{A}(3 \mathrm{~T})^{4}=\sigma 2 \mathrm{~A}\left(\mathrm{~T}^{\prime}\right)^{4}$
$16 \mathrm{~T}^{4}+81 \mathrm{~T}^{4}=2\left(\mathrm{~T}^{\prime}\right)^{4}$
$97 \mathrm{~T}^{4}=2\left(\mathrm{~T}^{\prime}\right)^{4}$
$\left(\mathrm{T}^{\prime}\right)^{4}=\frac{97}{2} \mathrm{~T}^{4}$
${ }^{2 \mathrm{~T}}| |^{3 \mathrm{~T}}$
$\therefore \quad \mathrm{T}^{\prime}=\left(\frac{97}{2}\right)^{1 / 4} \mathrm{~T}$
8. A small block is connected to one end of a massless spring of un-stretched length 4.9 m . The other end of the spring (see the figure) is fixed. The system lies on a horizontal frictionless surface. The block is stretched by 0.2 m and released from rest at $\mathrm{t}=0$. It then executes simple harmonic motion with angular frequency $\omega=\pi / 3 \mathrm{rad} / \mathrm{s}$. Simultaneously at $\mathrm{t}=0$, a small pebble is projected with speed v form point P at an angle of $45^{0}$ as shown in the figure. Point P is at a horizontal distance of 10 m from O. If the pebble hits the block at $\mathrm{t}=1 \mathrm{~s}$, the value of v is (take $\mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}$ )

(A) $\sqrt{50} \mathrm{~m} / \mathrm{s}$
(B) $\sqrt{51} \mathrm{~m} / \mathrm{s}$
(C) $\sqrt{52} \mathrm{~m} / \mathrm{s}$
(D) $\sqrt{53} \mathrm{~m} / \mathrm{s}$

Sol. (A)
$\frac{2 \mathrm{v} \sin 45^{\circ}}{\mathrm{g}}=1$
$\therefore \mathrm{v}=\sqrt{50} \mathrm{~m} / \mathrm{s}$.
9. Young's double slit experiment is carried out by using green, red and blue light, one color at a time. The fringe widths recorded are $\beta_{\mathrm{G}}, \beta_{\mathrm{R}}$ and $\beta_{\mathrm{B}}$, respectively. Then,
(A) $\beta_{G}>\beta_{B}>\beta_{R}$
(B) $\beta_{\mathrm{B}}>\beta_{\mathrm{G}}>\beta_{\mathrm{R}}$
(C) $\beta_{\mathrm{R}}>\beta_{\mathrm{B}}>\beta_{\mathrm{G}}$
(D) $\beta_{R}>\beta_{G}>\beta_{B}$

Sol. (D)
$\lambda_{\mathrm{R}}>\lambda_{\mathrm{G}}>\lambda_{\mathrm{B}}$
$\therefore \beta_{\mathrm{R}}>\beta_{\mathrm{G}}>\beta_{\mathrm{B}}$
10. Consider a thin spherical shell of radius R with centre at the origin, carrying uniform positive surface charge density. The variation of the magnitude of the electric field $|\vec{E}(r)|$ and the electric potential $V(r)$ with the distance r from the centre, is best represented by which graph?
(A)

(B)

(C)

(D)


## Sol. (D)

## SECTION II : Multiple Correct Answer(s) Type

This section contains 5 multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONE or MORE are correct.
11. Consider the motion of a positive point charge in a region where there are simultaneous uniform electric and magnetic fields $\vec{E}=E_{0} \hat{j}$ and $\vec{B}=B_{0} \hat{j}$. At time $t=0$, this charge has velocity $\vec{v}$ in the $x-y$ plane, making an angle $\theta$ with the $x$-axis. Which of the following option(s) is (are) correct for time $t>0$ ?
(A) If $\theta=0^{\circ}$, the charge moves in a circular path in the $x-z$ plane.
(B) If $\theta=0^{\circ}$, the charge undergoes helical motion with constant pitch along the $y$-axis.
(C) If $\theta=10^{\circ}$, the charge undergoes helical motion with its pitch increasing with time, along the $y$-axis.
(D) If $\theta=90^{\circ}$, the charge undergoes linear but accelerated motion along the $y$-axis.

Sol. (C, D)
If $\theta=90^{\circ}, \vec{B}$ exerts no force on $q$.
If $\theta=0^{\circ}, 10^{\circ}$; the charge particle moves in helix with increasing pitch due to $\vec{E}$ along y-axis.
12. A cubical region of side a has its centre at the origin. It encloses three fixed point charges, -q at $(0,-\mathrm{a} / 4,0),+3 \mathrm{q}$ at $(0,0,0)$ and -q at $(0,+\mathrm{a} / 4,0)$. Choose the correct options(s)

(A) The net electric flux crossing the plane $x=+a / 2$ is equal to the net electric flux crossing the plane $x=$ $-\mathrm{a} / 2$
(B) The net electric flux crossing the plane $y=+a / 2$ is more than the net electric flux crossing the plane $y=$ $-\mathrm{a} / 2$.
(C) The net electric flux crossing the entire region is $\frac{\mathrm{q}}{\varepsilon_{0}}$.
(D) The net electric flux crossing the plane $\mathrm{z}=+\mathrm{a} / 2$ is equal to the net electric flux crossing the plane $\mathrm{x}=$ $+\mathrm{a} / 2$.

Sol. (A, C, D)
Net flux through the cubical region $=\frac{-\mathrm{q}+3 \mathrm{q}-\mathrm{q}}{\varepsilon_{0}}=\frac{\mathrm{q}}{\varepsilon_{0}}$
The flux passing through the faces $x=\frac{-a}{2}, x=+\frac{a}{2}$ and $z=+\frac{a}{2}$ are same due to symmetry.
13. A person blows into open-end of a long pipe. As a result, a high pressure pulse of air travels down the pipe. When this pulse reaches the other end of the pipe,
(A) a high-pressure pulse starts travelling up the pipe, if the other end of the pipe is open.
(B) a low-pressure pulse starts travelling up the pipe, if the other end of the pipe is open.
(C) a low-pressure pulse starts travelling up the pipe, if the other end of the pipe is closed.
(D) a high-pressure pulse starts travelling up the pipe, if the other end of the pipe is closed.

Sol. (B, D)
At the open end, the phase of a pressure wave changes by $\pi$ radian due to reflection. At the closed end, there is no change in the phase of a pressure wave due to reflection.
14. A small block of mass of 0.1 kg lies on a fixed inclined plane $P Q$ which makes an angle $\theta$ with the horizontal. A horizontal force of 1 N acts on the block through its centre of mass as shown in the figure. The block remains stationary if (take $\mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}$ )

(A) $\theta=45^{\circ}$
(B) $\theta>45^{\circ}$ and a frictional force acts on the block towards P.
(C) $\theta>45^{\circ}$ and a frictional force acts on the block towards Q .
(D) $\theta<45^{\circ}$ and a frictional force acts on the block towards Q .

Sol. (A, C)
At $\theta=45^{\circ}, \mathrm{mg} \sin \theta=1 \times \cos \theta$
At $\theta>45^{\circ}, \mathrm{mg} \sin \theta>1 \times \cos \theta$ (friction acts upward)
At $\theta<45^{\circ}, \mathrm{mg} \sin \theta<1 \times \cos \theta$ (friction acts downward)
15. For the resistance network shown in the figure, choose the correct option(s)

(A) The current through PQ is zero.
(B) $\mathrm{I}_{1}=3 \mathrm{~A}$
(C) The potential at S is less than that at Q .
(D) $\mathrm{I}_{2}=2 \mathrm{~A}$

Sol. (A, B, C, D)
Nodes P and Q are equipotential and nodes S and T are equipotential from wheatstone bridge, no current passes through PQ and ST.
$\mathrm{I}_{1}=\frac{12}{4}=3 \mathrm{~A}$
$I_{2}=I_{1}\left(\frac{12}{6+12}\right)=2 \mathrm{~A}$


## SECTION III : Integer Answer Type

This section contains 5 questions. The answer to each question is single digit integer, ranging from 0 to 9 (both inclusive).
16. A circular wire loop of radius $R$ is placed in the $x-y$ plane centered at the origin $O$. A square loop of side $a(a \ll R)$ having two turns is placed with its centre at $z=\sqrt{3} R$ along the axis of the circular wire loop, as shown in figure. The plane of the square loop makes an angle of $45^{\circ}$ with respect to the z -axis. If the mutual inductance between the loops is given by $\frac{\mu_{0} a^{2}}{2^{p / 2} R}$, then the value of $p$ is


Sol. (7)
$\mathrm{M}=\frac{\mathrm{N} \phi}{\mathrm{I}}=\frac{2\left[\frac{\mu_{0} \mathrm{IR}^{2}}{2\left(8 \mathrm{R}^{3}\right)}\right] \mathrm{a}^{2} \cos 45^{\circ}}{\mathrm{I}} \frac{\mu_{0} \mathrm{a}^{2}}{8 \mathrm{R} 2^{1 / 2}}=\frac{\mu_{0} \mathrm{a}^{2}}{\mathrm{R} 2^{7 / 2}}$
So $P=7$
17. An infinitely long solid cylinder of radius $R$ has a uniform volume charge density $\rho$. It has a spherical cavity of radius $R / 2$ with its centre on the axis of the cylinder, as shown in the figure. The magnitude of the electric field at the point P , which is at a distance 2 R from the axis of the cylinder, is given by the expression $\frac{23 \rho \mathrm{R}}{16 \mathrm{k} \varepsilon_{0}}$. The value of k is


Sol. (6)
$\vec{E}=\frac{\lambda(\hat{j})}{2 \pi \varepsilon_{0}(2 R)}+\frac{K\left(\rho \frac{4}{3} \pi \frac{R^{3}}{8}\right)(-\hat{j})}{4 R^{2}}$
$\overrightarrow{\mathrm{E}}=\frac{\rho \pi \mathrm{R}^{2}(\hat{\mathrm{j}})}{4 \pi \varepsilon_{0} \mathrm{R}}+\frac{\mathrm{K} \pi \rho \mathrm{R}(-\hat{\mathrm{j}})}{24}$
$\overrightarrow{\mathrm{E}}=\operatorname{K} \rho \pi R(\hat{\mathrm{j}})+\frac{\mathrm{K}}{24} \rho \pi R(-\hat{\mathrm{j}})$

$\overrightarrow{\mathrm{E}}=\operatorname{K} \rho \pi \mathrm{R} \frac{23}{24}(\hat{\mathrm{j}})=\frac{23}{96 \varepsilon_{0}} \rho \mathrm{R}(\hat{\mathrm{j}})$
$\Rightarrow \mathrm{k}=6$
18. A proton is fired from very far away towards a nucleus with charge $\mathrm{Q}=120 \mathrm{e}$, where e is the electronic charge. It makes a closest approach of 10 fm to the nucleus. The de Broglie wavelength (in units of fm ) of the proton at its start is: (take the proton mass, $\mathrm{m}_{\mathrm{p}}=(5 / 3) \times 10^{-27} \mathrm{~kg}$; h/e $=4.2 \times 10^{-15} \mathrm{~J} . \mathrm{s} / \mathrm{C}$; $\frac{1}{4 \pi \varepsilon_{0}}=9 \times 10^{9} \mathrm{~m} / \mathrm{F} ; 1 \mathrm{fm}=10^{-15}$ )

Sol. (7)
$0+\frac{1}{2} \mathrm{mv}^{2}=\frac{\mathrm{K}(\mathrm{Q}) \mathrm{e}}{10 \times 10^{-15}}=\frac{\mathrm{K}(120 \mathrm{e}) \mathrm{e}}{10 \times 10^{-15}}$
$\frac{1}{2} \times \frac{5}{3} \times 10^{-27} \mathrm{v}^{2}=\frac{9 \times 10^{9} \times 120 \times\left(1.6 \times 10^{-19}\right)^{2}}{10 \times 10^{-15}}$
$\mathrm{v}=\frac{9 \times 6 \times 10^{9} \times 120 \times 2.56 \times 10^{-38}}{50 \times 10^{-42}}$
$\mathrm{v}=\sqrt{331.776 \times 10^{13}}$
$\lambda=\frac{\mathrm{h}}{\mathrm{mv}}$
$\lambda=\frac{4.2 \times 10^{-15} \times 1.6 \times 10^{-19}}{\frac{5}{3} \times 10^{-27} \times \sqrt{331.776 \times 10^{13}}}=\frac{4.2 \times 4.8 \times 10^{-34}}{57.6 \times 5 \times 10^{-21}}=0.07 \times 10^{-13}$
$\lambda=7 \times 10^{-15}=7 \mathrm{fm}$
19. A lamina is made by removing a small disc of diameter 2 R from a bigger disc of uniform mass density and radius 2 R , as shown in the figure. The moment of inertia of this lamina about axes passing though O and P is $I_{O}$ and $I_{P}$ respectively. Both these axes are perpendicular to the plane of the lamina. The ratio $I_{P} / I_{O}$ to the nearest integer is


Sol. (3)
$\mathrm{I}_{\mathrm{p}}=\left[\frac{4 \mathrm{mR}^{2}}{2}+\mathrm{m}\left(4 \mathrm{R}^{2}\right)\right]-\left[\frac{\mathrm{m}}{4} \frac{\mathrm{R}^{2}}{2}+\frac{\mathrm{m}}{4} 5 \mathrm{R}^{2}\right]$
$\mathrm{I}_{\mathrm{p}}=\mathrm{mR}^{2}\left[(2+4)-\left(\frac{1}{8}+\frac{5}{4}\right)\right]$
$\mathrm{I}_{\mathrm{p}}=\mathrm{mR}^{2}\left(6-\frac{11}{8}\right)=\frac{37}{8} \mathrm{mR}^{2}$
$\mathrm{I}_{0}=\left(\frac{4 \mathrm{mR}^{2}}{2}\right)-\left(\frac{\mathrm{m}}{4} \frac{\mathrm{R}^{2}}{2}+\frac{\mathrm{m}}{4} \mathrm{R}^{2}\right)$
$\mathrm{I}_{0}=\mathrm{mR}^{2}\left[2-\left(\frac{1}{8}+\frac{1}{4}\right)\right]=\mathrm{mR}^{2}\left[2-\frac{3}{8}\right]=\mathrm{mR}^{2}\left(\frac{13}{8}\right)$
So $\frac{\mathrm{I}_{\mathrm{p}}}{\mathrm{I}_{0}}=\frac{37 / 8}{13 / 8} \square 3$ (Nearest Integer)
20. A cylindrical cavity of diameter a exists inside a cylinder of diameter 2 a as shown in the figure. Both the cylinder and the cavity are infinity long. A uniform current density J flows along the length. If the magnitude of the magnetic field at the point $P$ is given by $\frac{N}{12} \mu_{0} a J$, then the value of $N$ is


Sol. (5)
$B=\frac{\mu_{0}\left(J \pi a^{2}\right)}{2 \pi a}-\frac{\mu_{0}\left(J \pi a^{2} / 4\right)}{2 \pi\left(\frac{3 a}{2}\right)}$
$\mathrm{B}=\frac{5 \mu_{0} \mathrm{Ja}}{12}=\frac{\mu_{0} \mathrm{NJa}}{12}$
So $N=5$

## PAPER-1 [Code - 8] <br> IITJEE 2012

## PART - II: CHEMISTRY

## SECTION - I: Single Correct Answer Type

This section contains $\mathbf{1 0}$ multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONLY ONE is correct.
21. A compound $\mathrm{M}_{\mathrm{p}} \mathrm{X}_{\mathrm{q}}$ has cubic close packing (ccp) arrangement of X . Its unit cell structure is shown below. The empirical formula of the compound is

(A) MX
(B) $\mathrm{MX}_{2}$
(C) $\mathrm{M}_{2} \mathrm{X}$
(D) $\mathrm{M}_{5} \mathrm{X}_{14}$

Sol. (B)

$$
\begin{aligned}
& X=8 \times \frac{1}{8}+6 \times \frac{1}{2}=4 \\
& M=4 \times \frac{1}{4}+1=2
\end{aligned}
$$

So, unit cell formula of the compound is $\mathrm{M}_{2} \mathrm{X}_{4}$ and the empirical formula of the compound is $\mathrm{MX}_{2}$.
22. The carboxyl functional group $(-\mathrm{COOH})$ is present in
(A) picric acid
(B) barbituric acid
(C) ascorbic acid
(D) aspirin

Sol. (D)

(Picric Acid)

(Barbituric Acid)

(Ascorbic Acid)

(Asprin)
23. As per IUPAC nomenclature, the name of the complex $\left[\mathrm{Co}\left(\mathrm{H}_{2} \mathrm{O}\right)_{4}\left(\mathrm{NH}_{3}\right)_{2}\right] \mathrm{Cl}_{3}$ is
(A) Tetraaquadiaminecobalt (III) chloride
(B) Tetraaquadiamminecobalt (III) chloride
(C) Diaminetetraaquacobalt (III) chloride
(D) Diamminetetraaquacobalt (III) chloride

Sol. (D)
$\left[\mathrm{Co}\left(\mathrm{H}_{2} \mathrm{O}\right)_{4}\left(\mathrm{NH}_{3}\right)_{2}\right] \mathrm{Cl}_{3}$
Diamminetetraaquacobalt (III) chloride
24. In allene $\left(\mathrm{C}_{3} \mathrm{H}_{4}\right)$, the type(s) of hybridization of the carbon atoms is (are)
(A) $s p$ and $s p^{3}$
(B) $s p$ and $s p^{2}$
(C) only $s p^{2}$
(D) $s p^{2}$ and $s p^{3}$

Sol. (B)
$\mathrm{Hp}_{2} \mathrm{C}=\stackrel{\text { sp }}{\mathrm{Sp}}=\mathrm{Cp}_{2} \mathrm{SH}_{2}$
25. The kinetic energy of an electron in the second Bohr orbit of a hydrogen atom is [ $\mathrm{a}_{0}$ is Bohr radius]
(A) $\frac{\mathrm{h}^{2}}{4 \pi^{2} \mathrm{ma}_{0}^{2}}$
(B) $\frac{\mathrm{h}^{2}}{16 \pi^{2} \mathrm{ma}_{0}^{2}}$
(C) $\frac{\mathrm{h}^{2}}{32 \pi^{2} \mathrm{ma}_{0}^{2}}$
(D) $\frac{\mathrm{h}^{2}}{64 \pi^{2} \mathrm{ma}_{0}^{2}}$

Sol. (C)
As per Bohr's postulate,
$\operatorname{mvr}=\frac{\mathrm{nh}}{2 \pi}$
So, $v=\frac{n h}{2 \pi m r}$
$\mathrm{KE}=\frac{1}{2} \mathrm{mv}^{2}$
So, $\mathrm{KE}=\frac{1}{2} \mathrm{~m}\left(\frac{\mathrm{nh}}{2 \pi \mathrm{mr}}\right)^{2}$
Since, $r=\frac{a_{0} \times n^{2}}{z}$
So, for $2^{\text {nd }}$ Bohr orbit
$r=\frac{a_{0} \times 2^{2}}{1}=4 a_{o}$
$\mathrm{KE}=\frac{1}{2} \mathrm{~m}\left(\frac{2^{2} \mathrm{~h}^{2}}{4 \pi^{2} \mathrm{~m}^{2} \times\left(4 \mathrm{a}_{0}\right)^{2}}\right)$
$\mathrm{KE}=\frac{\mathrm{h}^{2}}{32 \pi^{2} \mathrm{ma}_{0}^{2}}$
26. Which ordering of compounds is according to the decreasing order of the oxidation state of nitrogen?
(A) $\mathrm{HNO}_{3}, \mathrm{NO}, \mathrm{NH}_{4} \mathrm{Cl}, \mathrm{N}_{2}$
(B) $\mathrm{HNO}_{3}, \mathrm{NO}, \mathrm{N}_{2}, \mathrm{NH}_{4} \mathrm{Cl}$
(C) $\mathrm{HNO}_{3}, \mathrm{NH}_{4} \mathrm{Cl}, \mathrm{NO}, \mathrm{N}_{2}$
(D) $\mathrm{NO}, \mathrm{HNO}_{3}, \mathrm{NH}_{4} \mathrm{Cl}, \mathrm{N}_{2}$

Sol. (B)
$\mathrm{H}_{\mathrm{N}}^{\mathrm{N}} \mathrm{O}_{3}, \stackrel{+2}{\mathrm{~N}} \mathrm{O}, \stackrel{0}{\mathrm{~N}_{2}}, \stackrel{-3}{\mathrm{~N}} \mathrm{H}_{4} \mathrm{Cl}$
27. For one mole of a van der Waals gas when $b=0$ and $T=300 \mathrm{~K}$, the $P V$ vs. $1 / V$ plot is shown below. The value of the van der Waals constant $a\left(\right.$ atm. $\left.l i t r e^{2} \mathrm{~mol}^{-2}\right)$ is

(A) 1.0
(B) 4.5
(C) 1.5
(D) 3.0

Sol. (C)

van der Waals equation for 1 mole of real gas is,
$\left(\mathrm{P}+\frac{\mathrm{a}}{\mathrm{V}^{2}}\right)(\mathrm{V}-\mathrm{b})=\mathrm{RT}$
but, $\mathrm{b}=0$ (given)
$\Rightarrow\left(\mathrm{P}+\frac{\mathrm{a}}{\mathrm{V}^{2}}\right)(\mathrm{V})=\mathrm{RT}$
$\therefore \quad \mathrm{PV}=-\mathrm{a} \times \frac{1}{\mathrm{~V}}+\mathrm{RT}$
$y=m x+c$
Slope $=\tan (\pi-\theta)=-\mathrm{a}$
So, $\tan \theta=\mathrm{a}=\frac{21.6-20.1}{3-2}=1.5$
or, $\tan \theta=\frac{24.6-20.1}{3-0}=1.5$
28. The number of aldol reaction(s) that occurs in the given transformation is

(A) 1
(B) 2
(C) 3
(D) 4

Sol. (C)

29. The colour of light absorbed by an aqueous solution of $\mathrm{CuSO}_{4}$ is
(A) orange- red
(B) blue-green
(C) yellow
(D) violet

Sol. (A)
Aqueous solution of copper sulphate absorbs orange red light and appears blue (complementary colour).

30. The number of optically active products obtained from the complete ozonolysis of the given compound is

(A) 0
(B) 1
(C) 2
(D) 4

Sol. (A)



## SECTION II : Multiple Correct Answer (s) Type

This section contains 5 multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONE or MORE are correct.
31. Which of the following hydrogen halides react(s) with $\mathrm{AgNO}_{3}(\mathrm{aq})$ to give a precipitate that dissolves in Na${ }_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(\mathrm{aq})$ ?
(A) HCl
(B) HF
(C) HBr
(D) HI

Sol. (A, C, D)

$$
\mathrm{HX}+\mathrm{AgNO}_{3} \rightarrow \mathrm{AgX} \downarrow+\mathrm{HNO}_{3}(\mathrm{X}=\mathrm{Cl}, \mathrm{Br}, \mathrm{I})
$$

$$
\mathrm{AgX}+2 \mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3} \rightarrow \mathrm{Na}_{3}\left[\mathrm{Ag}\left(\mathrm{~S}_{2} \mathrm{O}_{3}\right)_{2}\right]+\mathrm{NaX}
$$

32. Identify the binary mixture(s) that can be separated into individual compounds, by differential extraction, as shown in the given scheme.

(A) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{OH}$ and $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{COOH}$
(B) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{COOH}$ and $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{OH}$
(C) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{OH}$ and $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{OH}$
(D) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{OH}$ and $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{COOH}$

Sol. (B, D)
(A) Both are soluble in NaOH , hence inseparable.
(B) Only benzoic acid $\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{COOH}\right)$ is soluble in NaOH and $\mathrm{NaHCO}_{3}$, while benzyl alcohol $\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{OH}\right)$ is not. Hence, separable.
(C) Although NaOH can enable separation between benzyl alcohol $\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{OH}\right)$ and phenol $\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{OH}\right)$ as only the later is soluble in NaOH . However, in $\mathrm{NaHCO}_{3}$, both are insoluble. Hence, inseparable.
(D) $\alpha$-phenyl acetic acid $\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{COOH}\right)$ is soluble in NaOH and $\mathrm{NaHCO}_{3}$. While benzyl alcohol $\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{OH}\right)$ is not. Hence, separable.
33. For an ideal gas, consider only $P-V$ work in going from an initial state $\boldsymbol{X}$ to the final state $\boldsymbol{Z}$. The final state $\boldsymbol{Z}$ can be reached by either of the two paths shown in the figure. Which of the following choice(s) is(are) correct? [Take $\Delta \mathrm{S}$ as change in entropy and $w$ as work done]

(A) $\Delta \mathrm{S}_{\mathrm{x} \rightarrow \mathrm{z}}=\Delta \mathrm{S}_{\mathrm{x} \rightarrow \mathrm{y}}+\Delta \mathrm{S}_{\mathrm{y} \rightarrow \mathrm{z}}$
(B) $\mathrm{w}_{\mathrm{x} \rightarrow \mathrm{z}}=\mathrm{w}_{\mathrm{x} \rightarrow \mathrm{y}}+\mathrm{w}_{\mathrm{y} \rightarrow \mathrm{z}}$
(C) $\mathrm{w}_{\mathrm{x} \rightarrow \mathrm{y} \rightarrow \mathrm{z}}=\mathrm{w}_{\mathrm{x} \rightarrow \mathrm{y}}$
(D) $\Delta \mathrm{S}_{\mathrm{x} \rightarrow \mathrm{y} \rightarrow \mathrm{z}}=\Delta \mathrm{S}_{\mathrm{x} \rightarrow \mathrm{y}}$

Sol. (A, C)
$\Delta \mathrm{S}_{\mathrm{X} \rightarrow \mathrm{Z}}=\Delta \mathrm{S}_{\mathrm{X} \rightarrow \mathrm{Y}}+\Delta \mathrm{S}_{\mathrm{Y} \rightarrow \mathrm{Z}}[$ entropy $(\mathrm{S})$ is a state function, hence additive $]$
$\mathrm{w}_{\mathrm{X} \rightarrow \mathrm{Y} \rightarrow \mathrm{Z}}=\mathrm{w}_{\mathrm{X} \rightarrow \mathrm{Y}}$ (work done in $\mathrm{Y} \rightarrow \mathrm{Z}$ is zero as it is an isochoric process)
34. Which of the following molecules, in pure form, is (are) unstable at room temperature?
(A)

(B)

(C)

(D)


Sol. (B, C)

35. Choose the correct reason(s) for the stability of the lyophobic colloidal particles.
(A) Preferential adsorption of ions on their surface from the solution
(B) Preferential adsorption of solvent on their surface from the solution
(C) Attraction between different particles having opposite charges on their surface
(D) Potential difference between the fixed layer and the diffused layer of opposite charges around the colloidal particles

Sol. (A, D)
Lyophobic colloids are stable due to preferential adsorption of ions on their surface from solution and potential difference between the fixed layer and the diffused layer of opposite charges around the colloidal particles that makes lyophobic sol stable.

## SECTION III: Integer Answer Type

This section contains $\mathbf{5}$ questions. The answer to each question is a single digit integer, ranging from 0 to 9 (both inclusive).
36. $29.2 \%(\mathrm{w} / \mathrm{w}) \mathrm{HCl}$ stock solution has density of $1.25 \mathrm{~g} \mathrm{~mL}^{-1}$. The molecular weight of HCl is $36.5 \mathrm{~g} \mathrm{~mol}^{-1}$. The volume ( mL ) of stock solution required to prepare a 200 mL solution of 0.4 M HCl is

## Sol. (8)

Stock solution of $\mathrm{HCl}=29.2 \%(\mathrm{w} / \mathrm{w})$
Molarity of stock solution of $\mathrm{HCl}=\frac{29.2 \times 1000 \times 1.25}{36.5 \times 100}$
If volume of stock solution required $=\mathrm{V} \mathrm{ml}$
$\mathrm{V} \times \frac{29.2}{36.5} \times \frac{1000}{80}=200 \times 0.4$
$\Rightarrow \mathrm{V}=8 \mathrm{ml}$
37. The substituents $\mathbf{R}_{\mathbf{1}}$ and $\mathbf{R}_{\mathbf{2}}$ for nine peptides are listed in the table given below. How many of these peptides are positively charged at $\mathrm{pH}=7.0$ ?


| Peptide | $\mathbf{\mathbf { R } _ { 1 }}$ | $\mathbf{R}_{\mathbf{2}}$ |
| :---: | :---: | :---: |
| I | H | H |
| II | H | $\mathrm{CH}_{3}$ |
| III | $\mathrm{CH}_{2} \mathrm{COOH}$ | H |
| IV | $\mathrm{CH}_{2} \mathrm{CONH}_{2}$ | $\left(\mathrm{CH}_{2}\right)_{4} \mathrm{NH}_{2}$ |
| V | $\mathrm{CH}_{2} \mathrm{CONH}_{2}{ }^{{f5d4838f9-53c9-462a-8b4b-8b79ba6c2bcc}}$ | $\mathrm{CH}_{3}$ |

Sol. (4)
Peptides with isoelectric point $(\mathrm{pI})>7$, would exist as cation in neutral solution $(\mathrm{pH}=7)$.
IV, VI, VIII and IX
38. An organic compound undergoes first-order decomposition. The time taken for its decomposition to $1 / 8$ and $1 / 10$ of its initial concentration are $t_{1 / 8}$ and $t_{1 / 10}$ respectively. What is the value of $\frac{\left[t_{1 / 8}\right]}{\left[t_{1 / 10}\right]} \times 10$ ? (take $\log _{10} 2=$ $0.3)$

Sol. (9)
$\mathrm{t}_{1 / 8}=\frac{2.303 \log 8}{\mathrm{k}}=\frac{2.303 \times 3 \log 2}{\mathrm{k}}$
$\mathrm{t}_{1 / 10}=\frac{2.303}{\mathrm{k}} \log 10=\frac{2.303}{\mathrm{k}}$
$\left[\frac{\mathrm{t}_{1 / 8}}{\mathrm{t}_{1 / 10}}\right] \times 10=\frac{\left(\frac{2.303 \times 3 \log 2}{\mathrm{k}}\right)}{\left(\frac{2.303}{\mathrm{k}}\right)} \times 10=9$
39. When the following aldohexose exists in its D-configuration, the total number of stereoisomers in its pyranose form is


Sol. (8)



Hence total number of stereoisomers in pyranose form of D-configuration $=2^{3}=8$
40. The periodic table consists of 18 groups. An isotope of copper, on bombardment with protons, undergoes a nuclear reaction yielding element $\mathbf{X}$ as shown below. To which group, element $\mathbf{X}$ belongs in the periodic table?
${ }_{29}^{63} \mathrm{Cu}+{ }_{1}^{1} \mathrm{H} \rightarrow 6{ }_{0}^{1} \mathrm{n}+\alpha+2{ }_{1}^{1} \mathrm{H}+\mathrm{X}$
Sol. (8)
${ }_{29}^{63} \mathrm{Cu}+{ }_{1} \mathrm{H}^{1} \rightarrow 6{ }_{0} \mathrm{n}^{1}+{ }_{2}^{4} \mathrm{He}+2{ }_{1} \mathrm{H}^{1}+{ }_{\mathrm{Z}}^{\mathrm{A}} \mathrm{X}$
Mass number: $63+1=1 \times 6+4+1 \times 2+\mathrm{A}$
$\mathrm{A}=64-12=52$
Atomic number: $29+1=6 \times 0+2+2 \times 1+Z$
$\mathrm{Z}=30-4=26$
${ }_{\mathrm{Z}}^{\mathrm{A}} \mathrm{X}={ }_{26}^{52} \mathrm{Fe}$
Hence $X$ is in group ' 8 ' in the periodic table.

## PAPER-1 [Code - 8]

## IITJEE 2012

## PART - III: MATHEMATICS

## SECTION I : Single Correct Answer Type

This section contains $\mathbf{1 0}$ multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONLY ONE is correct.
41. If $\lim _{x \rightarrow \infty}\left(\frac{x^{2}+x+1}{x+1}-a x-b\right)=4$, then
(A) $\mathrm{a}=1, \mathrm{~b}=4$
(B) $\mathrm{a}=1, \mathrm{~b}=-4$
(C) $\mathrm{a}=2, \mathrm{~b}=-3$
(D) $\mathrm{a}=2, \mathrm{~b}=3$

Sol. (B)
Given $\lim _{x \rightarrow \infty}\left(\frac{x^{2}+x+1}{x+1}-a x-b\right)=4$
$\Rightarrow \lim _{x \rightarrow \infty} \frac{x^{2}+x+1-a x^{2}-a x-b x-b}{(x+1)}=4 \Rightarrow \lim _{x \rightarrow \infty} \frac{(1-a) x^{2}+(1-a-b) x+(1-b)}{(x+1)}=4$
$\Rightarrow 1-\mathrm{a}=0$ and $1-\mathrm{a}-\mathrm{b}=4 \Rightarrow \mathrm{~b}=-4, \mathrm{a}=1$.

then the determinant of the matrix Q is
(A) $2^{10}$
(B) $2^{11}$
(C) $2^{12}$
(D) $2^{13}$

Sol. (D)
$|Q|=\left|\begin{array}{lll}2^{2} a_{11} & 2^{3} a_{12} & 2^{4} a_{13} \\ 2^{3} a_{21} & 2^{4} a_{22} & 2^{5} a_{23} \\ 2^{4} a_{31} & 2^{5} a_{32} & 2^{6} a_{33}\end{array}\right| \Rightarrow|Q|=2^{2} \cdot 2^{3} \cdot 2^{4}\left|\begin{array}{ccc}a_{11} & a_{12} & a_{13} \\ 2 a_{21} & 2 a_{22} & 2 a_{23} \\ 2^{2} a_{31} & 2^{2} a_{32} & 2^{2} a_{33}\end{array}\right|$
$|Q|=2^{9} \cdot 2 \cdot 2^{2}\left|\begin{array}{lll}a_{11} & a_{12} & a_{13} \\ a_{21} & a_{22} & a_{23} \\ a_{31} & a_{32} & a_{33}\end{array}\right|$
$\Rightarrow|\mathrm{Q}|=2^{12}|\mathrm{P}|$
$|\mathrm{Q}|=2^{13}$.
43. The locus of the mid-point of the chord of contact of tangents drawn from points lying on the straight line $4 x-5 y=20$ to the circle $x^{2}+y^{2}=9$ is
(A) $20\left(x^{2}+y^{2}\right)-36 x+45 y=0$
(B) $20\left(x^{2}+y^{2}\right)+36 x-45 y=0$
(C) $36\left(x^{2}+y^{2}\right)-20 x+45 y=0$
(D) $36\left(x^{2}+y^{2}\right)+20 x-45 y=0$

Sol. (A)
Equation of the chord bisected at $\mathrm{P}(\mathrm{h}, \mathrm{k})$
$h x+k y=h^{2}+k^{2}$
Let any point on line be $\left(\alpha, \frac{4}{5} \alpha-4\right)$
Equation of the chord of contact is
$\Rightarrow \alpha x+\left(\frac{4}{5} \alpha-4\right) y=9$
Comparing (i) and (ii)
$\frac{h}{\alpha}=\frac{k}{\frac{4}{5} \alpha-4}=\frac{h^{2}+k^{2}}{9}$
$\alpha=\frac{20 h}{4 h-5 k}$
Now, $\frac{h(4 h-5 k)}{20 h}=\frac{h^{2}+k^{2}}{9}$
$20\left(h^{2}+\mathrm{k}^{2}\right)=9(4 \mathrm{~h}-5 \mathrm{k})$
$20\left(x^{2}+y^{2}\right)-36 x+45 y=0$.

44. The total number of ways in which 5 balls of different colours can be distributed among 3 persons so that each person gets at least one ball is
(A) 75
(B) 150
(C) 210
(D) 243

Sol. (B)
Number of ways
$=3^{5}-{ }^{3} \mathrm{C}_{1} \cdot 2^{5}+{ }^{3} \mathrm{C}_{2} 1^{5}$
$=243-96+3=150$.
45. The integral $\int \frac{\sec ^{2} x}{(\sec x+\tan x)^{9 / 2}} d x$ equals (for some arbitrary constant K )
(A) $-\frac{1}{(\sec x+\tan x)^{11 / 2}}\left\{\frac{1}{11}-\frac{1}{7}(\sec x+\tan x)^{2}\right\}+K$
(B) $\frac{1}{(\sec x+\tan x)^{11 / 2}}\left\{\frac{1}{11}-\frac{1}{7}(\sec x+\tan x)^{2}\right\}+K$
(C) $-\frac{1}{(\sec x+\tan x)^{11 / 2}}\left\{\frac{1}{11}+\frac{1}{7}(\sec x+\tan x)^{2}\right\}+K$
(D) $\frac{1}{(\sec x+\tan x)^{11 / 2}}\left\{\frac{1}{11}+\frac{1}{7}(\sec x+\tan x)^{2}\right\}+K$

Sol. (C)
$\mathrm{I}=\int \frac{\sec ^{2} x}{(\sec x+\tan x)^{9 / 2}} d x$
Let $\sec \mathrm{x}+\tan \mathrm{x}=\mathrm{t}$
$\Rightarrow \sec x-\tan x=1 / t$
Now $\left(\sec x \tan x+\sec ^{2} x\right) d x=d t$
$\sec x(\sec x+\tan x) d x=d t$
$\sec \mathrm{xdx}=\frac{d t}{t}, \frac{1}{2}\left(t+\frac{1}{t}\right)=\sec \mathrm{x}$
$\mathrm{I}=\frac{1}{2} \int \frac{\left(t+\frac{1}{t}\right)}{t^{9 / 2}} \frac{d t}{t}$
$=\frac{1}{2} \int\left(t^{-9 / 2}+t^{-13 / 2}\right) d t$
$=\frac{1}{2}\left[\frac{t^{-9 / 2+1}}{-\frac{9}{2}+1}+\frac{t^{-13 / 2^{+1}}}{-\frac{13}{2}+1}\right]$
$=\frac{1}{2}\left[\frac{t^{-7 / 2}}{-\frac{7}{2}}+\frac{t^{-11 / 2}}{-\frac{11}{2}}\right]$
$=-\frac{1}{7} t^{-7 / 2}-\frac{1}{11} t^{-11 / 2}$
$=-\frac{1}{7} \frac{1}{t^{7 / 2}}-\frac{1}{11} \frac{1}{t^{11 / 2}}$
$=-\frac{1}{t^{11 / 2}}\left(\frac{1}{11}+\frac{t^{2}}{7}\right)=-\frac{1}{(\sec x+\tan x)^{11 / 2}}\left\{\frac{1}{11}+\frac{1}{7}(\sec x+\tan x)^{2}\right\}+k$
46. The point $P$ is the intersection of the straight line joining the points $Q(2,3,5)$ and $R(1,-1,4)$ with the plane $5 x-4 y-z=1$. If $S$ is the foot of the perpendicular drawn from the point $T(2,1,4)$ to $Q R$, then the length of the line segment PS is
(A) $\frac{1}{\sqrt{2}}$
(B) $\sqrt{2}$
(C) 2
(D) $2 \sqrt{2}$

Sol. (A)
D. R. of QR is $1,4,1$

Coordinate of $\mathrm{P} \equiv\left(\frac{4}{3}, \frac{1}{3}, \frac{13}{3}\right)$
D. R. of PT is $2,2,-1$

Angle between QR and PT is $45^{\circ}$
And PT = 1
$\Rightarrow \mathrm{PS}=\mathrm{TS}=\frac{1}{\sqrt{2}}$

47. Let $f(x)=\left\{\begin{array}{rl}x^{2}\left|\cos \frac{\pi}{x}\right|, & x \neq 0 \\ 0, \quad x=0\end{array} \quad, x \in \square, \quad\right.$ then f is
(A) differentiable both at $\mathrm{x}=0$ and at $\mathrm{x}=2$
(B) differentiable at $\mathrm{x}=0$ but not differentiable at $\mathrm{x}=2$
(C) not differentiable at $x=0$ but differentiable at $x=2$
(D) differentiable neither at $\mathrm{x}=0$ nor at $\mathrm{x}=2$

Sol. (B)
$f^{\prime}(0)=\lim _{h \rightarrow 0} \frac{f(0+h)-f(0)}{h}$
$=\lim _{h \rightarrow 0} \frac{h^{2}\left|\cos \frac{\pi}{h}\right|-0}{h}$
$=\lim _{h \rightarrow 0} h \cos \left(\frac{\pi}{h}\right)=0$
so, $f(x)$ is differentiable at $x=0$
$f^{\prime}\left(2^{+}\right)=\lim _{h \rightarrow 0} \frac{f(2+h)-f(2)}{h}$
$=\lim _{h \rightarrow 0} \frac{(2+h)^{2}\left|\cos \frac{\pi}{2+h}\right|-0}{h}$

$$
\begin{aligned}
& =\lim _{h \rightarrow 0} \frac{(2+h)^{2} \cos \left(\frac{\pi}{2+h}\right)}{h} \\
& f^{\prime}\left(2^{+}\right)=\lim _{h \rightarrow 0} \frac{(2+h)^{2}}{h} \sin \left(\frac{\pi}{2}-\frac{\pi}{2+h}\right) \\
& =\lim _{h \rightarrow 0} \frac{(2+h)^{2}}{h} \sin \left[\frac{\pi \cdot h}{2(2+h)}\right] \\
& =\lim _{h \rightarrow 0} \frac{(2+h)^{2}}{\frac{\pi h}{2(2+h)}} \sin \frac{\pi h}{2(2+h)} \times \frac{\pi}{2(2+h)}=\pi
\end{aligned}
$$

Again, $f^{\prime}\left(2^{-}\right)=\lim _{h \rightarrow 0} \frac{f(2-h)-f(2)}{-h}$
$=\lim _{h \rightarrow 0} \frac{(2-h)^{2}\left|\cos \left(\frac{\pi}{2-h}\right)\right|}{-h}$
$=\lim _{h \rightarrow 0} \frac{-(2-h)^{2} \cos \left(\frac{\pi}{2-h}\right)}{-h}$
$=\lim _{h \rightarrow 0} \frac{(2-h)^{2} \sin \left[\frac{\pi}{2}-\frac{\pi}{2-h}\right]}{h}$
$=\lim _{h \rightarrow 0} \frac{(2-h)^{2}}{h} \cdot \sin \left[\frac{-\pi h}{2(2-h)}\right]$
$=-\lim _{h \rightarrow 0} \frac{(2-h)^{2}}{\frac{\pi h}{2(2-h)}} \cdot \sin \frac{\pi h}{2(2-h)} \times \frac{\pi}{2(2-h)}=-\pi$
48. Let $z$ be a complex number such that the imaginary part of $z$ is nonzero and $a=z^{2}+z+1$ is real. Then $a$ cannot take the value
(A) -1
(B) $\frac{1}{3}$
(C) $\frac{1}{2}$
(D) $\frac{3}{4}$

Sol. (D)
Given equation is $\mathrm{z}^{2}+\mathrm{z}+1-\mathrm{a}=0$
Clearly this equation do not have real roots if
D < 0
$\Rightarrow 1-4(1-\mathrm{a})<0$
$\Rightarrow 4 \mathrm{a}<3$
a< $\frac{3}{4}$.
49. The ellipse $E_{1}: \frac{x^{2}}{9}+\frac{y^{2}}{4}=1$ is inscribed in a rectangle R whose sides are parallel to the coordinate axes. Another ellipse $E_{2}$ passing through the point $(0,4)$ circumscribes the rectangle $R$. The eccentricity of the ellipse $E_{2}$ is
(A) $\frac{\sqrt{2}}{2}$
(B) $\frac{\sqrt{3}}{2}$
(C) $\frac{1}{2}$
(D) $\frac{3}{4}$

Sol. (C)
Equation of ellipse is $(y+2)(y-2)+\lambda(x+3)(x-3)=0$
It passes through $(0,4) \Rightarrow \lambda=\frac{4}{3}$
Equation of ellipse is $\frac{x^{2}}{12}+\frac{y^{2}}{16}=1$
$\mathrm{e}=\frac{1}{2}$.


## Alternate

Let the ellipse be $\frac{x^{2}}{a^{2}}+\frac{y^{2}}{b^{2}}=1$ as it is passing through $(0,4)$ and $(3,2)$.
So, $b^{2}=16$ and $\frac{9}{a^{2}}+\frac{4}{16}=1$
$\Rightarrow \mathrm{a}^{2}=12$
So, $12=16\left(1-\mathrm{e}^{2}\right)$
$\Rightarrow \mathrm{e}=1 / 2$.
50. The function $\mathrm{f}:[0,3] \rightarrow[1,29]$, defined by $f(x)=2 x^{3}-15 x^{2}+36 x+1$, is
(A) one-one and onto
(B) onto but not one-one
(C) one-one but not onto
(D) neither one-one nor onto

Sol. (B)
$f(x)=2 x^{3}-15 x^{2}+36 x+1$
$\mathrm{f}^{\prime}(\mathrm{x})=6 \mathrm{x}^{2}-30 \mathrm{x}+36$
$=6\left(x^{2}-5 x+6\right)$
$=6(x-2)(x-3)$
$f(x)$ is increasing in $[0,2]$ and decreasing in $[2,3]$
$f(x)$ is many one
$\mathrm{f}(0)=1$
$\mathrm{f}(2)=29$
$\mathrm{f}(3)=28$
Range is [1, 29]
Hence, $f(x)$ is many-one-onto

## SECTION II : Multiple Correct Answer(s) Type

This section contains 5 multiple choice questions. Each question has four choices (A), (B), (C) and (D) out of which ONE or MORE are correct.
51. Tangents are drawn to the hyperbola $\frac{x^{2}}{9}-\frac{y^{2}}{4}=1$, parallel to the straight line $2 \mathrm{x}-\mathrm{y}=1$. The points of contact of the tangents on the hyperbola are
(A) $\left(\frac{9}{2 \sqrt{2}}, \frac{1}{\sqrt{2}}\right)$
(B) $\left(-\frac{9}{2 \sqrt{2}},-\frac{1}{\sqrt{2}}\right)$
(C) $(3 \sqrt{3},-2 \sqrt{2})$
(D) $(-3 \sqrt{3}, 2 \sqrt{2})$

Sol. (A, B)
Slope of tangent $=2$

The tangents are $y=2 x \pm \sqrt{9 \times 4-4}$
i.e., $2 x-y= \pm 4 \sqrt{2}$
$\Rightarrow \frac{x}{2 \sqrt{2}}-\frac{y}{4 \sqrt{2}}=1$ and $-\frac{x}{2 \sqrt{2}}+\frac{y}{4 \sqrt{2}}=1$
Comparing it with $\frac{x x_{1}}{9}-\frac{y y_{1}}{4}=1$
We get point of contact as $\left(\frac{9}{2 \sqrt{2}}, \frac{1}{\sqrt{2}}\right)$ and $\left(-\frac{9}{2 \sqrt{2}},-\frac{1}{\sqrt{2}}\right)$

## Alternate:

Equation of tangent at $P(\theta)$ is $\left(\frac{\sec \theta}{3}\right) x-\left(\frac{\tan \theta}{2}\right) y=1$
$\Rightarrow$ Slope $=\frac{2 \sec \theta}{3 \tan \theta}=2$
$\Rightarrow \sin \theta=\frac{1}{3}$
$\Rightarrow$ points are $\left(\frac{9}{2 \sqrt{2}}, \frac{1}{\sqrt{2}}\right)$ and $\left(-\frac{9}{2 \sqrt{2}},-\frac{1}{\sqrt{2}}\right)$.
52. Let $\theta, \varphi \in[0,2 \pi]$ be such that $2 \cos \theta(1-\sin \varphi)=\sin ^{2} \theta\left(\tan \frac{\theta}{2}+\cot \frac{\theta}{2}\right) \cos \varphi-1, \tan (2 \pi-\theta)>0$ and $-1<\sin \theta<-\frac{\sqrt{3}}{2}$. Then $\varphi$ cannot satisfy
(A) $0<\varphi<\frac{\pi}{2}$
(B) $\frac{\pi}{2}<\varphi<\frac{4 \pi}{3}$
(C) $\frac{4 \pi}{3}<\varphi<\frac{3 \pi}{2}$
(D) $\frac{3 \pi}{2}<\varphi<2 \pi$

Sol. (A, C, D)
$2 \cos \theta(1-\sin \varphi)=\frac{2 \sin ^{2} \theta}{\sin \theta} \cos \varphi-1=2 \sin \theta \cos \varphi-1$
$2 \cos \theta-2 \cos \theta \sin \varphi=2 \sin \theta \cos \varphi-1$
$2 \cos \theta+1=2 \sin (\theta+\varphi)$
$\tan (2 \pi-\theta)>0 \Rightarrow \tan \theta<0$ and $-1<\sin \theta<-\frac{\sqrt{3}}{2}$
$\Rightarrow \theta \in\left(\frac{3 \pi}{2}, \frac{5 \pi}{3}\right)$
$\frac{1}{2}<\sin (\theta+\varphi)<1$
$\Rightarrow 2 \pi+\frac{\pi}{6}<\theta+\varphi<\frac{5 \pi}{6}+2 \pi$
$2 \pi+\frac{\pi}{6}-\theta_{\max }<\varphi<2 \pi+\frac{5 \pi}{6}-\theta_{\text {min }}$
$\frac{\pi}{2}<\varphi<\frac{4 \pi}{3}$.
53. If $y(x)$ satisfies the differential equation $y^{\prime}-y \tan x=2 x \sec x$ and $y(0)=0$, then
(A) $y\left(\frac{\pi}{4}\right)=\frac{\pi^{2}}{8 \sqrt{2}}$
(B) $y^{\prime}\left(\frac{\pi}{4}\right)=\frac{\pi^{2}}{18}$
(C) $y\left(\frac{\pi}{3}\right)=\frac{\pi^{2}}{9}$
(D) $y^{\prime}\left(\frac{\pi}{3}\right)=\frac{4 \pi}{3}+\frac{2 \pi^{2}}{3 \sqrt{3}}$

Sol. (A, D)
$\frac{d y}{d x}-y \tan x=2 x \sec x$
$\cos x \frac{d y}{d x}+(-\sin x) y=2 x$
$\frac{d}{d x}(y \cos x)=2 x$
$y(x) \cos x=x^{2}+c$, where $\mathrm{c}=0$ since $\mathrm{y}(0)=0$
when $\mathrm{x}=\frac{\pi}{4}, \mathrm{y}\left(\frac{\pi}{4}\right)=\frac{\pi^{2}}{8 \sqrt{2}}$, when $\mathrm{x}=\frac{\pi}{3}, \mathrm{y}\left(\frac{\pi}{3}\right)=\frac{2 \pi^{2}}{9}$
when $\mathrm{x}=\frac{\pi}{4}, \mathrm{y}^{\prime}\left(\frac{\pi}{4}\right)=\frac{\pi^{2}}{8 \sqrt{2}}+\frac{\pi}{\sqrt{2}}$
when $x=\frac{\pi}{3}, y^{\prime}\left(\frac{\pi}{3}\right)=\frac{2 \pi^{2}}{3 \sqrt{3}}+\frac{4 \pi}{3}$
54. A ship is fitted with three engines $E_{1}, E_{2}$ and $E_{3}$. The engines function independently of each other with respective probabilities $\frac{1}{2}, \frac{1}{4}$ and $\frac{1}{4}$. For the ship to be operational at least two of its engines must function. Let X denote the event that the ship is operational and let $\mathrm{X}_{1}, \mathrm{X}_{2}$ and $\mathrm{X}_{3}$ denote respectively the events that the engines $E_{1}, E_{2}$ and $E_{3}$ are functioning. Which of the following is(are) true ?
(A) $P\left[X_{1}^{c} \mid X\right]=\frac{3}{16}$
(B) $\mathrm{P}[$ Exactly two engines of the ship are functioning $\mid \mathrm{X}]=\frac{7}{8}$
(C) $P\left[X \mid X_{2}\right]=\frac{5}{16}$
(D) $P\left[X \mid X_{1}\right]=\frac{7}{16}$

Sol. (B, D)
$P\left(X_{1}\right)=\frac{1}{2}, P\left(X_{2}\right)=\frac{1}{4}, P\left(X_{3}\right)=\frac{1}{4}$
$\mathrm{P}(\mathrm{X})=P\left(X_{1} \cap X_{2} \cap X_{3}^{C}\right)+P\left(X_{1} \cap X_{2}^{C} \cap X_{3}\right)+P\left(X_{1}^{C} \cap X_{2} \cap X_{3}\right)+P\left(X_{1} \cap X_{2} \cap X_{3}\right)=\frac{1}{4}$
(A) $P\left(X_{1}^{C} / X\right)=\frac{P\left(X \cap X_{1}^{C}\right)}{P(X)}=\frac{\frac{1}{32}}{\frac{1}{4}}=\frac{1}{8}$
(B) $\mathrm{P}[$ exactly two engines of the ship are functioning $\mid \mathrm{X}]=\frac{\frac{7}{32}}{\frac{1}{4}}=\frac{7}{8}$
(C) $P\left(\frac{X}{X_{2}}\right)=\frac{\frac{5}{32}}{\frac{1}{4}}=\frac{5}{8}$
(D) $P\left(\frac{X}{X_{1}}\right)=\frac{\frac{7}{32}}{\frac{1}{2}}=\frac{7}{16}$
55. Let $S$ be the area of the region enclosed by $y=e^{-x^{2}}, y=0, x=0$ and $x=1$. Then
(A) $\mathrm{S} \geq \frac{1}{e}$
(B) $\mathrm{S} \geq 1-\frac{1}{e}$
(C) $\mathrm{S} \leq \frac{1}{4}\left(1+\frac{1}{\sqrt{e}}\right)$
(D) $\mathrm{S} \leq \frac{1}{\sqrt{2}}+\frac{1}{\sqrt{e}}\left(1-\frac{1}{\sqrt{2}}\right)$

Sol. (A, B, D)
$\mathrm{S}>\frac{1}{e} \quad$ (As area of rectangle $\mathrm{OCDS}=1 / \mathrm{e}$ )
Since $e^{-x^{2}} \geq e^{-x} \forall x \in[0,1]$
$\Rightarrow S>\int_{0}^{1} e^{-x} d x=\left(1-\frac{1}{e}\right)$
Area of rectangle OAPQ + Area of rectangle $\mathrm{QBRS}>\mathrm{S}$
$S<\frac{1}{\sqrt{2}}(1)+\left(1-\frac{1}{\sqrt{2}}\right)\left(\frac{1}{\sqrt{e}}\right)$.


Since $\frac{1}{4}\left(1+\frac{1}{\sqrt{e}}\right)<1-\frac{1}{e}$
Hence, (C) is incorrect.

## SECTION III : Integer Answer Type

This section contains 5 questions. The answer to each question is single digit integer, ranging from 0 to 9 (both inclusive).
56. If $\vec{a}, \vec{b}$ and $\vec{c}$ are unit vectors satisfying $|\vec{a}-\vec{b}|^{2}+|\vec{b}-\vec{c}|^{2}+|\vec{c}-\vec{a}|^{2}=9$, then $|2 \vec{a}+5 \vec{b}+5 \vec{c}|$ is

Sol. (3)

$$
\begin{aligned}
& \text { As, }|\vec{a}-\vec{b}|^{2}+|\vec{b}-\vec{c}|^{2}+|\vec{c}-\vec{a}|^{2}=3\left(|\vec{a}|^{2}+|\vec{b}|^{2}+|\vec{c}|^{2}\right)-|\vec{a}+\vec{b}+\vec{c}|^{2} \\
& \Rightarrow 3 \times 3-|\vec{a}+\vec{b}+\vec{c}|^{2}=9 \\
& \Rightarrow|\vec{a}+\vec{b}+\vec{c}|=0 \Rightarrow \vec{a}+\vec{b}+\vec{c}=0 \\
& \Rightarrow \vec{b}+\vec{c}=-\vec{a} \\
& \Rightarrow|2 \vec{a}+5(\vec{b}+\vec{c})|=|-3 \vec{a}|=3|\vec{a}|=3
\end{aligned}
$$

57. Let $\mathrm{f}: \operatorname{IR} \rightarrow$ IR be defined as $\mathrm{f}(\mathrm{x})=|x|+\left|x^{2}-1\right|$. The total number of points at which f attains either a local maximum or a local minimum is

Sol. (5)

$$
\begin{aligned}
& f^{\prime}(x)=\frac{|x|}{x}+\frac{\left|x^{2}-1\right|}{x^{2}-1} \cdot(2 x) \\
& =\left\{\begin{array}{cc}
2 x-1 & , \\
-(2 x+1) & , \\
1-1<-1<x<0 \\
1-2 x & , \\
2 x+1 & 0<x<1
\end{array}\right. \\
& \hline
\end{aligned}
$$



So, $f^{\prime}(x)$ changes sign at points
$\mathrm{x}=-1,-\frac{1}{2}, 0, \frac{1}{2}, 1$
so, total number of points of local maximum or minimum is 5 .
58. Let $S$ be the focus of the parabola $y^{2}=8 x$ and let $P Q$ be the common chord of the circle $x^{2}+y^{2}-2 x-4 y=$ 0 and the given parabola. The area of the triangle PQS is

Sol. (4)
The parabola is $x=2 t^{2}, y=4 t$
Solving it with the circle we get :
$4 t^{4}+16 t^{2}-4 t^{2}-16 t=0$
$\Rightarrow \mathrm{t}^{4}+3 \mathrm{t}^{2}-4 \mathrm{t}=0 \Rightarrow \mathrm{t}=0,1$
so, the points P and Q are $(0,0)$ and $(2,4)$ which are also diametrically opposite points on the circle. The focus is $S \equiv(2,0)$.
The area of $\triangle \mathrm{PQS}=\frac{1}{2} \times 2 \times 4=4$.
59. Let $\mathrm{p}(\mathrm{x})$ be a real polynomial of least degree which has a local maximum at $\mathrm{x}=1$ and a local minimum at x $=3$. If $p(1)=6$ and $p(3)=2$, then $p^{\prime}(0)$ is

Sol. (9)
Let $\mathrm{p}^{\prime}(\mathrm{x})=\mathrm{k}(\mathrm{x}-1)(\mathrm{x}-3)$
$\Rightarrow \mathrm{p}(\mathrm{x})=k\left(\frac{x^{3}}{3}-2 x^{2}+3 x\right)+c$
Now, $p(1)=6 \Rightarrow \frac{4}{3} k+c=6$
also, $\mathrm{p}(3)=2 \Rightarrow \mathrm{c}=2$
so, $\mathrm{k}=3$, so, $\mathrm{p}^{\prime}(0)=3 \mathrm{k}=9$.


Sol. (4)
Let $\sqrt{4-\frac{1}{3 \sqrt{2}} \sqrt{4-\frac{1}{3 \sqrt{2}} \sqrt{4-\frac{1}{3 \sqrt{2}} \cdots}}}=y$
So, $4-\frac{1}{3 \sqrt{2}} y=y^{2} \quad(\mathrm{y}>0)$
$\Rightarrow y^{2}+\frac{1}{3 \sqrt{2}} y-4=0 \Rightarrow y=\frac{8}{3 \sqrt{2}}$
so, the required value is $6+\log _{3 / 2}\left(\frac{1}{3 \sqrt{2}} \times \frac{8}{3 \sqrt{2}}\right)$
$=6+\log _{3 / 2} \frac{4}{9}=6-2=4$.

